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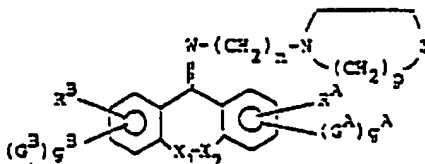
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(84) Tricyclic compounds.

(57) Tricyclic compounds represented by formula (I):



wherein

— represents single bond or double bond;

 $\overline{X_1-X_2}$  represents  $-CH_2O-$ .

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wherein 1 represents 0, 1 or 2,  $-\text{CH}_2-\text{CH}_2-$ , or  $-\text{CH}=\text{CH}-$ ;

W represents  $-\text{S}-$  or  $=\text{CH}-$ ;

n is 1, 2, 3, or 4;

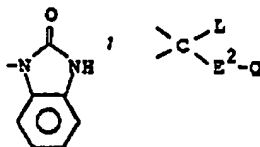
one of  $\text{R}^{\text{A}}$  and  $\text{R}^{\text{B}}$  represents hydrogen and the other represents  $-\text{Y}-\text{M}$  wherein Y represents single bond,  $-\text{CR}^1\text{R}^2-(\text{CH}_2)_m-$ , or  $-\text{CR}^1=\text{CR}^2-(\text{CH}_2)_m$  wherein each of  $\text{R}^1$  and  $\text{R}^2$  independently represents hydrogen or lower alkyl and m is 0, 1, 2, 3 or 4, in which the left side of each formula is bound to the mother nucleus; and M represents  $-\text{COOR}^3$  wherein  $\text{R}^3$  represents hydrogen or lower alkyl,  $-\text{CONR}^{\text{3a}}\text{R}^{\text{3b}}$  wherein each of  $\text{R}^{\text{3a}}$  and  $\text{R}^{\text{3b}}$  independently has the same significances for  $\text{R}^3$  as described above, or tetrazolyl;

each of  $\text{G}^{\text{A}}$  and  $\text{G}^{\text{B}}$  independently represents lower alkyl, halogen, hydroxyl, or lower alkoxy;

each of  $\text{g}^{\text{A}}$  and  $\text{g}^{\text{B}}$  independently represents 0, 1, 2 or 3;

Z represents  $\text{N}-\text{E}^1-\text{Q}$  wherein  $\text{E}^1$  represents single bond,  $-\text{CO}-$ ,  $-\text{COO}-$  wherein the left side of the formula is bound to the nitrogen atom, or  $-\text{SO}_2-$ ; and

Q represents optionally substituted aryl, optionally substituted aralkyl, optionally substituted aralkenyl, aromatic heterocyclic group, or

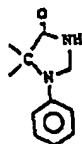


wherein L represents hydrogen, hydroxyl, or lower alkoxy;  $\text{E}^2$  represents single bond,  $-\text{CO}-$ , or  $-$

wherein  $\text{R}^4$  represents hydrogen or lower alkyl; and

Q has the same significance as described above;

$\text{C}=\text{CH}-\text{Q}$  wherein Q has the same significance as described above; or



p is 1, 2 or 3;

possess a potent antagonizing action against thromboxane  $\text{A}_2$  and also an antiallergic and/or antihistaminic activity, and are expected to have preventive and therapeutic effects on ischemic diseases; cerebrovascular diseases, etc.

## TRICYCLIC COMPOUNDS

BACKGROUND OF THE INVENTION

The present invention relates to novel tricyclic compounds which strongly antagonize an action of thromboxane  $A_2$  (hereafter referred to as  $TXA_2$ ) and possess an antiallergic and/or antihistaminic activity.

It is hitherto known that  $TXA_2$  strongly aggregates platelets and is a potent vasoconstrictor [cf. Arachidonic Acid Cascade and Drugs, edited by Shozo Yamamoto, Gendai Iryo Publishing Co., Ltd. (1985)]. Further  $TXA_2$  is a powerful vasoconstrictor against bronchus and bronchial smooth muscle. Therefore,  $TXA_2$  is considered to take part in pathological conditions over a wide range. As examples, the following diseases can be exemplified.

(1) Ischemic disease

For example, myocardial infarction, angina pectoris, and thrombosis

(2) Cerebro-vascular disease

For example, transient ischemic attack, migraine, cerebral hemorrhage, and cerebral infarction

(3) Peripheral vascular diseases and disease caused by unbalanced lipid metabolism

For example, atherosclerosis, capillary convulsion, peripheral circulation disorders, hypertension, and pulmonary embolism

(4) Inflammatory and allergic disease

For example, bronchial asthma, bronchitis, pneumonia, nephritis, and hepatitis

(5) Shock

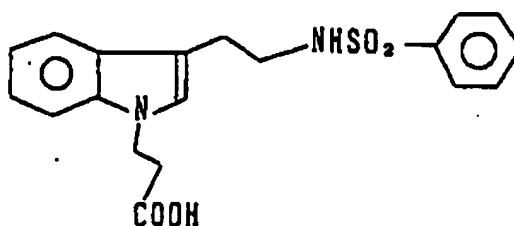
(6) Cancer metastasis

Accordingly, compounds that antagonize the action of  $TXA_2$  are expected to have therapeutic effects in preventing or treating optional one or more of the diseases describes above or other diseases involving  $TXA_2$ . Further where in drugs used for medical purposes heretofore, application thereof is limited due to side effects mediated by  $TXA_2$  or considered to be mediated by  $TXA_2$ , it is expected to alleviate the side effects by the use of compounds which antagonize the action of  $TXA_2$ .

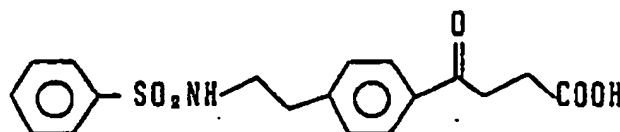
In recent years,  $TXA_2$  is also thought to play a role in the pathogenesis of allergic diseases, especially of an asthma [cf. J. Allergy Clin. Immunol., 77, 122 (1986); Arch. Pharmacol., 327, 148 (1984)].

As an antagonist of  $TXA_2$ , representative compounds are exemplified in Thrombosis Research, 44, 377 (1986).

Furthermore, an indole compound having the following structure:

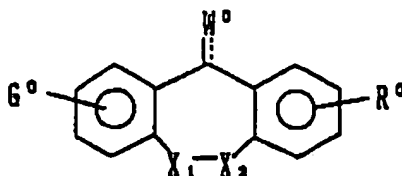


and the like are disclosed in Japanese Published Unexamined Patent Application No. 249960/1986 [West German Patent Application (DE) No. 3,514,696] and a compound having the following structure:



and the like are disclosed in Japanese Published Unexamined Patent Application No. 212552/1986 [West German Patent Application (DE) No. 3,508,692]. These compounds have a phenylsulfonamide group as a side chain and exhibit an activity of antagonizing  $TXA_2$ .

On the other hand, in tricyclic compounds represented by the following formula:



wherein  $R^\circ$  as a substituent on the aromatic ring has carboxyl or a derivative thereof (for example, an ester, an amide, etc.; hereafter collectively referred to as carboxylic acid group) directly or via an alkylene chain, etc. and  $W^\circ$  is hydrogen or a substituent such as oxo ( $=O$ ), methylene ( $=CH_2$ ), hydroxyl, alkoxy, etc., oxepine derivatives wherein  $X_1-X_2$  is  $-CH_2O-$  are known as showing antinflammatory or antiallergic activities, etc. [J. Med. Chem., 19, 941 (1976); *ibid.*, 20, 1499 (1977); *ibid.*, 21, 633 (1978); U.S. Patent No. 4,282,365 (Japanese Published Unexamined Patent Application No. 21679/1983); U.S. Patent No. 4,585,788; Japanese Published Unexamined Patent Application Nos. 152673/1986; 152674/1986 and 152675/1986].

Further, it is also known that oxepine derivatives wherein  $R^\circ$  is hydrogen or a substituent other than the carboxylic acid group, such as, alkyl, alkoxy, halogen, etc. and  $W^\circ$  has a (di)alkylaminoalkyl chain via  $-S-$  show antiallergic activities [Japanese Published Unexamined Patent Application No. 126883/1983 (EP 0085870A)]. It is also known that derivatives such as oxepine or thiepine (wherein  $X_1-X_2$  is  $-CH_2S-$ ) wherein  $W^\circ$  is alkylaminoalkylidene show an antidepressant action, etc. [U.S. Patent Nos. 3,354,155 and 3,420,851; Drugs, 13, 161 (1977); *Arz.-Forsch.*, 13, 1039 (1963); *ibid.*, 14, 100 (1964)]. Furthermore, it is also known that derivatives such as cycloheptene (wherein  $X_1-X_2$  is  $-CH=CH-$ ) or thiepine wherein  $W^\circ$  has an alkyl chain substituted with an alicyclic nitrogen-containing heterocyclic group such as piperazine, etc. at the terminal thereof via  $-NHCO-$  are known to have a calcium antagonizing activity [Japanese Published Unexamined Patent Application Nos. 47466/1986 (EP 191867A) and 153280/1987].

Further oxepine derivatives having an antiallergic activity wherein  $R^\circ$  has a carboxylic acid group and  $W^\circ$  has a (di)alkylaminoalkyl chain via  $-S-$  are known [Japanese Published Unexamined Patent Application Nos. 28972/1985 (U.S. Patent No. 4,596,804); 152669/1986, 152670/1986, 152671/1986 and 15672/1986 (all of them correspond to EP 188802A); 152676/1986 and 257981/1986]. Furthermore, oxepine or cycloheptene (wherein  $X_1-X_2$  is  $-CH_2CH_2-$ ) derivatives showing an antihistaminic activity wherein  $W^\circ$  is a (di)alkylaminoalkylidene are known [Japanese Published Unexamined Patent Application No. 45557/1986 (EP 214778A)]. Still further, oxepine derivatives wherein  $W^\circ$  is an alkylidene substituted with an alicyclic nitrogen-containing heterocyclic group such as 4-methylpiperazinyl, 4-methylhomopiperazinyl, piperidino, pyrrolidinyl, thiomorpholino or morpholino or with a (di)alkyl-substituted amino at the terminal thereof are known as showing an antiallergic and antinflammatory activity [Japanese Published Unexamined Patent Application No. 10784/1986 (EP 0235796A)].

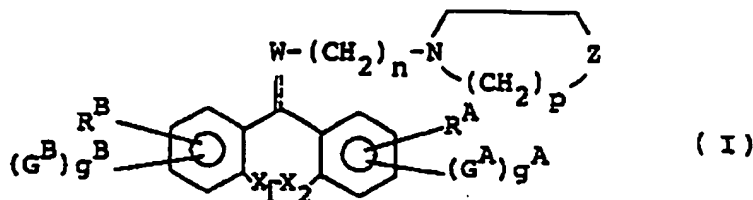
Novel and useful  $TXA_2$  antagonists are expected to have preventive and therapeutic effects on various diseases, and are in demand. Further antiallergic agents having a  $TXA_2$ -antagonizing activity are expected to have preventive and therapeutic effects on allergic diseases, and are in demand.

#### SUMMARY OF THE INVENTION

An object of the present invention is to provide novel tricyclic compounds having a  $TXA_2$ -antagonizing activity and antiallergic activity by containing both a carboxylic acid group as the foresaid  $R^\circ$ , and, as the foresaid  $W^\circ$ , an alkylthio chain or alkylidene chain substituted at the terminal thereof with an alicyclic nitrogen-containing heterocyclic group having a substituent such as aryl, aralkyl, etc. thereon.

The present invention relates to a tricyclic compound [hereafter referred to as Compound (I); compounds having other formula numbers are also the same] represented by formula (I):

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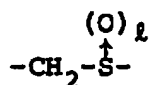
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wherein

— represents single bond or double bond;

 $\overline{X_1-X_2}$  represents  $-\text{CH}_2\text{O}-$ .

15

wherein  $l$  represents 0, 1 or 2,  $-\text{CH}_2-\text{CH}_2-$ , or  $-\text{CH}=\text{CH}-$ ;

20

 $W$  represents  $-\text{S}-$  or  $=\text{CH}-$ ; $n$  is 1, 2, 3, or 4;

one of  $R^A$  and  $R^B$  represents hydrogen and the other represents  $-\text{Y}-\text{M}$  wherein  $Y$  represents single bond,  $-\text{CR}^1\text{R}^2(\text{CH}_2)_m-$ , or  $-\text{CR}^1=\text{CR}^2(\text{CH}_2)_m$  wherein each of  $R^1$  and  $R^2$  independently represents hydrogen or lower alkyl and  $m$  is 0, 1, 2, 3 or 4, in which the left side of each formula is bound to the mother nucleus;

25

and  $M$  represents  $-\text{COOR}^3$  wherein  $R^3$  represents hydrogen or lower alkyl,  $-\text{CONR}^{3a}\text{R}^{3b}$  wherein each of  $R^{3a}$  and  $R^{3b}$  independently has the same significances for  $R^3$  as described above, or tetrazolyl;

each of  $G^A$  and  $G^B$  independently represents lower alkyl, halogen, hydroxyl, or lower alkoxy;

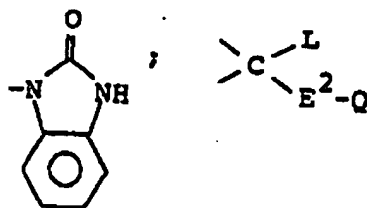
each of  $g^A$  and  $g^B$  independently represents 0, 1, 2 or 3;

$Z$  represents  $>\text{N}-\text{E}^1-\text{Q}$  wherein  $\text{E}^1$  represents single bond,  $-\text{CO}-$ ,  $-\text{COO}-$  wherein the left side of the formula is bound to the nitrogen atom, or  $-\text{SO}_2-$ ; and

30

$Q$  represents optionally substituted aryl, optionally substituted aralkyl, optionally substituted aralkenyl, aromatic heterocyclic group, or

35



40

wherein  $L$  represents hydrogen, hydroxyl, or lower alkoxy;  $\text{E}^2$  represents single bond,  $-\text{CO}-$ , or  $-\text{CH}-$   
 $\text{OR}^4$

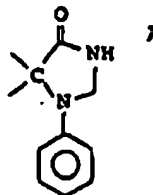
45

wherein  $R^4$  represents hydrogen or lower alkyl; and

$Q$  has the same significance as described above;

$>\text{C}=\text{CH}-\text{Q}$  wherein  $Q$  has the same significance as described above; or

50



55

p is 1, 2 or 3;  
and a pharmaceutically acceptable salt thereof.

## 5 DETAILED DESCRIPTION OF THE INVENTION

In the definition of Z in formula (I), the aryl is exemplified by phenyl and naphthyl having 6 to 10 carbon atoms, etc.; the aralkyl is exemplified by benzyl, phenethyl, benzhydryl and trityl, etc. having 7 to 20 carbon atoms, etc.; and the aralkenyl is exemplified by styryl and cinnamyl having 8 to 18 carbon atoms, etc. The  
10 substituent on each group means independently 1 to 3 substituents on the aromatic ring and includes a group selected from lower alkyl, halogen, trifluoromethyl, hydroxyl, lower alkoxy and methylenedioxy formed together with the ortho-position thereof. Likewise, the aromatic heterocyclic group shown by Q represents a group selected from furyl, thienyl, pyrrolyl, pyrimidinyl, quinolyl and isoquinolyl.

Further in the definition of each group in formula (I), the alkyl moiety in the lower alkyl and lower alkoxy  
15 is a straight or branched alkyl having 1 to 6 carbon atoms, for example, methyl, ethyl, propyl, isopropyl, butyl, isobutyl, sec-butyl, tert-butyl, pentyl, neopentyl, hexyl, etc.; the halogen includes, for example, fluorine, chlorine, bromine and iodine.

The pharmaceutically acceptable salt of Compound (I) includes an acid addition salt, a metal salt, an ammonium salt, an organic amine addition salt, an amino acid addition salt, etc. which are pharmaceutically  
20 acceptable.

As the pharmaceutically acceptable acid addition salt of Compound (I), mention may be made of the inorganic acid salt such as hydrochloride, sulfate, phosphate, etc. and the organic acid salt such as acetate, malate, fumarate, tartrate, citrate, etc. As the pharmaceutically acceptable metal salt, the alkali metal salt such as sodium salt, potassium salt, etc.; alkaline earth metal salt such as magnesium salt, calcium salt,  
25 etc. and further the aluminum salt and the zinc salt are appropriate. As the ammonium salt, mention may be made of the salt of ammonium, tetramethylammonium, etc. As the pharmaceutically acceptable organic amine addition salt, mention may be made of an addition salt of morpholine, piperidine, etc. As the pharmaceutically acceptable amino acid addition salt, an addition salt of lysine, glycine, phenylalanine and the like are mentioned.

Hereafter processes for producing Compound (I) are described below but the production of Compound (I) is not deemed to be limited thereto. Further in various processes, reaction conditions can be appropriately chosen from those described below.

A reaction solvent may be chosen from water or an organic solvent which does not participate in the reaction and can be used alone or in combination. The organic solvent includes, for example, an alcohol  
35 such as methanol, ethanol, propanol, isopropanol, etc.; an ether such as diethyl ether, dioxane, tetrahydrofuran, ethylene glycol monomethyl ether, ethylene glycol dimethyl ether, etc.; a hydrocarbon such as benzene, toluene, xylene, hexane, cyclohexane, petroleum ether, ligroin, decalin, etc.; a ketone such as acetone, methyl ethyl ketone, etc. an amide such as formamide, dimethylformamide, hexamethylphosphoric triamide, etc.; acetonitrile, ethyl acetate, dimethylsulfoxide, sulfolane or a halogenated hydrocarbon such as methylene chloride, dichloroethane, tetrachloroethane, chloroform or carbon tetrachloride, etc. Further in  
40 case that bases or acids later described are liquid, they may also be used as a solvent.

As the appropriate base, an inorganic or organic base can be used. These bases include an alkali metal hydroxide, for example, lithium hydroxide, sodium hydroxide or potassium hydroxide; an alkali metal carbonate, for example, sodium carbonate, sodium hydrogencarbonate or potassium carbonate; an alkali  
45 metal acetate, for example, sodium acetate or potassium acetate; an alkali metal alkoxide, for example, sodium methoxide, sodium ethoxide or potassium tert-butoxide; or an organic metal compound, for example, sodium hydride, n-butyl lithium, sec-butyl lithium; and an organic amine, for example, triethylamine, tri-n-butylamine, pyridine, N,N-dimethylaminopyridine, picoline, lutidine, N,N-dimethylaniline, dicyclohexylmethylamine, N-methylpiperidine, morpholine, diazabicyclooctane, diazabicycloundecene or N-benzyltrimethylammonium hydroxide (Triton B), etc.  
50

As the appropriate acid, an inorganic or organic acid or Lewis acid can be used. Examples of the inorganic acid include hydrochloric acid, hydrobromic acid, hydroiodic acid, sulfuric acid, nitric acid, hypochloric acid, sulfurous acid or nitrous acid, etc. Examples of the organic acid include formic acid, acetic acid, trifluoroacetic acid, benzoic acid, p-toluenesulfonic acid, camphorsulfonic acid or methanesulfonic acid,  
55 etc. Examples of the Lewis acid include aluminum chloride, zinc chloride, tin chloride, boron trifluoride, boron trifluoride diethyl ether complex, titanium tetrachloride, etc.

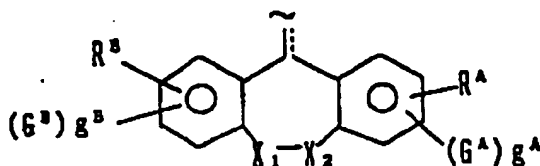
A reaction temperature is generally from -80° C to a boiling point of a solvent. Heating without a solvent is also possible. The reaction may generally be carried out under normal pressure but it is also possible to

apply pressure. In this case, the reaction temperature may be raised to a temperature higher than the boiling point of a solvent.

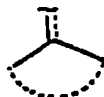
A reaction time is generally in a range of one minute to one week.

In the following description, preferred reaction conditions are given.

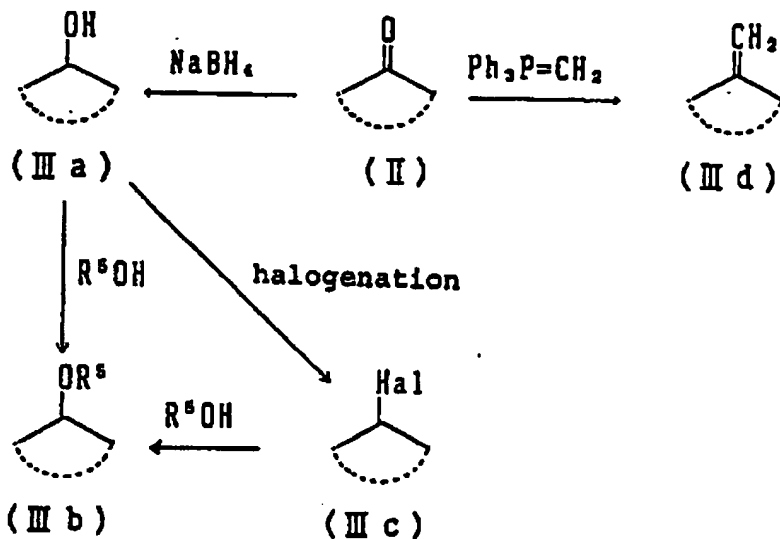
Further in the following description, the tricyclic moiety which does not directly participate in the reaction:



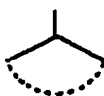
wherein  $\sim$ ,  $X_1-X_2$ ,  $R^A$ ,  $R^B$ ,  $G^A$ ,  $G^B$ ,  $g^A$  and  $g^B$  have the same significances as described above; is sometimes referred to as:



Compound (I) can be prepared from Compound (II) or from Compounds (IIIa through d), etc. obtained from Compound (II) according to the following reaction steps:



wherein



has the same significance as described above;  $R^5$  represents lower alkyl, Hal represents halogen and Ph represents phenyl.

Herein the halogen shown by Hal represents chlorine, bromine and iodine and the lower alkyl has the

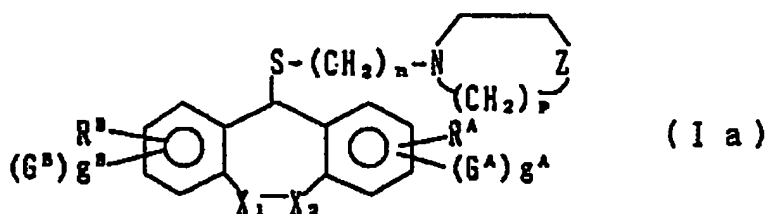
same significance as defined for the lower alkyl in each group in formula (I).

Compounds (II) are either described in J. Med. Chem., 19, 941 (1976); *ibid.*, 21, 1035 (1978); *ibid.*, 20, 1557 (1977); *ibid.*, 20, 1499 (1977); *ibid.*, 29, 2347 (1986); *ibid.*, 21, 633 (1978); *ibid.*, 20, 458 (1977); U.S. Patent Nos. 4,172,949 and 4,282,365; Japanese Published Unexamined Patent Application Nos. 21679/1983; 28972/1985; 152869/1986; 152672/1986; 152673/1986; 152675/1986 and 10784/1988 etc. or can be synthesized according to methods described in these publications or in a manner similar thereto.

Further the processes for producing Compounds (IIIa to d) from Compound (II) can be synthesized according to methods described in Japanese Published Unexamined Patent Application Nos. 150083/1981; 28972/1985; 152670/1986; 152671/1986; 152672/1986; 152675/1986 and 10784/1988, etc. or in a manner similar thereto.

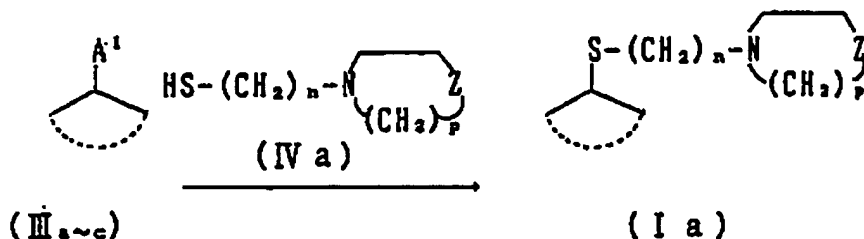
#### Method 1-1

[Synthesis of Compound (Ia) in Compound (I), wherein W is -S-(part 1)]

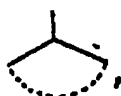


wherein  $X_1$ - $X_2$ ,  $R^A$ ,  $R^B$ ,  $G^A$ ,  $G^B$ ,  $Z$ ,  $g^A$ ,  $g^B$ ,  $n$  and  $p$  have the same significances as described above.

Compound (Ia) can be obtained from Compounds (IIIa to c) and (IVa) according to the following reaction step.



wherein  $A^1$  represents OH,  $OR^5$  or Hal; and



$Z$ ,  $R^5$ , Hal and  $p$  have the same significances as described above.

Compound (Ia) or acid addition salts thereof can be obtained by reacting Compound (IIIa) with 1 to 5 molar equivalents of an appropriate dehydration condensing agent, for example, trifluoroacetic anhydride at from 0°C to room temperature for 1 to 24 hours in an inert solvent such as methylene chloride, chloroform, etc., then adding 1 to 5 molar equivalents of Compound (IVa) or acid addition salts thereof (for example, hydrochloride, hydrobromide, acetate, trifluoroacetate, p-toluenesulfonate, etc.; the same is applied hereafter) to the reaction solution and carrying out the reaction at 0°C to a boiling point of the solvent for 1 to 24 hours, if necessary, in the presence of an appropriate acid catalyst, for example, boron trifluoride diethyl



ether complex.

Likewise, Compound (Ia) or acid addition salts thereof can be obtained by reacting Compound (IIIb) with 1 to 5 molar equivalents of Compound (IVa) or acid addition salts thereof in an inert solvent such as methylene chloride, chloroform, etc., at 0° C to a boiling point of the solvent for 1 to 24 hours, if necessary, in the presence of an appropriate acid catalyst, for example, boron trifluoride diethyl ether complex.

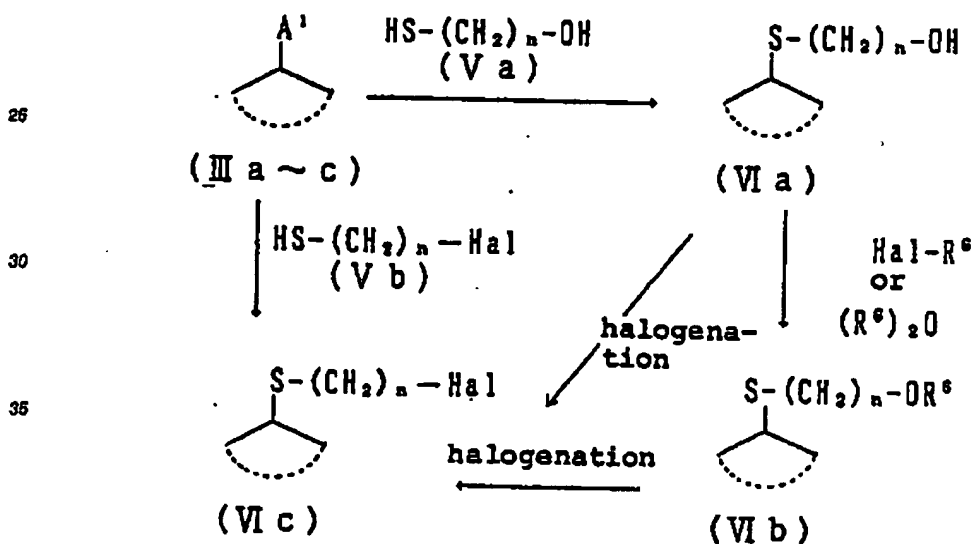
Furthermore, Compound (Ia) or acid addition salts thereof can be obtained by reacting Compound (IIIc) with 1 to 10 molar equivalents of Compound (IVa) or acid addition salts thereof in an inert solvent such as methylene chloride, chloroform, dimethylformamide, etc., at 0° C to a boiling point of the solvent for 1 to 24 hours, if necessary, in the presence of a base such as triethylamine, sodium hydride, etc.

#### Method 1-2

#### [Synthesis of Compound (Ia) (part 2)]

Likewise, Compound (Ia) can be obtained from Compounds (IIIa-c) according to the following reaction steps.

Firstly, Compound (VIb) or (VIc) is prepared from Compounds (IIIa-c) according to the following reaction steps.



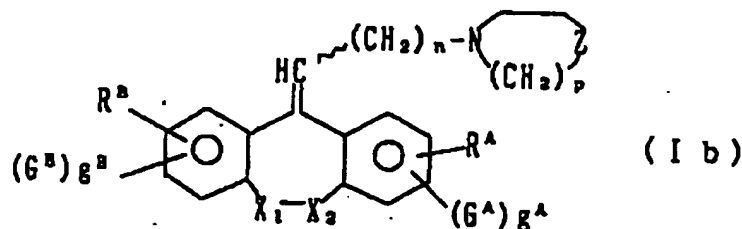
wherein

A<sup>1</sup>, Hal and n have the same significances as described above; and R<sup>6</sup> represents a group capable of being split as OR<sup>6</sup>.

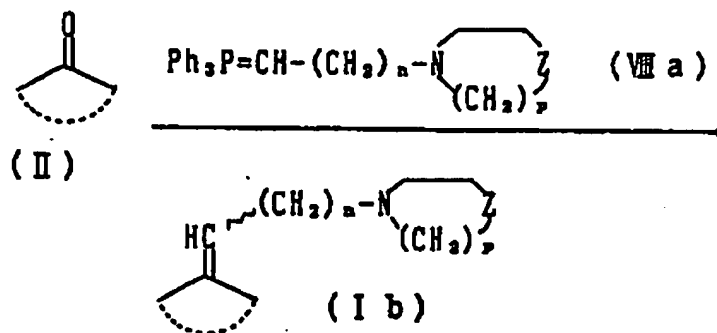
Herein, R<sup>6</sup> means, for example, alkylsulfonyl such as methanesulfonyl, trifluoromethanesulfonyl, etc. and arylsulfonyl such as phenylsulfonyl, p-toluenesulfonyl, etc.

The corresponding Compound (VIa) or (VIc) can be obtained by reacting Compound (IIIa) with 1 to 5 molar equivalents of an appropriate dehydrating and condensing agent, for example, trifluoroacetic anhydride, in an inert solvent such as methylene chloride, chloroform, etc., at a temperature of from 0° C to room temperature for 1 to 24 hours, then adding 1 to 10 molar equivalents of an alcohol (Va) or its halide (Vb) to this reaction solution and carrying out the reaction at a temperature of between at room temperature

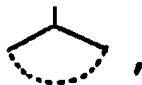




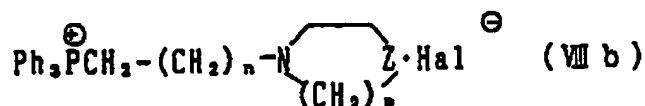
10 wherein  $X_1$ - $X_2$ ,  $R^A$ ,  $R^B$ ,  $G^A$ ,  $G^B$ ,  $z$ ,  $n$ ,  $g^A$ ,  $g^B$  and  $p$  have the same significances as described above.  
Compound (Ib) can be prepared according to the following reaction steps:



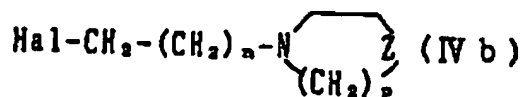
wherein



35 Z, Ph, n and p have the same significances as described above.  
Firstly, a phosphonium salt (VIIIb):



45 wherein Z, Hal, Ph, n and p have the same significances as described above, obtained by reacting triphenylphosphine ( $Ph_3P$ ) with a halide:



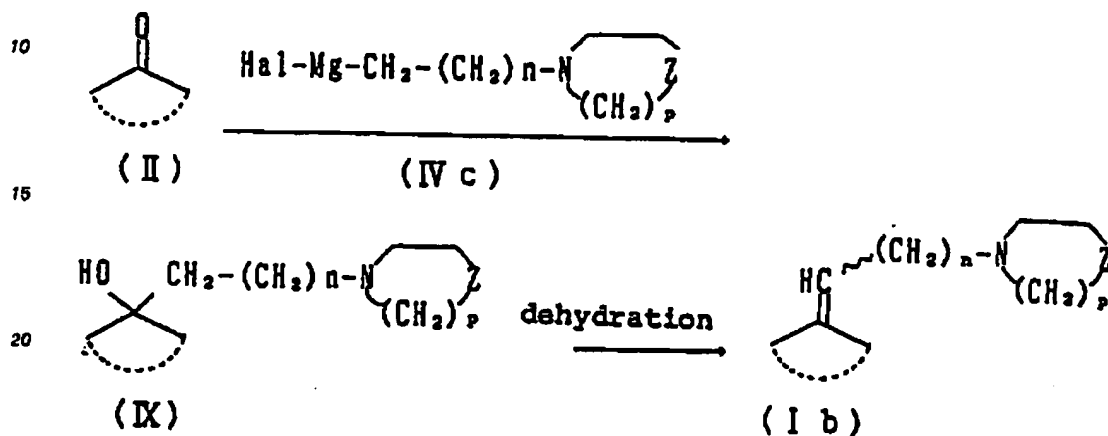
is treated with a molar equivalent of a base such as *n*-butyl lithium, etc. in an inert solvent, for example, tetrahydrofuran, etc., at 0°C to room temperature to give an ylide (VIIIa).

55 Compound (Ib) can be obtained by reacting 1 to 5 molar equivalents, based on Compound (II), of Compound (VIIIa), after or without isolation, with Compound (II) in an inert solvent, for example, tetrahydrofuran, etc., at a temperature of from 0°C and a boiling point of the solvent.

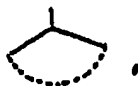
Method 2-2

## [Synthesis of Compound (Ib) (part 2)]

Compound (Ib) can also be obtained from Compound (II) according to the following reaction steps:



wherein



Z, Hal, n and p have the same significances as described above.

Firstly, Compound (II) is reacted with 1 to 5 molar equivalents of Grignard reagent (IVc) in an inert solvent such as tetrahydrofuran, diethyl ether, etc., at a temperature of from 0° C to room temperature, for 1 to 48 hours to give an alcohol (IX).

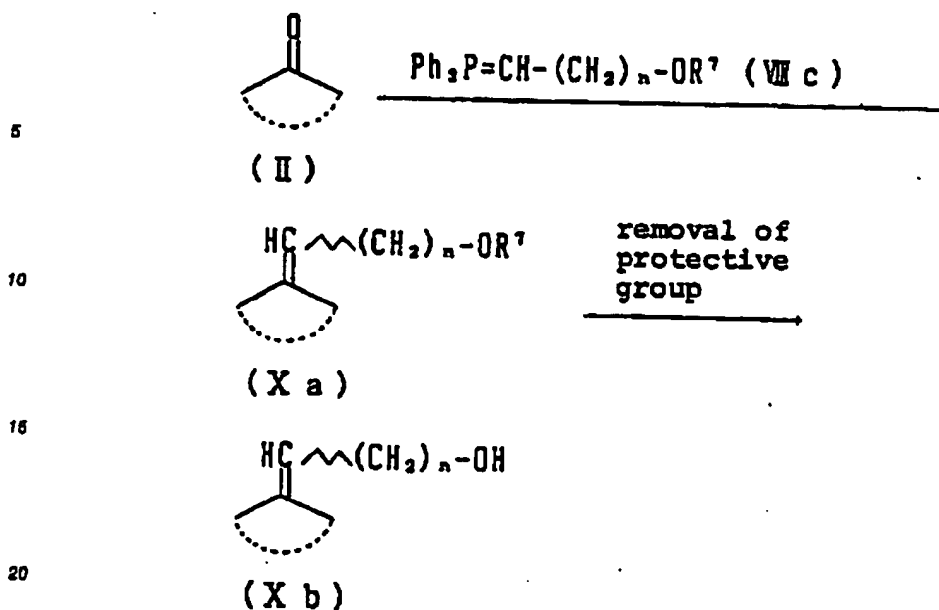
Compound (IVc) can be obtained by reacting corresponding Compound (IVb) with 0.5 to 2 molar equivalents of magnesium in an inert solvent such as tetrahydrofuran, diethyl ether, etc., if necessary and desired, in the presence of a trace amount of iodine, at a temperature of from 0° C to a boiling point of the solvent for 0.5 to 12 hours. The formed Grignard reagent formed is generally used in the following reaction without isolating the same.

The thus obtained Compound (IX) can be subjected to dehydration to give Compound (Ib). For the dehydration, a method which comprises performing the reaction in an inert solvent such as dioxane, etc., in the presence of an acid, for example, p-toluenesulfonic acid, etc., at a temperature of from room temperature to a boiling point of the solvent for 1 to 12 hours, or a method which comprises reacting with a halogenating agent such as thionyl chloride, etc., in an organic base such as pyridine, etc. at a temperature of from 0° C to a boiling point of the solvent for 1 to 12 hours, or the like is adopted.

Method 2-3

## [Synthesis of Compound (Ib) (part 3)]

Firstly, the carbonyl group of Compound (II) is converted into Compound (Xb) according to the following reaction steps:



Ph and n have the same significances as described above, and  $\text{R}^7$  represents a hydroxyl-protecting group.

Herein, as the hydroxyl-protecting group, groups generally used as protective groups for an alcoholic hydroxyl may be used; preferably used is, for example, tetrahydropyranyl or the like.

35 Firstly, an ylide (VIIc) in which hydroxyl is protected by an appropriate protective group (for example, tetrahydropyranyl, etc.) is formed in an inert solvent, for example, tetrahydrofuran [J. Org. Chem., 44, 3760 (1979)].

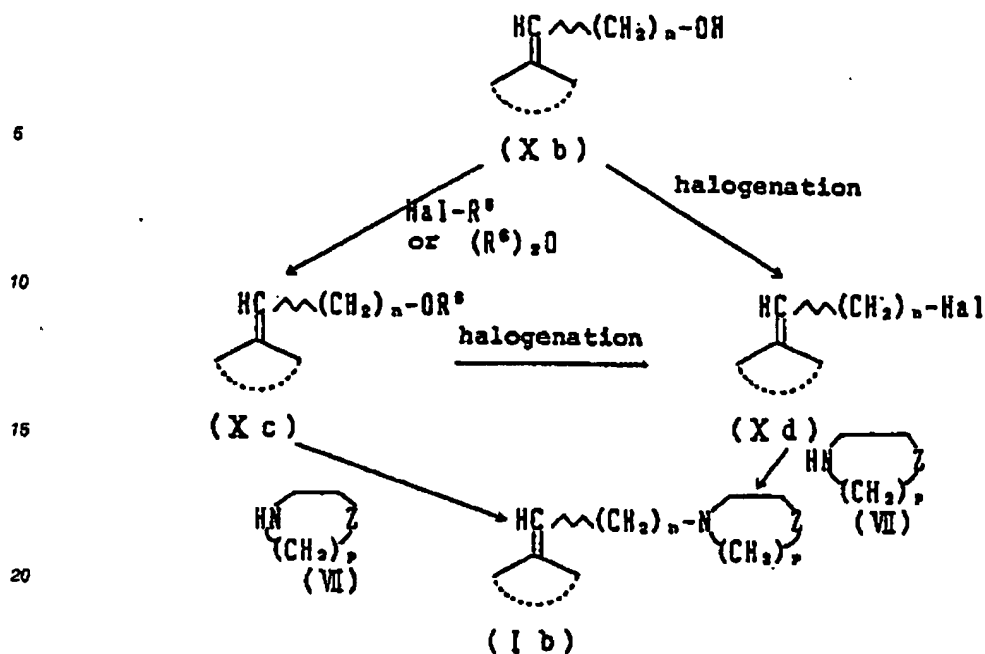
Then, the formed ylide (VIIc) is reacted with 0.2 to 1 molar equivalent of Compound (II) at a temperature of from  $-78^\circ\text{C}$  to a boiling point of the solvent for 1 to 48 hours to give Compound (Xa).

40 Compound (Xa) can be converted into Compound (Xb) by removing the protective group. The removal of protective group can be conducted in a conventional manner; in the case of using, for example, tetrahydropyranyl as a protective group, Compound (Xa) is treated with an acid catalyst such as p-toluenesulfonic acid, hydrochloric acid, etc. in a suitable hydrated solvent such as hydrated dioxane, hydrated tetrahydrofuran, etc., at a temperature of from  $0^\circ\text{C}$  to a boiling point of the solvent for 1 to 24 hours to give Compound (Xb).

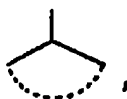
45 Compound (Xb) can be led to Compound (Ib) via Compound (Xc) or Compound (Xd) by the following reaction steps.

50

55



wherein

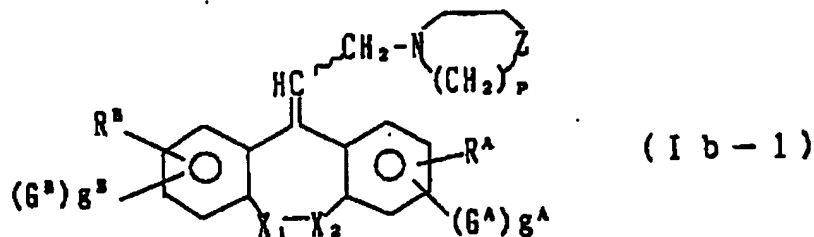


Z, R<sup>5</sup>, Hal, n and p have the same significances as described above.

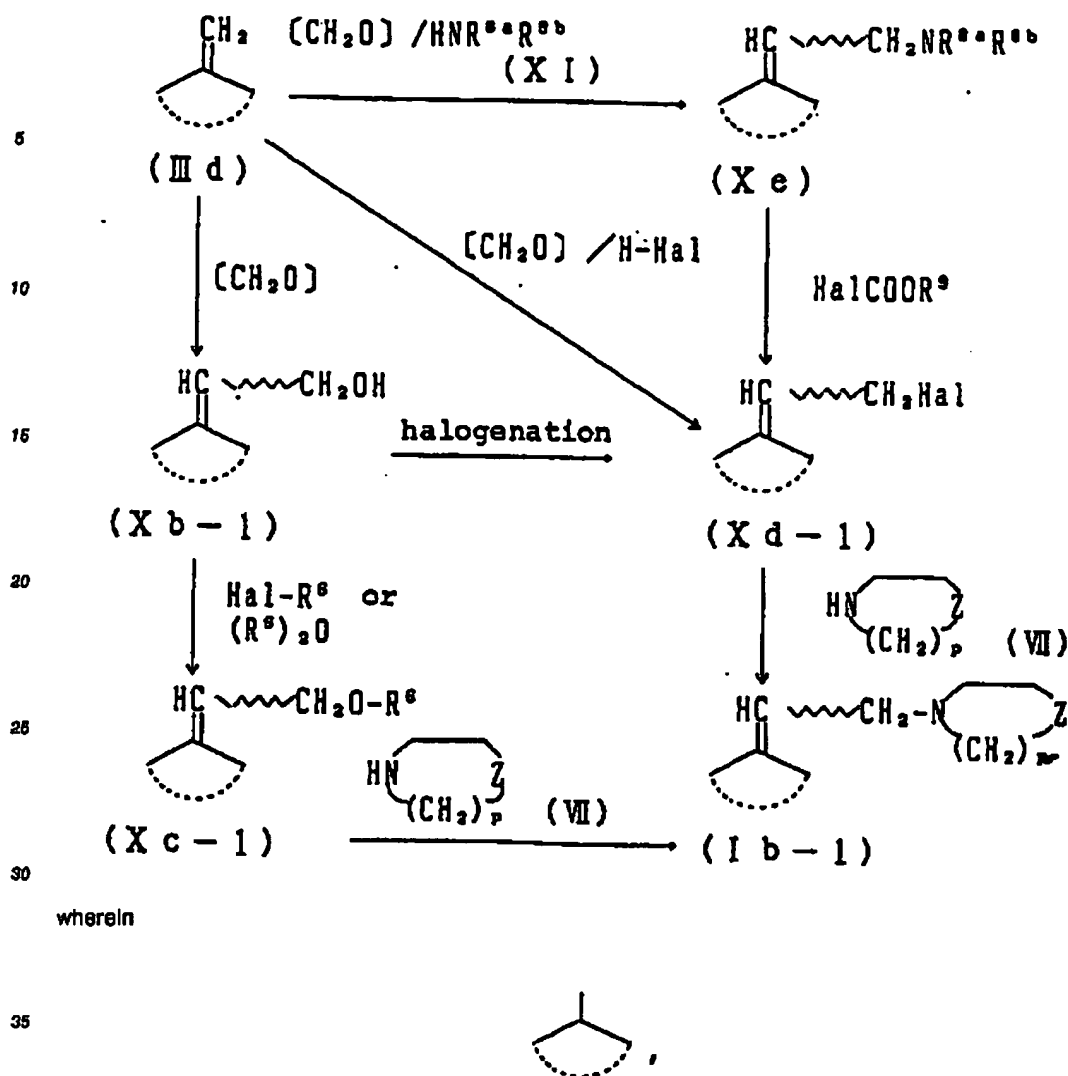
The reactions can be performed in a manner similar to the method for leading from Compound (Via) to Compound (Ia) described in Method 1-2.

#### Method 2-4

[Synthesis of Compound (Ib-1) In Compound (Ib) wherein n is 1]



55 wherein X<sub>1</sub>-X<sub>2</sub>, R<sup>A</sup>, R<sup>B</sup>, G<sup>A</sup>, G<sup>B</sup>, z, g<sup>A</sup>, g<sup>B</sup> and p have the same significances as described above.  
Compound (Ib-1) can be prepared according to the following reaction steps:



40 Z,  $\text{R}^{\text{c}}$ , Hal and p have the same significances as described above;  $[\text{CH}_2\text{O}]$  represents formaldehyde and/or a polymer thereof;  $\text{R}^{\text{aa}}$  and  $\text{R}^{\text{ab}}$ , which may be the same or different, each represents lower alkyl or may be combined to nitrogen adjacent thereto to form a heterocyclic ring and  $\text{R}^{\text{c}}$  represents lower alkyl.

45 Herein, the lower alkyl in the definitions of  $\text{R}^{\text{aa}}$ ,  $\text{R}^{\text{ab}}$  and  $\text{R}^{\text{c}}$  has the same significance as described for the lower alkyl in formula (I). As the heterocyclic ring formed by  $\text{R}^{\text{aa}}$  and  $\text{R}^{\text{ab}}$ , mention may be made of pyrrolidine, piperidine, N-methylpiperazine, morpholine, thiomorpholine, N-methylhomopiperazine and the like.

50 Compound (IIId) is reacted with 1 to 10 molar equivalents of formaldehyde and/or a formaldehyde polymer, for example, paraformaldehyde, either in a hydrohalogenic acid, preferably hydrochloric acid or in an inert solvent, for example, dioxane, saturated with hydrogen chloride and, if necessary and desired, in the presence of a strong acid such as sulfuric acid or trifluoroacetic acid, at a temperature of from room temperature to a boiling point of the solvent, for 1 to 24 hours to give Compound (Xd-1).

Further Compound (Xb-1) can be obtained under almost the same conditions as described above except that no hydrohalogenic acid is added.

55 Furthermore, Compound (Xd-1) can also be obtained as follows. That is, Compound (IIId) is reacted with 1 to 2 molar equivalents of formaldehyde and/or a formaldehyde polymer, for example, paraformaldehyde, and 1 to 3 molar equivalents of a secondary amine (XI) and trifluoroacetic acid, in an inert solvent such as methylene chloride, chloroform, dichloroethane, tetrachloroethane, etc., if necessary and desired, in the presence of acetic acid, at a temperature of from room temperature to a boiling point of the solvent, for 1 to 48 hours to give Compound (Xe) or acid addition salts thereof. Compound (Xe) can be led to Compound

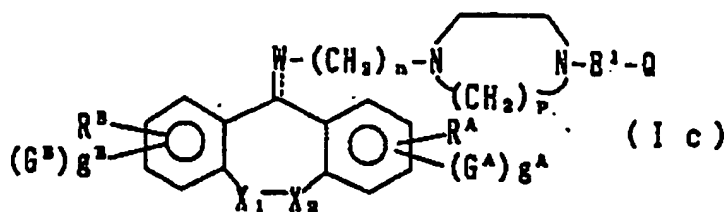
(Xd-1) by reacting with 1 to 10 molar equivalents of a halocarbonate, preferably ethyl chloroformate in an inert solvent such as methylene chloride, chloroform, dichloroethane, tetrachloroethane, etc., if necessary, in the presence of a base such as triethyl amine and sodium acetate between at 0 ° C and a boiling point of the solvent for 1 to 48 hours.

The thus obtained Compounds (Xb-1) and (Xd-1) can be converted into Compound (Ib-1) as in the synthesis of Compound (Ib) from the corresponding Compounds (Xb) and (Xd) described in Method 2-3.

Further Compound (Ib-1) can also be obtained according to the method for obtaining Compound (Xe) from Compound (IId) in which Compound (VII) is used in place of Compound (XI).

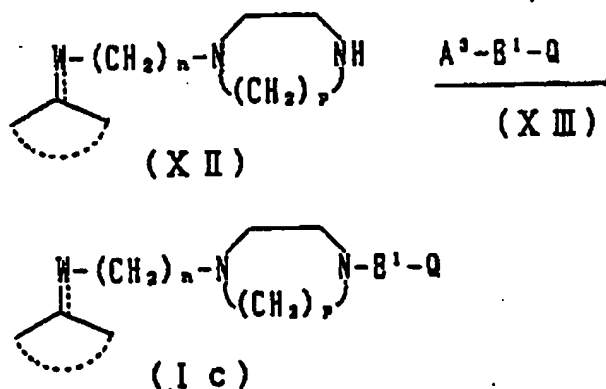
### Method 3

[Synthesis of Compound (Ic) in Compound (I), wherein Z is  $\text{>N-E}^1\text{-Q}$ ]

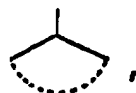


wherein  $\dots$ ,  $X_1$ - $X_2$ ,  $R^A$ ,  $R^B$ ,  $G^A$ ,  $G^B$ ,  $W$ ,  $E^1$ ,  $Q$ ,  $g^A$ ,  $g^B$ ,  $n$  and  $p$  have the same significances as described above.

Compound (Ic) can be prepared from Compounds (XII) and (XIII) obtained in a manner similar to Methods 1 and 2 described above by the following reaction steps.



wherein



$W$ ,  $E^1$ ,  $Q$ ,  $n$  and  $p$  have the same significances as described above and  $A^3$  represents a leaving group in  $\text{-E}^1\text{-Q}$ .

Herein, in the case that  $E^1$  is single bond, leaving group  $A^3$  represents halogen such as chlorine, bromine, iodine, etc. and in the case that  $E^1$  is  $\text{-CO-}$ ,  $A^3\text{-E}^1\text{-Q}$  represents  $\text{HOOC-Q}$  ( $A^3$  is  $\text{OH}$ ) or a carboxylic acid reactive derivative. The carboxylic acid reactive derivative includes an acid halide (acid chloride, acid

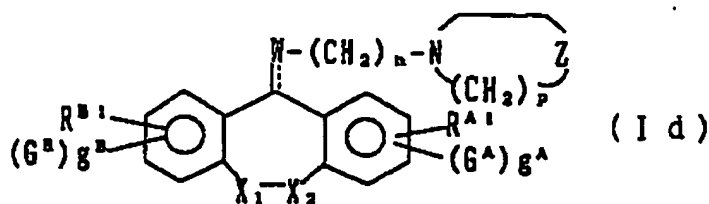


bromide, etc.), an acid anhydride (acid anhydride formed with a dehydrating condensing agent such as N, N'-dicyclohexylcarbodiimide, etc., in the reaction system, commercially available acid anhydrides, etc.), an activated ester (p-nitrophenyl ester, N-hydroxysuccinimide ester, etc.), a mixed acid anhydride (monoethyl carbonate, monoisobutyl carbonate, etc.) and the like. Further in the case that E<sup>1</sup> is -COO-, A<sup>1</sup> represents halogen as described above and in the case that E<sup>1</sup> is -SO<sub>2</sub>-, A<sup>3</sup> represents halogen or -O-SO<sub>2</sub>-Q.

Compound (1c) can be obtained by reacting Compound (XII) or acid addition salts thereof with 1 to 5 molar equivalents of Compound (XIII), either in an inert solvent such as methylene chloride, chloroform, etc., in the presence of a base such as pyridine, etc., or in a basic organic solvent such as pyridine or triethylamine, etc., at a temperature of 0°C to room temperature for 1 to 24 hours.

#### Method 4-1

[Synthesis of Compound (Id) in Compound (I), wherein M is -COOH]



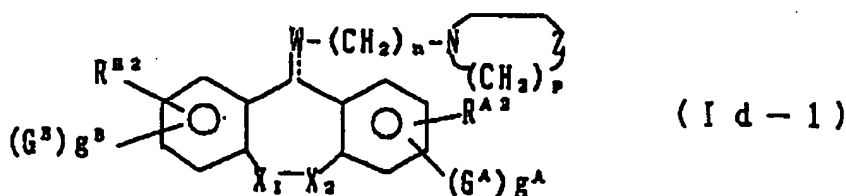
wherein one of R<sup>A1</sup> and R<sup>B1</sup> represents -Y-COOH and the other represents hydrogen; and ....., X<sub>1</sub>-X<sub>2</sub>, G<sup>A</sup>, G<sup>B</sup>, W, Z, Y, n, g<sup>A</sup>, g<sup>B</sup> and p have the same significances as described above.

Compound (Id) can be obtained by hydrolysis of the corresponding carboxylic acid ester.

That is, Compound (Id) can be obtained by subjecting Compound (Ie) [in Compound (I), compound wherein M is -COOR<sup>3c</sup> (wherein R<sup>3c</sup> represents lower alkyl in the definitions for R<sup>3</sup> described above)] synthesized according to Methods 1 to 3, to an appropriate hydrolysis method, for example, by reacting with a molar equivalent to an excess of sodium hydroxide or potassium hydroxide, etc. in a solvent mixture of a lower alcohol such as methanol, ethanol, etc. and water, at a temperature of from room temperature to a boiling point of the solvent for 1 to 48 hours.

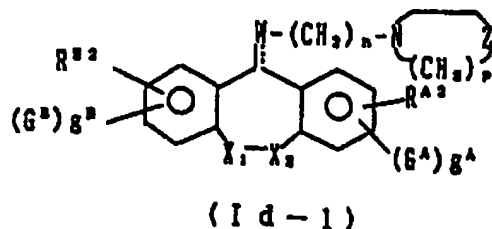
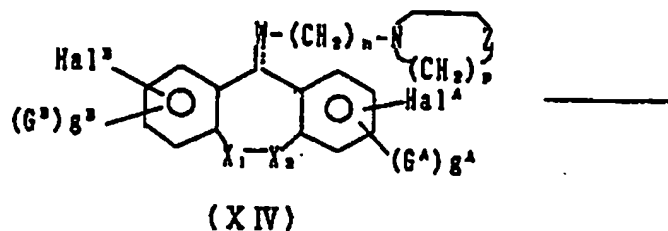
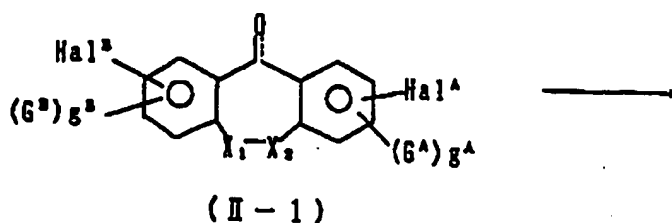
#### Method 4-2

[Synthesis of Compound (Id-1) in Compound (I), wherein Y is single bond]



wherein one of R<sup>A2</sup> and R<sup>B2</sup> represents -COOH and the other represents hydrogen; and ....., X<sub>1</sub>-X<sub>2</sub>, G<sup>A</sup>, G<sup>B</sup>, W, Z, n, g<sup>A</sup>, g<sup>B</sup> and p have the same significances as described above.

Compound (Id-1) can be obtained according to the following reaction steps.



wherein one of Hal<sup>A</sup> and Hal<sup>B</sup> represents Hal and the other represents hydrogen; and ....., X<sub>1</sub>-X<sub>2</sub>, R<sup>A2</sup>, R<sup>B2</sup>, G<sup>A</sup>, G<sup>B</sup>, W, Z, Hal, n, g<sup>A</sup>, g<sup>B</sup> and p have the same significances as described above.

Compound (Id-1) can be obtained by carboxylating Compound (XIV) synthesized from Compound (II-1) in a manner similar to Methods 1 to 3.

Carboxylation can be performed by reacting, for example, Compound (XIV) with 1 molar equivalent of a metallizing agent, e.g., n-butyl lithium, in an inert solvent such as tetrahydrofuran, etc., at a temperature of from -78°C to room temperature, for 10 minutes to 12 hours followed by reacting the resulting reaction mixture with 1 molar equivalent to a largely excessive amount of carbon dioxide at a temperature of from -78°C to room temperature, for 10 minutes to 12 hours. Alternatively, Compound (Id-1) can also be obtained by preparing the corresponding Grignard reagent from Compound (XIV) and magnesium in an inert solvent such as diethyl ether, etc. in a manner similar to Method 2-2 and reacting the reagent with carbon dioxide, and the like method.

#### Method 5

In the methods for production shown by Methods 1 through 4, where groups defined in Compound (I) change under reaction conditions for practicing the method or are inappropriate for practicing the method, the groups may be subjected to conventional means used in organic synthesis chemistry, for example, means for protecting functional groups, means for removing protection, etc. [for example, cf., Green, Protective Groups in Organic Synthesis, John Wiley & Sons Incorporated (1981)], methods for oxidation, reduction, hydrolysis, etc. [for example, cf., SHIN-JIKKEN KAGAKU KOZA, vols. 14 & 15, Maruzen (1977)].

For example, in case that group M is -COOH, 4,4-dimethyloxazoline, etc. are preferably used as a protecting group for -COOH (for example, Japanese Published Unexamined Patent Application No. 10784/1988) in the method in which the corresponding ester is hydrolyzed (cf. Method 4-1 described above) or in the reaction using a Grignard reagent (cf., for example, Method 2-2). Further, a desired compound can be obtained by hydrolyzing (removing a protecting group in) a compound obtained by Methods 1 through 4, etc. in which group -Y-M is -Y'-CH<sub>2</sub>OR<sup>10</sup> [wherein Y' represents a group obtained by removing CH<sub>2</sub> from Y and R<sup>10</sup> represents a protecting group for hydroxyl (e.g., acetyl, tetrahydropyranyl, etc.)] to convert into -Y'-CH<sub>2</sub>OH and oxidizing the compound.

The intermediates and objective compounds in the respective methods described above can be isolated

and purified by purification methods conventionally used in organic synthesis chemistry, for example, filtration, extraction, washing, drying, concentration, recrystallization, various column chromatographies, etc. Further the intermediates may also be provided in the subsequent reaction, without being particularly purified.

5 In Compound (I) obtained by the foregoing methods, compounds wherein W is shown by =CH- include geometrical isomers of E-form and Z-form with respect to stereochemistry. In general, the methods described above give a mixture of these isomers. Isolation and purification of these isomers can be made in a conventional manner in organic synthesis chemistry, for example, column chromatography, recrystallization, etc. It is also possible to isolate the isomers at stages of intermediates (Xa to Xe) by the various  
10 methods described above.

Further, if desired, E- and Z-forms may be isomerized from each other. This can be made by treating each isomer under reflux in, e.g., acetic acid, for 1 to 24 hours, in the presence of an appropriate acid catalyst such as p-toluenesulfonic acid, etc.

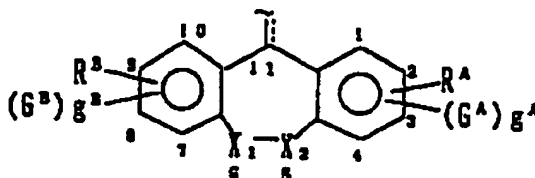
15 In the present invention, Compound (I) includes not only the E/Z isomers described above but also all possible stereoisomers and a mixture thereof.

In case that salts of Compound (I) are desired to obtain, when Compound (I) is obtained in the form of a salt, Compound (I) may be purified as it is.

Further in case that Compound (I) is obtained in a free form, salts may be formed in a conventional manner. Furthermore, Compound (I) and pharmaceutically acceptable salts thereof may also be present in the form of addition products to water or various solvents; these adducts including the pharmaceutically  
20 acceptable salts are also included in the present invention.

Specific examples of Compound (I) obtained by various methods are shown in Table 1.

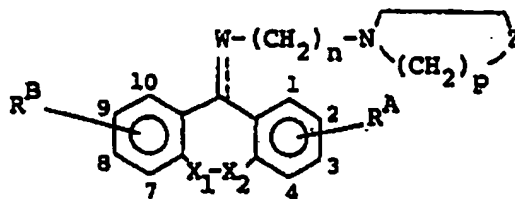
25 Numbering of substitution positions in Table 1 and Table 6 later described does not necessarily harmonize with the correct nomenclature [cf. see (NOTE) below]; but for purpose of simplicity, numbering of the substitution positions is systematically made as illustrated below.

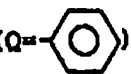
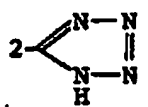

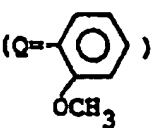


#### (NOTE)



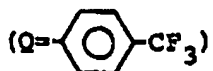
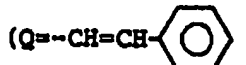

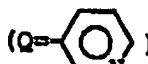
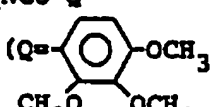

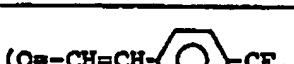

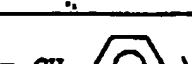
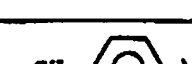
35 In cycloheptene derivatives (wherein  $X_1-X_2$  is  $-CH_2CH_2-$  or  $-CH=CH-$ ), despite the positional number in the general formula above, for example, a substituent on the carbon at the 2-position in the formula above is correctly given as a substituent at the 3-position. However, in the tables, according to the positional numbering in the formula described above,  $-COOH$  on the carbon at the 2-position is indicated to be 2-COOH (correctly 3-COOH).  
40  
45  
50  
55

Table 1

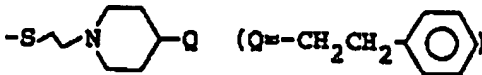
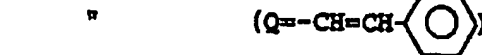
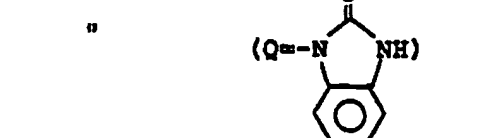
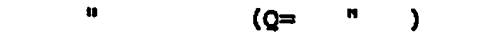
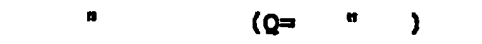

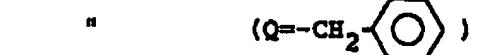
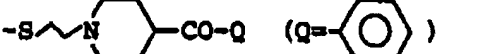
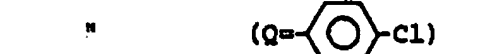

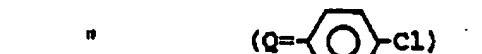
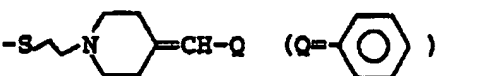


$X_1-X_2$	$W-(CH_2)_n-N-(CH_2)_p-Z$	$R^A / R^B$	Compound No.
$-CH_2O-$	$-S-CH_2-CH_2-N(CH_2)_6-N-Q$ ( $Q=$  )	2-COOCH <sub>3</sub> 2-COOH	1a 1b
"	" ( $Q=$ " )	2-CH <sub>2</sub> COOCH <sub>3</sub> 2-CH <sub>2</sub> COOH	2a 2b
"	" ( $Q=$ " )	2-CH <sub>2</sub> CH <sub>2</sub> COOCH <sub>3</sub> 2-CH <sub>2</sub> CH <sub>2</sub> COOH	3a 3b
"	" ( $Q=$ " )	2-CH(CH <sub>3</sub> )COOCH <sub>3</sub> 2-CH(CH <sub>3</sub> )COOH	4a 4b
"	" ( $Q=$ " )		5
"	" ( $Q=$ " )	3-COOCH <sub>3</sub> 3-COOH	6a 6b
"	" ( $Q=$ " )	9-COOCH <sub>3</sub> 9-COOH	7a 7b
"	" ( $Q=$  )	2-COOCH <sub>3</sub> 2-COOH	8a 8b
"	" ( $Q=$  )	2-COOCH <sub>3</sub> 2-COOH	9a 9b

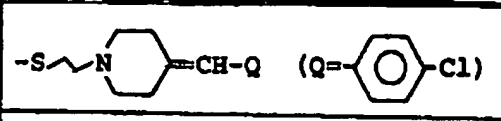
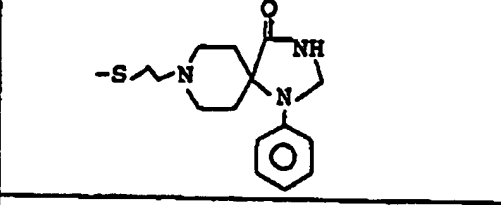
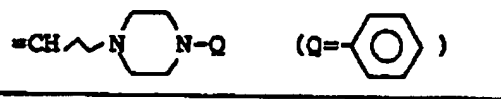
6	$X_1-X_2$	$---W-(CH_2)_n-N-(CH_2)_p-2$	$R^A / R^B$	Compound No.
	$-CH_2O-$	$-S-CH_2-CH_2-N-(CH_2)_4-N-Q$ ( $Q = \text{C}_6\text{H}_4\text{OCH}_3$ )	2-COOCH <sub>3</sub> 2-COOH	10a 10b
10	"	" ( $Q = -CH_2-C_6H_5$ )	2-COOCH <sub>3</sub> 2-COOH	11a 11b
15	"	" ( $Q = -CH_2-C_6H_4-Cl$ )	2-COOCH <sub>3</sub> 2-COOH	12a 12b
	"	" ( $Q = \text{---}$ )	2-C(CH <sub>3</sub> ) <sub>2</sub> COOCH <sub>3</sub> 2-C(CH <sub>3</sub> ) <sub>2</sub> COOH	13a 13b
20	"	" ( $Q = -CH_2-C_6H_4\text{OCH}_3$ )	2-COOCH <sub>3</sub> 2-COOH	14a 14b
25	"	" ( $Q = -CH_2-C_6H_3(O)-$ )	2-COOCH <sub>3</sub> 2-COOH	15a 15b
30	"	" ( $Q = -CH-$ biphenyl)	2-COOCH <sub>3</sub> 2-COOH	16a 16b
35	"	" ( $Q = -CH-$ 2,6-difluorobiphenyl)	2-COOCH <sub>3</sub> 2-COOH	17a 17b
40	"	" ( $Q = -C(C_6H_5)_3$ )	2-COOCH <sub>3</sub> 2-COOH	18a 18b
45	"	" ( $Q = -CH_2CH_2-C_6H_5$ )	2-COOCH <sub>3</sub> 2-COOH	19a 19b
50	"	" ( $Q = -CH_2-CH=CH-C_6H_5$ )	2-COOCH <sub>3</sub> 2-COOH	20a 20b

5	$X_1-X_2$	$-----W-(CH_2)_n-N-(CH_2)_p-Z$	$R^A / R^B$	Compound No.
10	$-CH_2O-$	$-S-CH_2-CH_2-N(CH_2)_6-NSO_2Q$ ( $Q=$  )	2-COOCH <sub>3</sub> 2-COOH	21a 21b
15	"	" ( $Q=$  )	2-COOCH <sub>3</sub> 2-COOH	22a 22b
20	"	" ( $Q=$  )	2-COOCH <sub>3</sub> 2-COOH	23a 23b
25	"	" ( $Q=$  )	2-COOCH <sub>3</sub> 2-COOH	24a 24b
30	"	" ( $Q=$  )	2-COOCH <sub>3</sub> 2-COOH	25a 25b
35	"	" ( $Q=$  )	2-COOCH <sub>3</sub> 2-COOH	26a 26b
40	"	$-S-CH_2-CH_2-N(CH_2)_6-NCO-Q$ ( $Q=$  )	2-COOCH <sub>3</sub> 2-COOH	27a 27b
45	"	" ( $Q=$  )	2-COOCH <sub>3</sub> 2-COOH	28a 28b
50	"	" ( $Q=$  )	2-COOCH <sub>3</sub> 2-COOH	29a 29b
55	"	$-S-CH_2-CH_2-N(CH_2)_6-NCO-O-Q$ ( $Q=$  )	2-COOCH <sub>3</sub> 2-COOH	30a 30b
60	"	" ( $Q=$  )	2-COOCH <sub>3</sub> 2-COOH	31a 31b
65	"	$-S-CH_2-CH_2-N(CH_2)_6-N-Q$ ( $Q=$  )	2-COOCH <sub>3</sub> 2-COOH	32a 32b

6	$X_1-X_2$	$---W-(CH_2)_n-N(CH_2)_p-Z$	$R^A / R^B$	Compound No.
10	$-CH_2O-$	$-S-CH_2-N(CH_2)_4-N-Q$ ( $Q=-CH_2-CH=CH-C_6H_5$ )	2-COOCH <sub>3</sub> 2-COOH	33a 33b
15	"	$-S-CH_2-N(CH_2)_4-NSO_2-Q$ ( $Q=C_6H_4-C_6H_4$ )	2-COOCH <sub>3</sub> 2-COOH	34a 34b
20	"	" ( $Q=-CH=CH-C_6H_5$ )	2-COOCH <sub>3</sub> 2-COOH	35a 35b
25	"	$-S-CH_2-N(CH_2)_4-Q$ ( $Q=C_6H_4$ )	2-COOCH <sub>3</sub> 2-COOH	37a 37b
30	"	" ( $Q=-CH_2-C_6H_5$ )	2-COOCH <sub>3</sub> 2-COOH	38a 38b
35	"	" ( $Q=$ " )	2-CH <sub>2</sub> COOCH <sub>3</sub> 2-CH <sub>2</sub> COOH	39a 39b
40	"	" ( $Q=$ " )	2-CH(CH <sub>3</sub> )COOCH <sub>3</sub> 2-CH(CH <sub>3</sub> )COOH	40a 40b
45	"	" ( $Q=$ " )	2-C(CH <sub>3</sub> ) <sub>2</sub> COOCH <sub>3</sub> 2-C(CH <sub>3</sub> ) <sub>2</sub> COOH	41a 41b
50	"	" ( $Q=$ " )	2-CH <sub>2</sub> CH <sub>2</sub> COOCH <sub>3</sub> 2-CH <sub>2</sub> CH <sub>2</sub> COOH	42a 42b
55	"	" ( $Q=-CH_2-C_6H_4-Cl$ )	2-COOCH <sub>3</sub> 2-COOH	43a 43b
60	"	" ( $Q=$ " )	2-C(CH <sub>3</sub> ) <sub>2</sub> COOCH <sub>3</sub> 2-C(CH <sub>3</sub> ) <sub>2</sub> COOH	44a 44b





$X_1-X_2$	$---W-(CH_2)_n-N-(CH_2)_p-Z$	$R^A / R^B$	Compound No.
-CH <sub>2</sub> O-		2-COOCH <sub>3</sub> 2-COOH	45a 45b
"		2-COOCH <sub>3</sub> 2-COOH	46a 46b
"		2-COOCH <sub>3</sub> 2-COOH	47a 47b
"		2-CH <sub>2</sub> COOCH <sub>3</sub> 2-CH <sub>2</sub> COOH	48a 48b
"		2-C(CH <sub>3</sub> ) <sub>2</sub> COOCH <sub>3</sub> 2-C(CH <sub>3</sub> ) <sub>2</sub> COOH	49a 49b
"		2-COOCH <sub>3</sub> 2-COOH	50a 50b
"		2-COOCH <sub>3</sub> 2-COOH	51a 51b
"		2-COOCH <sub>3</sub> 2-COOH	52a 52b
"		2-COOCH <sub>3</sub> 2-COOH	53a 53b
"		2-COOCH <sub>3</sub> 2-COOH	54a 54b
"		2-COOCH <sub>3</sub> 2-COOH	55a 55b
"		2-COOCH <sub>3</sub> 2-COOH	56a 56b



5	$X_1-X_2$	$---W-(CH_2)_n-N-(CH_2)_p-2$	$R^A / R^B$	Compound No.
	$-CH_2O-$		2-COOCH <sub>3</sub> 2-COOH	57a 57b
10	"		2-COOCH <sub>3</sub> 2-COOH	58a 58b
15	"	"	2-C(CH <sub>3</sub> ) <sub>2</sub> COOCH <sub>3</sub> 2-C(CH <sub>3</sub> ) <sub>2</sub> COOH	59a 59b
20	"		2-COOCH <sub>3</sub> 2-COOH	60a 60b
25	"	" (Q= " )	2-CH <sub>2</sub> COOCH <sub>3</sub> 2-CH <sub>2</sub> COOH	61a 61b
30	"	" (Q=-CH <sub>2</sub> -C <sub>6</sub> H <sub>4</sub> )	2-COOCH <sub>3</sub> 2-COOH	62a 62b
35	"	" (Q= " )	2-CH <sub>2</sub> COOCH <sub>3</sub> 2-CH <sub>2</sub> COOH	63a 63b
40	"	" (Q=-CH <sub>2</sub> -C <sub>6</sub> H <sub>4</sub> -Cl)	2-COOCH <sub>3</sub> 2-COOH	64a 64b
45	"	" (Q= " )	2-C(CH <sub>3</sub> ) <sub>2</sub> COOCH <sub>3</sub> 2-C(CH <sub>3</sub> ) <sub>2</sub> COOH	65a 65b
50	"	" (Q=-CH <sub>2</sub> -C <sub>6</sub> H <sub>4</sub> -OCH <sub>3</sub> )	2-COOCH <sub>3</sub> 2-COOH	66a 66b
55	"	" (Q=-CH <sub>2</sub> -C <sub>6</sub> H <sub>4</sub> -OCH <sub>3</sub> )	2-COOCH <sub>3</sub> 2-COOH	67a 67b

$X_1-X_2$	$---W-(CH_2)_n-\overbrace{N-(CH_2)_p}^Z$	$R^A / R^B$	Compound No.
-CH <sub>2</sub> O-	$=CH \sim N \begin{array}{c} \diagup \quad \diagdown \\ \quad \quad \quad \end{array} N-Q$ $(Q = -CH_2-CH=CH-\text{C}_6\text{H}_5)$	2-COOCH <sub>3</sub>	68a
		2-COOH	68b
"	$=CH \sim N \begin{array}{c} \diagup \quad \diagdown \\ \quad \quad \quad \end{array} NSO_2-Q$ $(Q = \text{C}_6\text{H}_5)$	2-COOCH <sub>3</sub>	69a
		2-COOH	69b
"	$"$ $(Q = \text{C}_6\text{H}_4-F)$	2-COOCH <sub>3</sub>	70a
		2-COOH	70b
"	$"$ $(Q = \text{C}_4\text{H}_3\text{S})$	2-COOCH <sub>3</sub>	71a
		2-COOH	71b
"	$=CH \sim N \begin{array}{c} \diagup \quad \diagdown \\ \quad \quad \quad \end{array} N-Q$ $(Q = \text{C}_6\text{H}_5)$	2-COOCH <sub>3</sub>	72a
		2-COOH	72b
"	$"$ $(Q = \text{C}_6\text{H}_4-F)$	2-COOCH <sub>3</sub>	73a
		2-COOH	73b
"	$"$ $(Q = -CH_2-\text{C}_6\text{H}_5)$	2-COOCH <sub>3</sub>	74a
		2-COOH	74b
"	$"$ $(Q = -CH_2-CH=CH-\text{C}_6\text{H}_5)$	2-COOCH <sub>3</sub>	75a
		2-COOH	75b
"	$"$ $(Q = \text{C}_4\text{H}_3\text{N}_2)$	2-COOCH <sub>3</sub>	76a
		2-COOH	76b
"	$=CH \sim N \begin{array}{c} \diagup \quad \diagdown \\ \quad \quad \quad \end{array} NSO_2-Q$ $(Q = \text{C}_6\text{H}_4-F)$	2-COOCH <sub>3</sub>	77a
		2-COOH	77b
"	$"$ $(Q = -CH=CH-\text{C}_6\text{H}_5)$	2-COOCH <sub>3</sub>	78a
		2-COOH	78b
"	$"$ $(Q = \text{C}_4\text{H}_3\text{S})$	2-COOCH <sub>3</sub>	79a
		2-COOH	79b

5	$X_1-X_2$	$W-(CH_2)_n-N-(CH_2)_p-Z$	$R^A / R^B$	Compound No.
10	$-CH_2O-$	$=CH-CH_2-N(CH_2)_4-NSO_2-Q$ ( $Q = \text{thiophene}$ )	2- $CH_2COOCH_3$ 2- $CH_2COOH$	80a 80b
15	"	" ( $Q = \text{phenyl}$ )	2- $CH(CH_3)COOCH_3$ 2- $CH(CH_3)COOH$	81a 81b
20	"	" ( $Q = \text{phenyl}$ )	2- $C(CH_3)_2COOCH_3$ 2- $C(CH_3)_2COOH$	82a 82b
25	"	" ( $Q = \text{phenyl}$ )	2- $CH_2CH_2COOCH_3$ 2- $CH_2CH_2COOH$	83a 83b
30	"	$=CH-CH_2-N(CH_2)_4-Q$ ( $Q = \text{phenyl}$ )	2- $COOCH_3$ 2- $COOH$	84a 84b
35	"	" ( $Q = -CH_2-\text{phenyl}$ )	2- $COOCH_3$ 2- $COOH$	85a 85b
40	"	" ( $Q = \text{phenyl}$ )	2- $CH_2COOCH_3$ 2- $CH_2COOH$	86a 86b
45	"	" ( $Q = \text{phenyl}$ )	2- $CH(CH_3)COOCH_3$ 2- $CH(CH_3)COOH$	87a 87b
50	"	" ( $Q = \text{phenyl}$ )	2- $C(CH_3)_2COOCH_3$ 2- $C(CH_3)_2COOH$	88a 88b
55	"	" ( $Q = \text{phenyl}$ )	2- $CH_2CH_2COOCH_3$ 2- $CH_2CH_2COOH$	89a 89b
	"	" ( $Q = -CH_2-\text{phenyl}-Cl$ )	2- $COOCH_3$ 2- $COOH$	90a 90b
	"	" ( $Q = \text{phenyl}$ )	2- $C(CH_3)_2COOCH_3$ 2- $C(CH_3)_2COOH$	91a 91b
	"	" ( $Q = -CH_2-\text{phenyl}-OCH_3$ )	2- $COOCH_3$ 2- $COOH$	92a 92b

5	$X_1-X_2$	$---W-(CH_2)_n-N-(CH_2)_p-Z$	$R^A / R^B$	Compound No.
10	$-CH_2O-$	$=CH-CH_2-N(CH_2)_4-Q$ ( $Q = -N(=O)-NH-$  )	2-COOCH <sub>3</sub> 2-COOH	93a 93b
15	"	$=CH-CH_2-N(CH_2)_4-Q$ ( $Q = -C_6H_4-$ )	2-COOCH <sub>3</sub> 2-COOH	94a 94b
20	"	" ( $Q = -CH_2-C_6H_4-$ )	2-COOCH <sub>3</sub> 2-COOH	95a 95b
25	"	" ( $Q = -CH_2-C_6H_4-$ )	2-CH <sub>2</sub> COOCH <sub>3</sub> 2-CH <sub>2</sub> COOH	96a 96b
30	"	" ( $Q = -N(=O)-NH-$  )	2-COOCH <sub>3</sub> 2-COOH	97a 97b
35	"	" ( $Q = -CH_2-C_6H_4-$ )	2-CH <sub>2</sub> COOCH <sub>3</sub> 2-CH <sub>2</sub> COOH	98a 98b
40	$-CH_2S-$	$-S-CH_2-N(CH_2)_4-CH_2-C_6H_4-$	2-COOCH <sub>3</sub> 2-COOH	99a 99b
45	"	$=CH-CH_2-N(CH_2)_4-CH_2-C_6H_4-$	2-COOCH <sub>3</sub> 2-COOH	100a 100b
50	$-CH_2CH_2-$	$-S-CH_2-N(CH_2)_4-CH_2-C_6H_4-$	2-COOCH <sub>3</sub> 2-COOH	101a 101b
55	"	$=CH-CH_2-N(CH_2)_4-NSO_2-$ 	2-COOCH <sub>3</sub> 2-COOH	102a 102b
	$-CH=CH-$	$-S-CH_2-N(CH_2)_4-CH_2-C_6H_4-$	2-COOCH <sub>3</sub> 2-COOH	103a 103b
	"	$=CH-CH_2-N(CH_2)_4-NSO_2-$ 	2-COOCH <sub>3</sub> 2-COOH	104a 104b

The thus prepared Compound (I) possesses a potent TXA<sub>2</sub> antagonizing activity and some of them also possess an antiallergic activity and/or antihistaminic activity. Preferred examples of Compound (I) are shown in Table 2.

Names of the compounds given in Table 2, reference examples and examples later described are indicated by correct nomenclature.

Table 2

	<u>Compound</u>	<u>No.</u>
15	11-[2-(4-Phenyl-1-piperazinyl)ethyl]thio-6,11-dihydrodibenz[b,e]oxepin-2-carboxylic acid	1b
	11-[2-(4-Phenyl-1-piperazinyl)ethyl]thio-6,11-dihydrodibenz[b,e]oxepin-2-acetic acid	2b
20	11-[2-(4-Benzyl-1-piperazinyl)ethyl]thio-6,11-dihydrodibenz[b,e]oxepin-2-carboxylic acid	11b
	11-[2-(4-Chlorobenzyl-1-piperazinyl)ethyl]thio-6,11-dihydrodibenz[b,e]oxepin-2-carboxylic acid	12b
25	2-Methyl-2-[11-[2-(4-chlorobenzyl-1-piperazinyl)-ethyl]thio-6,11-dihydrodibenz[b,e]oxepin-2-yl]-propionic acid	13b
30	11[2-[4-[Bis(4-fluorophenyl)methyl]-1-piperazinyl]-ethyl]thio-6,11-dihydrodibenz[b,e]oxepin-2-carboxylic acid	17b
	11-[2-(4-Benzyl-1-homopiperazinyl)ethyl]thio-6,11-dihydrodibenz[b,e]oxepin-2-carboxylic acid	32b
35	11-[2-(4-Phenyl-1-piperidinyl)ethyl]thio-6,11-dihydrodibenz[b,e]oxepin-2-carboxylic acid	37b
	11-[2-(4-Benzyl-1-piperidinyl)ethyl]thio-6,11-dihydrodibenz[b,e]oxepin-2-carboxylic acid	38b
40	11-[2-(4-Benzyl-1-piperidinyl)ethyl]thio-6,11-dihydrodibenz[b,e]oxepin-2-acetic acid	39b
	2-Methyl-2-[11-[2-(4-benzyl-1-piperidinyl)-ethyl]thio-6,11-dihydrodibenz[b,e]oxepin-2-yl]-propionic acid	41b
45	11-[2-[4-(4-Chlorobenzyl)-1-piperidinyl]ethyl]thio-6,11-dihydrodibenz[b,e]oxepin-2-carboxylic acid	43b
50	2-Methyl-2-[11-[2-[4-(4-chlorobenzyl-1-piperidinyl)-ethyl]thio-6,11-dihydrodibenz[b,e]oxepin-2-yl]-propionic acid	44b

5	11-[2-[4-(2(3H)-Benzimidazol-1-yl)piperidino]-ethyl]thio-6,11-dihydrodibenz[b,e]oxepin-2-carboxylic acid	47b
	11-[2-[4-(2(3H)-Benzimidazol-1-yl)piperidino]-ethyl]thio-6,11-dihydrodibenz[b,e]oxepin-2-acetic acid	48b
10	2-Methyl-2-[11-[2-[4-(2(3H)-benzimidazol-1-yl)-piperidino]ethyl]thio-6,11-dihydrodibenz[b,e]-oxepin-2-carboxylic acid	49b
15	11-[2-[1-Phenyl-1,3,8-triazaspiro[4.5]decan-4-on-8-yl]ethyl]thio-6,11-dihydrodibenz[b,e]oxepin-2-carboxylic acid	58b
	2-Methyl-2-[11-[2-[1-phenyl-1,3,8-triazaspiro[4.5]-decan-4-one-8-yl]ethyl]thio-6,11-dihydrodibenz[b,e]oxepin-2-yl]propionic acid	59b
20	11-[3-(4-Benzyl-1-piperazinyl)propylidene]-6,11-dihydrodibenz[b,e]oxepin-2-carboxylic acid	62b
	11-[3-(4-Benzyl-1-piperazinyl)propylidene]-6,11-dihydrodibenz[b,e]oxepin-2-acetic acid	63b
25	2-Methyl-2-[11-[3-(4-benzyl-1-piperazinyl)propylidene]-6,11-dihydrodibenz[b,e]oxepin-2-yl]propionic acid	64b
	11-[3-[4-(4-Chlorobenzyl-1-piperazinyl)propylidene]-6,11-dihydrodibenz[b,e]oxepin-2-carboxylic acid	65b
30	2-Methyl-2-[11-[3-[4-(4-chlorobenzyl-1-piperazinyl)-propylidene]-6,11-dihydrodibenz[b,e]oxepin-2-yl]-propionic acid	66b
35	11-[2-[4-(4-Thienylsulfonyl)-1-piperazinyl]-ethylidene]-6,11-dihydrodibenz[b,e]oxepin-2-carboxylic acid	79b
	11-[2-[4-(2-Thienylsulfonyl)-1-piperazinyl]-ethylidene]-6,11-dihydrodibenz[b,e]oxepin-2-acetic acid	80b
40	2-[11-[2-[4-(2-Thienylsulfonyl)-1-piperazinyl]-ethylidene]-6,11-dihydrodibenz[b,e]oxepin-2-yl]-propionic acid	81b
45	2-Methyl-2-[11-[2-[4-(2-thienylsulfonyl)-1-piperazinyl]ethylidene]-6,11-dihydrodibenz[b,e]oxepin-2-yl]propionic acid	82b
	3-[11-[2-[4-(2-Thienylsulfonyl)-1-piperazinyl]-ethylidene]-6,11-dihydrodibenz[b,e]oxepin-2-yl]-propionic acid	83b
50	11-[3-(4-Phenyl-1-piperidinyl)propylidene]-6,11-dihydrodibenz[b,e]oxepin-2-carboxylic acid	84b
55	11-[3-(4-Benzyl-1-piperidinyl)propylidene]-6,11-dihydrodibenz[b,e]oxepin-2-carboxylic acid	85b

5	11-[3-(4-Benzyl-1-piperidinyl)propylidene]-6,11-dihydrodibenz[b,e]oxepin-2-acetic acid	86b
	2-Methyl-2-[11-[3-(4-benzyl-1-piperidinyl)-propylidene]-6,11-dihydrodibenz[b,e]oxepin-2-yl]-propionic acid	88b
10	11-[3-[4-(4-Chlorobenzyl)-1-piperidinyl]propylidene]-6,11-dihydrodibenz[b,e]oxepin-2-acetic acid	90b
	2-Methyl-2-[11-[3-[4-(4-chlorobenzyl)-1-piperidinyl)-propylidene]-6,11-dihydrodibenz[b,e]oxepin-2-yl]-propionic acid	91b
15	11-[3-[4-[2(3H)-Benzimidazolone-1-yl]piperidino]-propylidene]-6,11-dihydrodibenz[b,e]oxepin-2-carboxylic acid	93b
20	11-[2-[4-[2(3H)-Benzimidazolone-1-yl]piperidino]-ethylidene]-6,11-dihydrodibenz[b,e]oxepin-2-acetic acid	98b
	11-[2-(4-Benzyl-1-piperidinyl)ethyl]thio-6,11-dihydrodibenzo[b,e]thiepin-2-carboxylic acid	99b
25	11-[3-(4-Benzyl-1-piperidinyl)propylidene]-6,11-dihydrodibenzo[b,e]thiepin-2-carboxylic acid	100b
	5-[2-(4-Benzyl-1-piperidinyl)ethyl]thio-10,11-dihydro-5H-dibenzo[a,d]cyclohepten-3-carboxylic acid	101b
30	5-[2-[4-(2-Thienyl)sulfonyl-1-piperidinyl]ethylidene]-10,11-dihydro-5H-dibenzo[a,d]cyclohepten-3-carboxylic acid	102b
	5-[2-(4-Benzyl-1-piperidinyl)ethyl]thio-5H-dibenzo[a,d]cyclohepten-3-carboxylic acid	103b
35	5-[2-[4-(2-Thienyl)sulfonyl-1-piperidinyl]ethylidene]-5H-dibenzo[a,d]cyclohepten-3-carboxylic acid	104b

40 Next, TXA<sub>2</sub> antagonizing activity of Compound (I) is described below.

#### Test on anti-platelet activity (TXA<sub>2</sub> antagonizing test)

45 Using guinea pig platelet, influence of the compounds in accordance with the present invention on platelet aggregation induced by U-48819 (9,11-dideoxy-9 $\alpha$ ,11 $\alpha$ -methanoepoxyprostaglandin F<sub>2</sub> $\alpha$ ; manufactured by Cayman Chemical Co., Ltd.) which was a TXA<sub>2</sub>/prostaglandin H<sub>2</sub> receptor stimulant was examined.

Male guinea pig (Hartley strain; body weight, 300 to 600 g) was anesthetized by intraperitoneal administration of sodium pentobarbital (30 mg/kg) and blood was collected from the descending aorta of the abdomen with a 1/10 volume of sodium citrate. By centrifugation (KC-70; manufactured by Kubota Co., Ltd.)  
50 at 800 rpm for 10 minutes, platelet rich plasma (PRP) was collected. Platelet aggregation induced by U-48819 (0.5 - 1  $\mu$ M) was determined by photometry [Born, G.V.R. et al., Nature (London), 194, 927-929 (1962)]. A test compound was pretreated before 3 minutes and an ability of inhibiting aggregation was measured. The minimum concentration of inhibiting platelet aggregation by 30% or more was defined as the minimum effective concentration (MEC) of the test compound.

55 The results are shown in Table 3.

#### Antiallergic activity test

Antiallergic activity was examined in accordance with the passive cutaneous anaphylaxis (PCA) test using rats. As test animals, Wistar strain male rats weighing 180 to 220 g were used to collect antiserum and for the PCA test, Wistar strain male rats weighing 120 to 140 g were used.

6

#### A) Preparation of anti-egg white albumin (EWA) rat serum

Anti-EWA rat serum was prepared according to the method of Stotland and Share [Can. J. Physiol. Pharmacol., 52, 1114 (1974)]. That is, 1 mg of EWA was mixed with 20 mg of aluminum hydroxide gel and 0.5 ml of pertussis-diphtheria-tetanus mixed vaccine. The mixture was subcutaneously administered to the rat paw in 4 portions. Fourteen days after, blood was collected from the carotid artery and the serum was isolated and stored in a state frozen at -80 °C. Titer of this antiserum to homologous PCA for 48 hours was 1 : 32.

15

#### B) Homologous PCA test for 48 hours using rats

Three rats were used per group. After the hair at the back was cut, 0.05 ml each of anti-EWA rat serum diluted to 8-fold with physiological saline was intracutaneously injected to each rat at the back in 2 places to passively sensitize the rat. A test compound or its solution (physiological saline solution or CMC solution) was orally administered 47 hours after and 0.5 ml/100 g of 1% Evans Blue physiological saline containing 2 mg of antigen EWA was then intravenously administered to the tail vein an hour after the administration. Thirty minutes after, the animal was bled to death. The skin was peeled off and a leaked dye amount at the blue dyed portion was measured according to the method of Katayama et al. [Microbiol. Immunol., 22, 89 (1978)]. That is, the blue dyed portion was cut off with scissors, put into a test tube charged with 1 ml of 1 N potassium hydroxide and incubated at 37 °C for 24 hours. After 9 ml of a mixture of 0.6 N phosphate : acetone (5 : 13) was added to the system, the mixture was shaken followed by centrifugation at 2500 rpm for 10 minutes. An absorbancy of the supernatant at 620 nm was measured and the leaked dye amount was quantitatively determined by comparing with a calibration curve previously prepared. A value of one rat was obtained from a mean value of the two portions and an inhibition rate of each rat was calculated according to the following equation.

$$\text{Inhibition rate (\%)} = \left( \frac{\text{A mean amount leaked in the group administered with solvent} - \text{A leaked amount in the group administered with test compound}}{\text{A mean amount leaked in the group administered with solvent}} \right) \times 100$$

The case showing that the inhibition rate is 50% or more is judged to be positive in the PCA inhibition activity and the minimum dose in which at least one out of 3 rats was recognized to be positive was defined as the minimum effective dose (MED) of the test compound.

The results are shown in Table 3.

#### Acute toxicity test

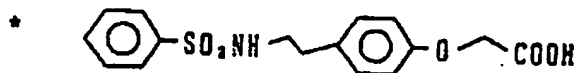
Using three dd strain male mice weighing  $20 \pm 1$  g, a test compound was orally (po; 300 mg/kg) or intraperitoneally (ip; 100 mg/kg) administered. MLD (the minimum lethal dose) was determined by observing the mortality for seven days after administration.

The results are shown in Table 3.



Table 3

Compound No.**	Acute Toxicity (MLD) mg/kg		Antiallergic Activity (MED) mg/kg po	TXA <sub>2</sub> Antagonizing Activity (MEC) µg/ml
	po	ip		
1b'	> 300	> 100	10	3
8b	> 300	> 100	10	1
9b'	> 300	> 100	10	1
11b	> 300	> 100	10	1
25b	> 300	> 100	10	3
37b	> 300	> 100	1	10
38b	> 300	> 100	1	1
E-73b	> 300	> 100	> 100	10
E-75b'	> 300	> 100	100	0.1
E-78b	> 300	> 100	> 100	3
E-79b	> 300	> 100	> 100	30
BM13177* (reference compound)	> 300	> 100	> 100	3



\*\* In Compound No., symbol ' indicates an addition product of the corresponding compound in Tables 1 through 2 to a salt or solvent and, symbols E and Z represents E-form and Z-form, respectively (the same shall apply hereinafter).

As demonstrated in Table 3, Compound (I) and pharmaceutically acceptable salts thereof possess an excellent TXA<sub>2</sub> antagonizing activity and some compounds also possess an antiallergic activity.

Binding study using <sup>3</sup>H-U46619 (TXA<sub>2</sub> receptor binding activities test)

Receptor binding was performed by a modified one of the method of Kattelman et al. [Thrombosis Research, 41, 471 (1988)].

To a washed platelet suspension ( $1 \times 10^8$  platelets) of guinea pig were added 12 nM of tritium-labeled U46619 ( $^3\text{H}$ -U46619; NEN Co., Ltd.) and a test compound (final concentration: 1  $\mu\text{M}$ ) solution. After keeping at 37°C for 20 minutes, the mixture was rapidly filtered through a glass fiber filter paper (GF/C; Whatman). Immediately, the filter paper was washed 5 times with 3 ml of ice-cold 50 mM Tris-hydroxymethylaminomethane buffer (containing 100 mM sodium chloride). The glass fiber filter was transferred to a vial and a scintillator (Ex-H; manufactured by Wako Pure Chemical Industries Co., Ltd.) was added thereto. A radioactivity was determined with a liquid scintillation counter.

It is already known that  $^3\text{H}$ -U46619 is capable of binding with  $\text{TXA}_2$  receptor (see the publication supra). An inhibitory activity of the test compound against this binding means an ability of binding with the  $\text{TXA}_2$  receptor. It is also known that a good relationship exists between this binding activity and the anti-platelet activity ( $\text{TXA}_2$  antagonizing test).

An inhibitory activity (binding activity) of the test compound was calculated by the following equation.

$$(\%) \text{ Inhibition} = \left( 1 - \frac{\text{Amount bound in the presence of a test compound} - \text{Amount of non-specific binding}}{\text{Amount of total binding} - \text{Amount of non-specific binding}} \right) \times 100$$

(Notes)

Amount of the total binding is an amount of  $^3\text{H}$ -U46619-bound radioactivity in the absence of a test compound.

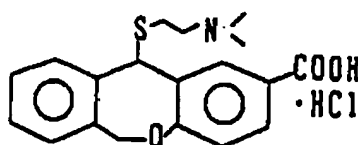
The amount of non-specific binding is used to mean an amount of  $^3\text{H}$ -U46619-bound radioactivity in the presence of 10  $\mu\text{M}$  U-46619.

The amount bound in the presence of a test compound is an amount of  $^3\text{H}$ -U46619-bound radioactivity in the presence of a test compound in various concentrations.

The results are shown in Table 4.

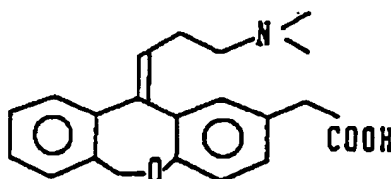
Table 4

Compound No.	% Inhibition
11b	70
25b	78
38b	81
E-73b	85
E-75b'	85
E-78b	83
E-79b	77
Reference compounds:	
Ketotifen	-3
BM13177	31



(Compound A)

-4



(Compound B)

1

As demonstrated in Table 4, Compound (I) and pharmaceutically acceptable salts thereof have an excellent binding ability to TXA<sub>2</sub> receptor.

This inhibitory activity is not observed with hitherto known antiallergic agents, for example, Ketotifen [Merck Index, 10th edition, page 782 (1983)] and known antiallergic agents such as Compound (A) (Japanese Published Unexamined Patent Application No. 28972/1985) and Compound (B) (Japanese Published Unexamined Patent Application No. 10784/1988); one of excellent activity of Compound (I) is TXA<sub>2</sub> receptor binding ability.

#### Binding study using <sup>3</sup>H-Pyrilamine (Histamine H<sub>1</sub> receptor binding)

Receptor binding was performed by the method of Chang et al. [J. Neurochem., 32, 1953 (1979)].

The cerebellum of guinea pig was suspended in a 40-fold amount (V/W) of chilled 50 mM phosphate buffer (pH 7.5) with Polytron Homogenizer (Kinematica Co., Ltd.). The suspension was centrifuged (35,500 x g, 10 minutes). The resulting precipitates were resuspended in an equal amount of fresh buffer followed by centrifugation. A 100-fold amount (V/W) of chilled buffer was added to the final precipitates.

To 1 ml of the membrane suspension adjusted as described above were added 50 µl of tritium-labeled pyrilamine (<sup>3</sup>H-pyrilamine; NEN Co., Ltd.) and 50 µl of a test compound. The mixture was allowed to stand at 25°C for 30 minutes, 4 ml of chilled buffer was added thereto. The mixture was rapidly filtered through a glass fiber filter (GF/C; Whatman). The filter paper was further washed three times with 5 ml of buffer.

Hereafter procedures were performed in a manner similar to the TXA<sub>2</sub> receptor binding test to determine a radioactivity.

An inhibitory activity of the test compound against H<sub>1</sub> receptor was calculated by the following equation.

$$(\%) \text{ Inhibition} = \left( 1 - \frac{\text{Amount bound in the presence of a test compound} - \text{Amount of non-specific binding}}{\text{Amount of total binding} - \text{Amount of non-specific binding}} \right) \times 100$$

(Notes)

Amount of the total binding is an amount of <sup>3</sup>H-pyramine-bound radioactivity in the absence of a test compound.

The amount of non-specific binding is an amount of <sup>3</sup>H-pyramine-bound radioactivity in the presence of 3 μM Astemizole (manufactured by Janesen Pharmaceutica).

The amount bound in the presence of a test compound is an amount of <sup>3</sup>H-pyramine-bound radioactivity in the presence of a test compound in various concentrations.

Inhibitory activity of the test compounds at 10<sup>-6</sup>M are shown in table 5.

Table 5

<u>Compound No.</u>	<u>(%) Inhibition</u>
1b'	68
11b	95
25b	95
38b	92
<b>Reference Compounds:</b>	
Ketotifen	100
BM13177	0
Compound A	88

Compound (I) and pharmaceutically acceptable salts thereof may be administered singly as they are but in general, it is preferably administered as various medical preparations. These medical preparations are used for animal and human beings.

The medical preparation in accordance with the present invention may contain, as an active ingredient, Compound (I) or pharmaceutically acceptable salts thereof singly or as admixture with other optional effective components for different treatment. Further these medical preparations can be produced by optional procedures well known in the pharmaceutical field, by mixing the active ingredient together with one or more pharmaceutically acceptable carriers.

Herein, as the optional effective components for different treatment contained together with Compound (I) or pharmaceutically acceptable salts thereof, mention may be made of, for example, a steroid, a non-steroid antiinflammatory agent, a peripheral analgesic, a leucotriene antagonist, a leucotriene biosynthesis inhibitor, an H<sub>2</sub> receptor antagonist, an antihistaminic agent, a histamine release inhibitor, a bronchodilator, an angiotensin converting enzyme inhibitor, a thromboxane A<sub>2</sub> biosynthesis inhibitor, an H<sup>+</sup>-K<sup>+</sup>ATPase inhibitor, a coronary dilator, a calcium antagonist, a diuretic, a xanthine oxidase inhibitor, a cerebral

circulation improving agent, a cerebral metabolism activator, a cerebral protecting agent, a liver protecting agent, an antiplatelet agent, a thrombolytic agent, an adrenaline  $\alpha$  receptor antagonist, an adrenergic  $\beta$  receptor agent, an adrenaline  $\beta$  receptor antagonist, a serotonin antagonist, a platelet activation factor (PAF) antagonist, an adenosine receptor antagonist, an antihyperlipidemic agent, a cholesterol biosynthesis inhibitor, an immunostimulating agent, an immunosuppressive agent, an anticancer agent, etc.

It is preferred that the most effective route for treatment be selected as a route for administration. Oral or parenteral administration such as intrarectal, topical, intranasal, intraocular, intrabuccal, subcutaneous, intramuscular and intravenous routes, etc. are mentioned.

As the form of administration, there are a capsule, a tablet, a granule, a powder, a syrup, an emulsion, a suppository, an ointment, an eyedrop, a nosedrop, a troche, an aerosol, an injection, etc.

A liquid preparation suited for oral administration, for example, an emulsion and a syrup can be prepared using water; sugars such as sucrose, sorbitol, fructose, etc.; glycols such as polyethylene glycol, propylene glycol, etc.; oils such as sesame oil, olive oil, soybean oil, etc.; antiseptics such as a p-hydroxybenzoic acid ester, etc.; flavors such as strawberry flavor, pepper mint, etc. Further a capsule, a tablet, a powder and a granule, etc. can be prepared using an excipient such as lactose, glucose, sucrose, mannitol, etc.; a disintegrator such as starch, sodium alginate, etc.; a lubricant such as magnesium stearate, talc, etc.; a binder such as polyvinyl alcohol, hydroxypropyl cellulose, gelatin, etc.; a surfactant such as an aliphatic ester, etc.; a plasticizer such as glycerine, etc.

A preparation suited for parenteral administration is composed of a sterile aqueous preparation containing active compounds which are preferably isotonic to blood of recipient. For example, with an injection, a solution for injection is prepared using carriers composed of a saline solution, a glucose solution or a mixture of saline water and glucose solution.

A nasal spray preparation is composed of a purified aqueous solution of the active compounds which contains an antiseptic and an isotonic agent. Such a preparation is adjusted to pH compatible with nasal membrane and to an isotonic state.

An ocular preparation is prepared in a manner similar to the nasal spray, except that pH and isotonic factors are controlled so as to fit those of eyes.

A topical preparation is prepared by dissolving or suspending the active compound in one or more media, for example, a mineral oil, petroleum, a polyvalent alcohol or other bases used for topical medical preparations.

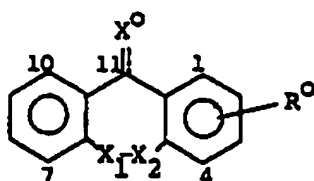
A preparation for rectal administration is provided as a suppository using conventional carriers, for example, cacao fat, hydrogenated fat or hydrogenated fat carboxylic acid, etc.

Further these parenteral preparations may also be added with one or more auxiliary components such as a diluent, a fragrance, an antiseptic (including an antioxidant), an excipient, a disintegrator, a lubricant, a binder, a surfactant, a plasticizer and the like.



Effective dose and number of administration of Compound (I) or pharmaceutically acceptable salts thereof vary depending upon mode of administration, age and body weight of the patient and properties or severity of conditions to be treated. In general, daily dose is 0.01 to 1,000 mg/person and number of administration is once a day, or the dosage may be divided into several ones.

Hereafter, the present invention is described by referring to Reference Examples and Examples below. Intermediates obtained in the following Reference Examples are shown in Table 6.

Table 6



Compound (Reference Example)	$X_1-X_2$	$R^O$	$\text{---}X^O$
a (1)	$-\text{CH}_2\text{O}-$	2-COOCH <sub>3</sub>	=O
b (2)	"	"	-OH
c (3)	"	"	=CHCH <sub>2</sub> Cl
d (4)	"	2-CH <sub>2</sub> COOCH <sub>3</sub>	=O
e (5)	"	"	=CHCH <sub>2</sub> Cl
f (6)	"	2-C(CH <sub>3</sub> ) <sub>2</sub> COOCH <sub>3</sub>	=O
g (7)	"	"	-OH
h (8)	"	2-COOCH <sub>3</sub>	-S-CH <sub>2</sub> CH <sub>2</sub> OH
i (9)	"	"	-S-CH <sub>2</sub> CH <sub>2</sub> Cl
j (10)	"	"	-S-CH <sub>2</sub> CH <sub>2</sub> I
k (11)	"	"	-S-CH <sub>2</sub> CH <sub>2</sub> N(CH <sub>2</sub> ) <sub>4</sub> NH
l (12)	"	2-C(CH <sub>3</sub> ) <sub>2</sub> COOCH <sub>3</sub>	-S-CH <sub>2</sub> CH <sub>2</sub> OH

Compound (Reference Example)	X <sub>1</sub> -X <sub>2</sub>	R <sup>o</sup>	---X <sup>o</sup>
m (13)	-CH <sub>2</sub> O-	2-C(CH <sub>3</sub> ) <sub>2</sub> COOCH <sub>3</sub>	-S ~ Cl
n (14)	"	"	-S ~ I
o (15)	-CH <sub>2</sub> -CH <sub>2</sub> -	2-COOCH <sub>3</sub>	-S ~ OH
p (16)	"	"	-S ~ I
q (17)	-CH=CH-	"	-S ~ OH
r (18)	"	"	-S ~ I
s (19)	-CH <sub>2</sub> O-	"	=CHCH <sub>2</sub> N 
t (20)	"	"	=CHCH <sub>2</sub> CH <sub>2</sub> OSO <sub>2</sub> CH <sub>3</sub>
u (21)	"	"	-S ~ N 

## Reference Example 1

Methyl 11-oxo-8,11-dihydrodibenz[b,e]oxepin-2-carboxylate (Compound a):

A mixture of 348.9 g of methyl p-hydroxybenzoate sodium salt, 402.4 g of phthalide and 200 g of sodium chloride was stirred at 150 °C for 8 hours. After completion of the reaction, the reaction mixture was cooled to room temperature and 4 liters of 10% acetic acid aqueous solution were added thereto. The mixture was allowed to stand at room temperature overnight. After stirring at room temperature for 3 hours, crystals were recovered by filtration. To the crystals was added 6 liters of water. After stirring at room temperature for 30 minutes, the crystals were taken out by filtration. To the crystals was added 3 liters of toluene. The mixture was stirred at room temperature for an hour. The crystals were taken out by filtration and dried by heating under reduced pressure to give 383.9 g of 2-(4-methoxycarbonylphenoxy)methyl benzoate.

IR (KBr tablet, cm<sup>-1</sup>): 3400, 1700, 1610, 1260, 1235

In 5.0 liters of methylene chloride was suspended 392.7 g of the thus obtained 2-(4-methoxycarbonylphenoxy)methyl benzoate and, 266.0 g of trifluoroacetic anhydride was added to the suspension. After stirring at room temperature for an hour, 19.4 g of boron trifluoride ethyl ether complex was added to the mixture followed by stirring at room temperature for 2 hours. The reaction solution was poured into ice water. After fractionation, the organic phase was washed with a diluted sodium hydroxide aqueous solution and then with water, dried over anhydrous magnesium sulfate and concentrated under reduced pressure.

The concentrate was recrystallized from isopropyl ether to obtain 335.3 g of the objective compound as white crystals.

Melting point: 128 - 129 °C

Elemental analysis: as C<sub>15</sub>H<sub>12</sub>O<sub>4</sub>

	C	H
Found (%)	71.55	4.48
Calcd. (%)	71.63	4.51

NMR (CDCl<sub>3</sub>, δ, ppm): 3.84(s, 3H), 5.14(s, 2H), 6.87-8.93(m, 7H)

IR (KBr tablet, cm<sup>-1</sup>): 1710, 1650, 1610, 1250, 1010

## Reference Example 2

Methyl 11-hydroxy-6,11-dihydrodibenz[b,e]oxepin-2-carboxylate (Compound b):

Compound a, 50 g, obtained in Reference Example 1 was suspended in 300 ml of methanol and 6.3 g of sodium borohydride was added to the suspension. The mixture was stirred at room temperature for 2 hours. After completion of the reaction, 10 ml of acetic acid and 300 ml of water were added thereto followed by stirring for 30 minutes. Insoluble matters were taken out by filtration and washed with methanol and then with water. By drying with heating under reduced pressure, 40 g of the objective compound was obtained.

NMR (CDCl<sub>3</sub>, δ, ppm): 2.16(s, 6H), 2.30-2.76(m, 4H), 3.83(s, 3H), 4.83 and 6.40(ABq, J = 12.6 Hz, 2H), 5.01(s, 1H), 6.79-7.93(m, 7H)

IR (neat, cm<sup>-1</sup>): 2950, 1710, 1240, 1015

## Reference Example 3

Methyl (E)-11-(2-chloroethylidene)-6,11-dihydrodibenz[b,e]oxepin-2-carboxylate (Compound c):

4-Methylpiperazine, 30 ml, and 74 g of paraformaldehyde were dissolved in 2 l of tetrachloroethane and 100 ml of trifluoroacetic acid was dropwise added to the solution. After stirring at 60 °C for 2 hours, a solution of 36 g of methyl 11-methylene-6,11-dihydrodibenz[b,e]oxepin-2-carboxylate in 600 ml of tetrachloroethane was dropwise added to the reaction mixture followed by stirring at 80 °C for further 3 hours. The reaction mixture was concentrated to dryness under reduced pressure and 4 N hydrochloric acid aqueous solution was added to the residue to adjust pH to 1. Then, the mixture was washed with diethyl ether. Thereafter, 10 N sodium hydroxide aqueous solution was added to adjust pH to 13. Extraction was performed with 3 l of methylene chloride. After washing with saturated sodium chloride aqueous solution and drying over anhydrous sodium sulfate, the solvent was distilled off under reduced pressure. The resulting residue was subjected to silica gel column chromatography (eluent; hexane : ethyl acetate : triethylamine = 5 : 5 : 1) to give 44 g of methyl 11-{2-(4-methyl-1-piperazinyl)ethylidene}-6,11-dihydrodibenz[b,e]oxepin-2-carboxylate as colorless oil.

MS (m/z): 378 (M<sup>+</sup>)

NMR (CDCl<sub>3</sub>, δ, ppm): 2.24(s, 3H), 2.45(s, 6H), 2.94-3.32(m, 2H), 3.84(s, 3H), 5.22(bs, 2H), 5.85 and 6.22(t, J = 6.8 Hz, 1H), 6.66-8.07(m, 7H)

E-form compound, 21.5 g, isolated from the Z/E mixture described above in a conventional manner and 23.5 g of sodium acetate were suspended in 250 ml of dichloroethane and, 27.1 ml of ethyl chloroformate was dropwise added to the suspension. After completion of the dropwise addition, the mixture was stirred at room temperature for an hour and the solvent was distilled off under reduced pressure. The residue was extracted with 400 ml of ethyl acetate. After washing with saturated sodium chloride aqueous solution and drying over anhydrous sodium sulfate, the solvent was distilled off under reduced pressure. The obtained crude product was recrystallized from isopropanol to give 14.3 g of the objective compound as white crystals.



Melting point: 134 - 135° C

Elemental analysis: as C<sub>18</sub>H<sub>15</sub>ClO<sub>3</sub>

	C	H
Found (%)	68.55	4.77
Calcd. (%)	68.68	4.80

NMR (CDCl<sub>3</sub>, δ, ppm): 3.80(s, 3H), 4.16(d, J=8.1Hz, 2H), 4.88(bs, 1H), 5.57(bs, 1H), 6.31(t, J=8.1Hz, 1H), 6.79-8.04(m, 7H)

#### Reference Example 4

Methyl 11-oxo-6,11-dihydrodibenz[b,e]oxepin-2-acetate (Compound d):

The corresponding starting material was used, treated in a manner similar to Reference Example 1 and recrystallized from methanol to give the objective compound as light yellow crystals.

Melting point: 75 - 76° C

#### Reference Example 5

Methyl (E)-11-(2-chloroethylidene)-6,11-dihydrodibenz[b,e]oxepin-2-acetate (Compound e):

The corresponding starting material was used, treated in a manner similar to Reference Example 3 and recrystallized from isopropanol to give the objective compound as white crystals.

Melting point: 127 - 128° C

Elemental analysis: as C<sub>19</sub>H<sub>17</sub>ClO<sub>3</sub>

	C	H
Found (%)	69.37	5.40
Calcd. (%)	69.41	5.21

NMR (CDCl<sub>3</sub>, δ, ppm): 3.55(s, 2H), 3.69(s, 3H), 4.14 (d, J=8.1Hz, 2H), 4.7-5.4(m, 2H), 6.23(t, J=8.1Hz, 1H), 6.74(d, J=8.1Hz, 1H), 6.95-7.4(m, 6H)

IR (KBr tablet, cm<sup>-1</sup>): 1731, 1487, 1258, 1137, 1004

#### Reference Example 6

Methyl 2-methyl-2-[(11-oxo-6,11-dihydrodibenz[b,e]oxepin)-2-yl]propionate (Compound f):

Compound d, 30 g, obtained in Reference Example 4 was suspended in 300 ml of dimethylformamide and 30 g of potassium hydroxide as powder and then 27 ml of methyl iodide were added to the suspension. The mixture was stirred at room temperature for 2 hours. The solvent was distilled off under reduced pressure and the residue was extracted with 500 ml of ethyl acetate. The extract was washed successively with saturated sodium bicarbonate aqueous solution and saturated sodium chloride aqueous solution. After drying over anhydrous magnesium sulfate, the solvent was distilled off under reduced pressure. The obtained residue was subjected to silica gel column chromatography (eluent: hexane : ethyl acetate = 10 : 1) for purification to give 3.6 g of the objective compound as colorless oil.

NMR (CDCl<sub>3</sub>, δ, ppm): 1.62(s, 6H), 3.68(s, 3H), 5.17 (s, 2H), 7.01(d, J=8.8Hz, 1H), 7.2-7.9(m, 5H), 8.22(d, J=2.6Hz, 1H)

IR (neat,  $\text{cm}^{-1}$ ): 2874, 1729, 1648, 1597, 1490, 1016

#### Reference Example 7

Methyl 2-methyl-2-[(11-hydroxy-6,11-dihydrodibenz[b,e]oxepin)-2-yl]propionate (Compound g):

The objective compound was obtained as colorless oil from Compound f obtained in Reference Example 6 in a manner similar to Reference Example 2.

NMR ( $\text{CDCl}_3$ ,  $\delta$ , ppm): 1.56(s, 6H), 3.62(s, 3H), 4.97 and 5.90(ABq,  $J=13.1\text{Hz}$ , 2H), 5.61(s, 1H), 6.87 (d,  $J=8.4\text{Hz}$ , 1H), 7.06-7.40(m, 6H)

IR (neat,  $\text{cm}^{-1}$ ): 2974, 1728, 1497, 1010

#### Reference Example 8

Methyl 11-(2-hydroxyethyl)thio-6,11-dihydrodibenz[b,e]oxepin-2-carboxylate (Compound h):

In 400 ml of methylene chloride was suspended 40.0 g of Compound b obtained in Reference Example 2. Trifluoroacetic anhydride, 21.0 ml, was added to the suspension followed by stirring at room temperature for an hour. Then, 10.7 ml of 2-mercaptoethanol was added to the mixture followed by stirring for further 4 hours. After 100 ml of methylene chloride was added to the reaction mixture, the mixture was washed with saturated aqueous sodium chloride solution. After drying over anhydrous magnesium sulfate, the solvent was distilled off under reduced pressure. The obtained crude product was recrystallized from toluene to give 37.6 g of the objective compound as white crystals.

Melting point:  $128-130^\circ\text{C}$

Elemental analysis: as  $\text{C}_{18}\text{H}_{18}\text{O}_4\text{S}$

	C	H
Found (%)	65.28	5.55
Calcd. (%)	65.43	5.49

NMR ( $\text{CDCl}_3$ ,  $\delta$ , ppm): 2.66(dt,  $J=2.1$ , 6.0Hz, 2H), 3.69 (t,  $J=5.9\text{Hz}$ , 2H), 3.89(s, 3H), 4.91 and 6.43(ABq,  $J=12.7\text{Hz}$ , 2H), 5.09(s, 1H), 6.82-7.98(m, 7H)

IR (KBr tablet,  $\text{cm}^{-1}$ ): 3420, 1708, 1683, 1610, 1437, 1318, 1008

#### Reference Example 9

Methyl 11-(2-chloroethyl)thio-6,11-dihydrodibenz[b,e]oxepin-2-carboxylate (Compound I):

Compound h, 20.0 g, obtained in Reference Example 8 was dissolved in 200 ml of dimethylformamide and, 12 ml of 2,4,6-chollidine and 4.0 g of lithium chloride were added to the solution. Under ice cooling, 5.4 ml of methanesulfonyl chloride was dropwise added to the mixture.

After stirring at room temperature overnight, the solvent was distilled off under reduced pressure. The residue was extracted with ethyl acetate. The extract was washed successively with 1 N hydrochloric acid aqueous solution and saturated sodium chloride aqueous solution. After drying over anhydrous magnesium sulfate, the solvent was distilled off under reduced pressure. The obtained residue was purified by silica gel column chromatography (eluent; hexane : ethyl acetate = 5 : 1) and crystallized from hexane to give 20.8 g of the objective compound.

Melting point:  $100-102^\circ\text{C}$

Elemental analysis: as  $\text{C}_{18}\text{H}_{17}\text{ClO}_2\text{S}$

	C	H
Found (%)	61.77	4.80
Calcd. (%)	61.97	4.91

NMR (CDCl<sub>3</sub>,  $\delta$ , ppm): 2.54-3.62(m, 4H), 3.84(s, 3H), 5.04(s, 1H), 4.87 and 6.37(ABq, J = 13.2Hz, 2H), 6.78-8.12(m, 7H)

IR (CHCl<sub>3</sub>, cm<sup>-1</sup>): 1714, 1611, 1322, 1295, 1132, 1008

#### Reference Example 10

Methyl 11-(2-iodoethyl)thio-6,11-dihydrodibenz[b,e]oxepin-2-carboxylate (Compound j):

Compound i, 10.1 g, obtained in Reference Example 9 was dissolved in 150 ml of acetonitrile and, 14.8 g of sodium iodide was added to the solution. The mixture was heated under reflux for 5 hours. After allowing it to stand for cooling, the reaction mixture was extracted with ethyl acetate and the extract was washed twice with saturated sodium chloride aqueous solution. After drying over anhydrous magnesium sulfate, the solvent was distilled off under reduced pressure. The residue was purified by silica gel column chromatography (eluent; hexane : ethyl acetate = 10 : 1) and further solidified with hexane to give 3.1 g of the objective compound as white solid.

Melting point: 111 - 113 °C

Elemental analysis: as C<sub>18</sub>H<sub>17</sub>IO<sub>3</sub>S

	C	H
Found (%)	49.08	3.71
Calcd. (%)	49.10	3.89

NMR (CDCl<sub>3</sub>,  $\delta$ , ppm): 2.68-3.22(m, 4H), 3.88(s, 3H), 5.08(s, 1H), 4.91 and 6.37(ABq, J = 13.2Hz, 2H), 6.78-8.08(m, 7H)

IR (CHCl<sub>3</sub>, cm<sup>-1</sup>): 1714, 1611, 1295, 1120, 1008

#### Reference Example 11

Methyl [2-(1-piperazinyl)ethyl]thio-6,11-dihydrodibenz[b,e]oxepin-2-carboxylate (Compound k):

Compound j, 7.0 g, obtained in Reference Example 10 and 6.8 g of piperazine were heated under reflux for 8 hours in 300 ml of ethanol. The solvent was distilled off under reduced pressure. 4 N hydrochloric acid aqueous solution was added to the residue to adjust pH to 3 followed by washing with diethyl ether. Next, 10 N sodium hydroxide aqueous solution was added to adjust pH to 13. Extraction was performed with 500 ml of ethyl acetate. After washing successively with saturated sodium bicarbonate aqueous solution and saturated sodium chloride aqueous solution, the extract was dried over anhydrous sodium sulfate. The solvent was distilled off under reduced pressure to give 5.9 g of the objective compound as light yellow oil. The product was provided for the following reaction without further purification.

NMR (CDCl<sub>3</sub>,  $\delta$ , ppm): 2.19-3.33(m, 12H), 3.84(s, 3H), 5.08(s, 1H), 4.86 and 8.44(ABq, J = 12.7Hz, 2H), 6.78-8.12(m, 7H)

MS (m/z): 402 (M<sup>+</sup>)

#### Reference Example 12

Methyl 2-methyl-2-[11-(2-hydroxyethyl)thio-6,11-dihydrodibenz[b,e]oxepin]-2-yl]propionate (Compound l):

The objective compound was obtained as colorless oil in a manner similar to Reference Example 8 using Compound g obtained in Reference Example 7.

NMR (CDCl<sub>3</sub>, δ, ppm): 1.56(s, 6H), 2.64(dt, J=2.2, 6.0Hz, 2H), 3.65(s, 3H), 4.85 and 6.27(ABq, J= 13.1Hz, 2H), 5.01(s, 1H), 6.7-7.4(m, 7H)

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#### Reference Example 13

10 Methyl 2-methyl-2-[[11-(2-chloroethyl)thio-6,11-dihydrodibenz[b,e]oxepin]-2-yl]propionate (Compound m):

The objective compound was obtained as colorless oil in a manner similar to Reference Example 9 using Compound l obtained in Reference Example 12.

15 NMR (CDCl<sub>3</sub>, δ, ppm): 1.67(s, 6H), 2.6-2.9(m, 2H), 3.2-3.4(m, 2H), 3.65(s, 3H), 4.85 and 6.23(ABq, J= 13.1Hz, 2H), 5.04(s, 1H), 6.7-7.4(m, 7H)

#### Reference Example 14

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Methyl 2-methyl-2-[[11-(2-iodoethyl)thio-6,11-dihydrodibenz[b,e]oxepin]-2-yl]propionate (Compound n):

The objective compound was obtained as colorless oil in a manner similar to Reference Example 10 using Compound m obtained in Reference Example 13.

25 NMR (CDCl<sub>3</sub>, δ, ppm): 1.56(s, 6H), 3.64(s, 3H), 5.27 (s, 1H)

#### Reference Example 15

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Methyl 5-(2-hydroxyethyl)thio-10,11-dihydro-5H-dibenzo[a,d]cyclohepten-3-carboxylate (Compound o):

In 15.3 ml of methylene chloride was suspended 0.51 g of methyl 5-hydroxy-10,11-dihydro-5H-dibenzo-  
[a,d]cyclohepten-3-carboxylate. Under ice cooling, 0.27 ml of trifluoroacetic anhydride was added to the  
35 suspension followed by stirring at room temperature for an hour. Then, 0.14 ml of 2-mercaptoethanol was added to the mixture followed by stirring for further an hour. After 10 ml of methylene chloride was added to the reaction mixture, the mixture was washed with saturated aqueous sodium chloride solution. After drying over anhydrous magnesium sulfate, the solvent was distilled off under reduced pressure. The obtained residue was purified by silica gel column chromatography (eluent; hexane : ethyl acetate = 1 : 1) and  
40 recrystallized from toluene to give 0.51 g of the objective compound as light yellow crystals.

Melting point: 107 - 109 °C

NMR (CDCl<sub>3</sub>, δ, ppm): 2.51-2.65(m, 3H), 2.75-3.03(m, 2H), 3.56-3.98(m, 4H), 3.87(s, 3H), 5.16(s, 1H), 7.14-7.25(m, 5H), 7.82(dd, J= 1.5, 7.9Hz, 1H), 7.91(s, 1H)

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#### Reference Example 16

50 Methyl 5-(2-iodoethyl)thio-10,11-dihydro-5H-dibenzo[a,d]cyclohepten-3-carboxylate (Compound p):

Compound o 0.46 g, obtained in Reference Example 15 and 1.10 g of triphenyl phosphine were suspended in 9.2 ml of benzene. While stirring under ice cooling, 0.65 ml of diethyl azodicarboxylate was added to the suspension followed by stirring at room temperature for 10 minutes. Then, 0.26 ml of iodomethane was added to the mixture followed by stirring at room temperature for 30 minutes. After 10 ml  
55 of ethyl acetate and 10 ml of water were added thereto, the mixture was extracted with ethyl acetate. The extract was washed with saturated aqueous sodium chloride solution. After drying over anhydrous magnesium sulfate, the solvent was distilled off under reduced pressure. The obtained residue was purified by silica gel column chromatography (eluent; hexane : ethyl acetate = 3 : 1) to give 0.57 g of the objective

compound as pale yellow oil.

NMR (CDCl<sub>3</sub>,  $\delta$ , ppm): 2.70-3.08(m, 8H), 3.83-3.81(m, 2H), 3.89(s, 3H), 5.17(s, 1H), 7.17-7.32(m, 5H), 7.84(dd, J = 1.7, 7.8Hz, 1H), 7.92(s, 1H)

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#### Reference Example 17

Methyl 5-(2-hydroxyethyl)thio-5H-dibenzo[a,d]cyclohepten-3-carboxylate (Compound q):

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The corresponding starting material was used, treated in a manner similar to Reference Example 15 and recrystallized from toluene to give the objective compound as pale yellow crystals.

Melting point: 130 - 132 °C

NMR (CDCl<sub>3</sub>,  $\delta$ , ppm): 2.30-2.50(m, 3H), 3.51(t, J = 6.2Hz, 2H), 3.89(s, 3H), 5.31(s, 1H), 7.00(s, 1H), 7.02(s, 1H), 7.09-7.41(m, 5H), 7.91(dd, J = 1.5, 7.9Hz, 1H), 8.01(s, 1H)

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#### Reference Example 18

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Methyl 5-(2-iodoethyl)thio-5H-dibenzo[a,d]cyclohepten-3-carboxylate (Compound r):

Compound q obtained in Reference Example 17 was used and treated in a manner similar to Reference Example 16 to give the objective compound as pale yellow oil.

NMR (CDCl<sub>3</sub>,  $\delta$ , ppm): 2.57-3.07(m, 4H), 3.92(s, 3H), 5.31(s, 1H), 7.01(s, 1H), 7.02(s, 1H), 7.09-7.41 (m, 5H), 7.91(dd, J = 1.5, 7.9Hz, 1H), 8.01(s, 1H)

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#### Reference Example 19

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Methyl (E)-11-[2-(1-piperazinyl)ethyl]thio-6,11-dihydrodibenz[b,e]oxepin-2-acetate (Compound s):

The objective compound was obtained as colorless oil from Compound c (E-form) obtained in Reference Example 3 in a manner similar to Reference Example 11.

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NMR (CDCl<sub>3</sub>,  $\delta$ , ppm): 1.95-3.0(m, 7H), 3.10(d, J = 7.0Hz, 2H), 3.83(s, 3H), 4.75-5.5(m, 2H), 6.20(t, J = 7.0Hz, 1H), 6.71(d, J = 8.5Hz, 1H), 6.95-7.45(m, 4H), 7.73(dd, J = 2.2, 8.5Hz, 1H), 7.99(d, J = 2.2Hz, 1H)

#### Reference Example 20

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Methyl (E,Z)-11-[3-(methylsulfonyl)oxy]propylidene-6,11-dihydrodibenz[b,e]oxepin-2-carboxylate (Compound t):

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[3-[(Tetrahydro-2H-pyran-2-yl)oxy]propyl]triphenylphosphonium bromide, 40.0 g, was suspended in 250 ml of tetrahydrofuran. Under ice cooling in nitrogen atmosphere, 50 ml of n-butyl lithium/hexane solution (1.6 N) was dropwise added to the suspension. After stirring at room temperature for further an hour, 15.0 g of Compound a obtained in Reference Example 1 was added thereto followed by stirring at room temperature for 12 hours. After 50 ml of water was added to the reaction mixture, the mixture was extracted with 1 L of ethyl acetate. After washing thrice with saturated sodium chloride aqueous solution and drying over anhydrous sodium sulfate, the solvent was distilled off under reduced pressure. The obtained residue was dissolved in 500 ml of dioxane and 200 ml of water and 1.0 g of p-toluenesulfonic acid were added to the solution followed by heating under reflux for an hour. The mixture was concentrated under reduced pressure and the obtained residue was extracted with 1 L of ethyl acetate. After washing successively with saturated sodium bicarbonate aqueous solution and saturated sodium chloride aqueous solution, the extract was dried over anhydrous sodium sulfate. The solvent was distilled off under reduced pressure. The resulting residue was subjected to silica gel column chromatography (eluent: toluene : ethyl acetate = 1 :

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1) to give 9.8 g of methyl (E,Z)-11-(3-hydroxy)propylidene-6,11-dihydrodibenz[b,e]oxepin-2-carboxylate. A ratio of E/Z was approximately 3 : 7 by NMR.

The obtained product, 4.8 g, mainly composed of the Z-form and 10.0 g of p-toluenesulfonic acid were stirred in 250 ml of acetic acid at 100 °C for 42 hours. After completion of the reaction, the solvent was distilled off under reduced pressure and 200 ml of methanol was added to the residue followed by heating under reflux for 3 hours. The solvent was distilled off under reduced pressure and the residue was extracted with 500 ml of ethyl acetate. After washing with saturated sodium bicarbonate aqueous solution and drying over anhydrous sodium sulfate, the solvent was distilled off under reduced pressure. The obtained residue was subjected to silica gel column chromatography (eluent; hexane : ethyl acetate = 1 : 1) to give 4.5 g of the product showing an E/Z ratio of about 1 : 1.

NMR (CDCl<sub>3</sub>, δ, ppm): 2.17-2.72(m, 2H), 3.37-3.78(m, 2H), 3.77(s, 3H), 4.68-5.43(m, 1H), 5.70(t, J = 7.4Hz, 1.5H; Z form), 6.40(t, J = 6.9Hz, 1.5H; E form), 6.52-8.12(m, 7H)

The aforesaid product, 3.5 g, showing an E/Z ratio of about 1 : 1 was dissolved in 50 ml of pyridine and 1.7 ml of methanesulfonyl chloride was dropwise added under ice cooling. After stirring for further an hour under ice cooling, the solvent was distilled off under reduced pressure. The residue was extracted with 200 ml of ethyl acetate. The extract was washed in succession with 1N hydrochloric acid aqueous solution, saturated sodium bicarbonate aqueous solution and saturated sodium chloride aqueous solution. After drying over anhydrous magnesium sulfate, the solvent was distilled off under reduced pressure to give 4.3 g of the objective compound as colorless oil.

NMR (CDCl<sub>3</sub>, δ, ppm): -SO<sub>2</sub>CH<sub>3</sub> ; 2.93(s, 1.5H; E form), 3.00(s, 1.5H; Z form)

This product was provided for use in the following reaction step without particular purification. A ratio of E/Z was about 1 : 1.

#### Reference Example 21

Methyl 11-[2-(1-homopiperaziny)ethyl]thio-6,11-dihydrodibenz[b,e]oxepin-2-carboxylate (Compound u):

The objective compound was obtained as colorless oil in a manner similar to Reference Example 11, using Compound j obtained in Reference Example 10 and homopiperazine.

NMR (CDCl<sub>3</sub>, δ, ppm): 1.4-2.0(m, 2H), 2.4-3.1(m, 12H), 3.85(s, 3H), 5.02(s, 1H), 4.87 and 6.44(ABg, J = 12.9Hz, 2H), 6.75-8.05(m, 7H)

IR (neat, cm<sup>-1</sup>): 2928, 1716, 1610, 1249, 1118, 1007

#### Example 1

Methyl 11-[2-(4-phenyl-1-piperaziny)ethyl]thio-6,11-dihydrodibenz[b,e]oxepin-2-carboxylate (Compound 1a):

After 6.1 g of Compound j obtained in Reference Example 10 and 6.0 ml of 1-phenylpiperazine were heated under reflux in 300 ml of ethanol for 7.5 hours, the solvent was distilled off under reduced pressure. The residue was extracted with 500 ml of ethyl acetate. The extract was washed in succession with saturated sodium bicarbonate aqueous solution and saturated sodium chloride aqueous solution and dried over anhydrous sodium sulfate. Thereafter, the solvent was distilled off under reduced pressure. The obtained residue was subjected to silica gel column chromatography (eluent; hexane : ethyl acetate : triethylamine = 30 : 10 : 3). The resulting crude product was solidified with diethyl ether to give 3.1 g of the objective compound.

#### Example 2

Methyl 11-[2-(4-phenyl-1-piperaziny)ethyl]thio-6,11-dihydrodibenz[b,e]oxepin-2-carboxylate dihydrochloride (Compound 1a):

Compound 1a, 1.2 g, obtained in Example 1 was dissolved in 200 ml of isopropanol and 2 ml of 6 N hydrochloric acid/isopropanol solution was added to the solution followed by stirring at room temperature for

2 hours. The solvent was distilled off under reduced pressure. The obtained crude product was recrystallized from isopropanol to give 0.9 g of the objective compound.

In Examples 3 through 7 below, the objective compound was prepared using Compound j obtained in Reference Example 10 and the corresponding piperazine compound in a manner similar to Example 1.

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### Example 3

Methyl 11-[2-[4-(4-fluorophenyl)-1-piperazinyl]ethyl]thio-6,11-dihydrodibenz[b,e]oxepin-2-carboxylate  
(Compound 8a)

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### Example 4

Methyl 11-[2-[4-(2-methoxyphenyl)-1-piperazinyl]ethyl]thio-6,11-dihydrodibenz[b,e]oxepin-2-carboxylate  
(Compound 9a)

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### Example 5

Methyl 11-[2-[4-(4-benzyl-1-piperazinyl)ethyl]thio-6,11-dihydrodibenz[b,e]oxepin-2-carboxylate (Compound 11a)

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### Example 6

Methyl 11-[2-(4-piperonyl-1-piperazinyl)ethyl]thio-6,11-dihydrodibenz[b,e]oxepin-2-carboxylate (Compound 15a)

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### Example 7

Methyl 11-[2-[4-(diphenylmethyl)-1-piperazinyl]ethyl]thio-6,11-dihydrodibenz[b,e]oxepin-2-carboxylate  
(Compound 16a)

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In Examples 8 through 14 below, the objective compound was prepared using Compound j obtained in Reference Example 10 and the corresponding piperidine compound in a manner similar to Example 1.

### Example 8

Methyl 11-[2-(4-phenyl-1-piperidinyl)ethyl]thio-6,11-dihydrodibenz[b,e]oxepin-2-carboxylate (Compound 37a)

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### Example 9

Methyl 11-[2-(4-benzyl-1-piperidinyl)ethyl]thio-6,11-dihydrodibenz[b,e]oxepin-2-carboxylate (Compound 38a)

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### Example 10

Methyl 11-[2-[4-[2(3H)-benzimidazol-1-yl]piperidino]ethyl]thio-6,11-dihydrodibenz[b,e]oxepin-2-carboxylate  
(Compound 47a)

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### Example 11

Methyl 11-[2-(4-phenyl-4-hydroxy-1-piperidinyl)ethyl]thio-6,11-dihydrodibenz[b,e]oxepin-2-carboxylate  
(Compound 50a)

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### Example 12

Methyl 11-[2-(4-benzyl-4-hydroxy-1-piperidinyl)ethyl]thio-6,11-dihydrodibenz[b,e]oxepin-2-carboxylate  
(Compound 51a)

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### Example 13

Methyl 11-[2-[4-(4-chlorobenzoyl)-1-piperidinyl]ethyl]thio-6,11-dihydrodibenz[b,e]oxepin-2-carboxylate (Compound 53a)

5 Example 14

Methyl 11-[2-(1-phenyl-1,3,8-triazaspiro[4.5]decan-4-on-8-yl)ethyl]thio-6,11-dihydrodibenz[b,e]oxepin-2-carboxylate (Compound 58a)

10 In Examples 15 through 17 below, the objective compound was prepared using the corresponding starting material n, p or r and 4-benzylpiperidine in a manner similar to Example 1.

Example 15

15 Methyl 2-methyl-2-[11-2-(4-benzyl-1-piperidinyl)ethyl]thio-6,11-dihydrodibenz[b,e]oxepin-2-yl]propanoate (Compound 41a)

Example 16

20 Methyl 5-[2-(4-benzyl-1-piperidinyl)ethyl]thio-10,11-dihydro-5H-dibenzo[a,d]cycloheptan-3-carboxylate (Compound 101a)

Example 17

25 Methyl 5-[2-(4-benzyl-1-piperidinyl)ethyl]thio-5H-dibenzo[a,d]cycloheptan-3-carboxylate (Compound 103a)

Example 18

30 Methyl (E,Z)-11-[3-(4-benzyl-1-piperazinyl)propylidene]-6,11-dihydrodibenz[b,e]oxepin-2-carboxylate (Compound E/Z-85a)

The objective compound was prepared using Compound t obtained in Reference Example 20 and 4-benzylpiperidine in a manner similar to Example 1. A ratio of E/Z was 1 : 1.

35 In Examples 19 through 22 below, the objective compound was prepared using the corresponding starting material c or e and the corresponding piperazine or piperidine compound in a manner similar to Example 1.

40 Example 19

Methyl (E)-11-[2-[4-(4-fluorophenyl-1-piperazinyl)ethylidene]-6,11-dihydrodibenz[b,e]oxepin-2-carboxylate (Compound E-73a)

45 Example 20

Methyl (E,Z)-11-[2-[4-(2-pyrimidyl)-1-piperazinyl]ethylidene]-6,11-dihydrodibenz[b,e]oxepin-2-carboxylate (Compound E/Z-78a)

50 Example 21

Methyl (E)-11-[2-(4-benzyl-1-piperidinyl)ethylidene]-6,11-dihydrodibenz[b,e]oxepin-2-acetate (Compound E-98a)

55 Example 22

Methyl (E)-11-[2-[4-[2(3H)-benzimidazol-1-yl]piperidinyl]ethylidene]-6,11-dihydrodibenz[b,e]oxepin-2-acetate



tate (Compound E-88a)

Physicochemical properties of compounds obtained in Examples 1 through 22 are shown in Table 7-1. In Tables 7-1 to 7-5, symbol " in solvent for recrystallization means solidification by tritilation.

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Table 7-1

Example No. (Compound No.)	Appearance	MP (°C) [Solvent for recrystallization]	NMR (solvent) $\delta$ , ppm	IR (method) $\text{cm}^{-1}$	MS m/z	Elemental analysis (%) Upper column: found Lower column: calcd.
1 (1 a)	white crystal	129 - 130 [isopropanol]	(CDCl <sub>3</sub> ) 2.30-3.34(m, 12H), 3.82 (s, 1H), 5.03(s, 1H), 4.83 & 6.39 (ABq, J=13.1Hz, 2H), 6.30-8.06 (m, 12H)	(KBr tablet) 3450, 2948, 2826, 1711, 1610, 1598, 1242	474 (M <sup>+</sup> )	C <sub>28</sub> H <sub>30</sub> N <sub>2</sub> O <sub>3</sub> C H N 71.11 6.23 5.80 70.86 6.37 5.90
2 (1a')	white crystal	188 (dec.) [isopropanol]	-	(KBr tablet) 1714, 1599, 1494, 1118, 1006	-	C <sub>28</sub> H <sub>32</sub> Cl <sub>2</sub> N <sub>2</sub> O <sub>3</sub> C H N 61.75 6.21 5.15 61.42 5.89 5.12
3 (8 a)	yellow amorphous	-	(CDCl <sub>3</sub> ) 2.4-3.3(m, 12H), 3.81 (s, 3H), 3.83(s, 3H), 4.86 & 6.41 (ABq, J=11.7Hz, 2H), 5.03(s, 1H), 6.7-7.4(m, 8H), 7.67(dd, J=2.5, 8.5Hz, 1H), 7.88(d, J=2.5Hz, 1H)	(CHCl <sub>3</sub> ) 2824, 1713, 1611, 1508, 1294, 1132, 1120, 1005	492 (M <sup>+</sup> )	-
4 (9 a)	colorless oil	-	(CDCl <sub>3</sub> ) 2.4-3.3(m, 12H), 3.81 (s, 3H), 3.83(s, 3H), 4.85 & 6.44 (ABq, J=12.8Hz, 2H), 5.07(s, 1H), 6.7-7.4(m, 9H), 7.76(dd, J=2.3, 8.5Hz, 1H), 7.98(d, J=2.3Hz, 1H)	(CHCl <sub>3</sub> ) 2936, 2806, 1708, 1610, 1496, 1005	504 (M <sup>+</sup> )	-

Example No. (Compound No.)	Appearance	MP (°C) [Solvent for re-crystallization]	NMR (solvent) $\delta$ , ppm	IR (method) $\text{cm}^{-1}$	MS m/z	Elemental analysis (%) Upper column: found Lower column: calcd.
5 (11a)	colorless oil	-	(CDCl <sub>3</sub> ) 2.05-2.65(m, 12H), 3.46(s, 2H), 3.83(s, 3H), 4.86 & 6.45(ABq, J=12.0Hz, 2H), 5.06(s, 1H), 6.98-7.44(m, 9H), 7.78(dd, J=2.2, 8.6Hz, 1H), 7.98(d, J=2.2Hz, 1H)	(CHCl <sub>3</sub> ) 2816, 1714, 1611, 1120, 1006	487(M <sup>+</sup> )	-
6 (15a)	colorless oil	-	(CDCl <sub>3</sub> ) 2.25-2.75(m, 12H), 3.36(s, 2H), 3.81(s, 3H), 4.82 & 6.40(ABq, J=12.2Hz, 2H), 5.03(s, 1H), 5.85(s, 2H), 6.55-6.8(m, 3H), 6.82(d, J=4.6Hz, 1H), 7.1-7.35(m, 4H), 7.72(dd, J=2.5, 8.5Hz, 1H), 7.93(d, J=2.5Hz, 1H)	(CHCl <sub>3</sub> ) 1702, 1602, 1296, 1121, 1004	532(M <sup>+</sup> )	-
7 (16a)	colorless oil	-	(CDCl <sub>3</sub> ) 3.84(s, 3H), 4.20(s, 1H), 4.87 & 6.43(ABq, J=12.7Hz, 2H), 5.07(s, 1H), 6.84(d, J=8.6Hz, 1H), 7.78(dd, J=2.2, 8.6Hz, 1H), 7.97(d, J=2.2Hz, 1H)	(CHCl <sub>3</sub> ) 1715, 1610, 1495, 1241, 1117, 1005	564(M <sup>+</sup> )	-
8 (37a)	colorless oil	-	(CDCl <sub>3</sub> ) 1.5-3.2(m, 13H), 3.84(s, 3H), 4.95 & 6.50(ABq, J=12.7Hz, 2H), 5.14(s, 1H), 6.87(d, J=8.5Hz, 1H), 7.05-7.45(m, 9H), 7.81(dd, J=2.2, 8.5Hz, 1H), 8.03(d, J=2.2Hz, 1H)	(CHCl <sub>3</sub> ) 1719, 1609, 1496, 1247, 1128, 1009	473(M <sup>+</sup> )	-

Example No. (Compound No.)	Appearance	MP (°C) [Solvent for re-crystallization]	NMR (solvent) $\delta$ , ppm	IR (method) $\text{cm}^{-1}$	MS m/z	Elemental analysis (%) Upper column: found Lower column: calcd.
9 (38a)	colorless oil	-	(CDCl <sub>3</sub> ) 1.0-2.2 (m, 7H), 2.2-3.2 (m, 8H), 3.83 (s, 3H), 4.85 & 6.41 (ABq, J=13.9Hz, 2H), 5.06 (s, 1H), 6.8-7.4 (m, 10H), 7.75 (dd, J=2.0, 8.5Hz, 1H), 7.95 (d, J=2.0Hz, 1H)	(CHCl <sub>3</sub> ) 2924, 1713, 1611, 1295, 1120, 1007	487 (M <sup>+</sup> )	-
10 (47a)	colorless oil	-	(DMSO-d <sub>6</sub> ) 1.5-3.0 (m, 12H), 3.82 (s, 3H), 4.12 (m, 1H), 5.08 & 6.31 (ABq, J=12.7Hz, 2H), 5.51 (s, 1H), 6.91 (d, J=8.6Hz, 1H), 6.95-7.5 (m, 8H), 7.75 (dd, J=2.2, 8.6Hz, 1H), 8.04 (d, J=2.2Hz, 1H)	(CHCl <sub>3</sub> ) 1718, 1686, 1251, 1119, 1015	529 (M <sup>+</sup> )	-
11 (50a)	colorless oil	-	(CDCl <sub>3</sub> ) 3.87 (s, 3H), 4.90 & 6.45 (ABq, J=12.9Hz, 2H), 5.09 (s, 1H), 6.86 (d, J=8.6Hz, 1H), 7.79 (dd, J=2.2, 8.6Hz, 1H), 8.00 (d, J=2.2Hz, 1H)	(CHCl <sub>3</sub> ) 1715, 1610, 1247, 1117, 1005	489 (M <sup>+</sup> )	-
12 (51a)	colorless oil	-	(CDCl <sub>3</sub> ) 1.2-2.7 (m, 12H), 2.73 (s, 2H), 3.86 (s, 3H), 4.88 & 6.44 (ABq, J=12.7Hz, 2H), 5.06 (s, 1H), 6.84 (d, J=8.6Hz, 1H), 7.00-7.40 (m, 9H), 7.78 (dd, J=2.2, 8.6Hz, 1H), 7.98 (d, J=2.2Hz, 1H)	(neat) 2916, 1696, 1609, 1570, 1491, 1113, 1008	503 (M <sup>+</sup> )	-

Example No. (Compound No.)	Appearance	MP (°C) [Solvent for recrystallization]	NMR (solvent) $\delta$ , ppm	IR (method) $\text{cm}^{-1}$	MS m/z	Elemental analysis (%) Upper column: found Lower column: calcd.
13 (53a)	colorless oil	-	(CDCl <sub>3</sub> ) 1.80-3.50 (m, 13H), 3.87 (s, 3H), 4.90 $\pm$ 6.43 (ABq, J=12.7 Hz, 2H), 5.09 (s, 1H), 6.86 (d, J=8.6 Hz, 1H), 7.20-7.35 (m, 4H), 7.43 (d, J=8.6 Hz, 2H), 7.85 (d, J=8.6 Hz, 2H), 7.70-7.95 (m, 2H)	(CHCl <sub>3</sub> ) 2954, 2930, 1715, 1293, 1132, 1121	536 (M <sup>+</sup> )	-
14 (58a)	colorless oil	-	(CDCl <sub>3</sub> ) 1.6-1.8 (m, 2H), 2.4-3.0 (m, 10H), 3.86 (s, 3H), 4.72 (s, 2H), 4.90 $\pm$ 6.46 (ABq, J=12.8 Hz, 2H), 5.13 (s, 1H), 6.6-7.4 (m, 10H), 7.75 (dd, J=2.0, 8.5 Hz, 1H), 8.00 (d, J=2.0 Hz, 1H)	(CHCl <sub>3</sub> ) 2926, 1703, 1600, 1499, 1376, 1243, 1118, 1006	543 (M <sup>+</sup> )	-
15 (41a)	colorless oil	-	(CDCl <sub>3</sub> ) 1.1-2.9 (m, 15H), 1.55 (s, 6H), 3.61 (s, 3H), 4.99 (s, 1H), 4.80 $\pm$ 6.29 (ABq, J=13.0 Hz, 2H), 6.7-7.35 (m, 12H)	(CHCl <sub>3</sub> ) 2922, 1726, 1496, 1265, 1147, 1124, 1105, 1013, 906	529 (M <sup>+</sup> )	-
16 (101a)	pale yellow oil	-	(CDCl <sub>3</sub> ) 1.13-1.99 (m, 9H), 2.35-2.99 (m, 8H), 3.62-3.96 (m, 2H), 3.83 (s, 3H), 5.16 (s, 1H), 6.95-7.30 (m, 10H), 7.80 (dd, J=1.5, 7.9 Hz, 1H), 7.92 (s, 1H)	(CHCl <sub>3</sub> ) 1717, 1301, 1280, 1105	485 (M <sup>+</sup> )	-

Example No. (Compound No.)	Appearance	MP (°C) [solvent for re-crystallization]	NMR (solvent) $\delta$ , ppm	IR (method) $\text{cm}^{-1}$	MS m/z	Elemental analysis (%) Upper column: found Lower column: calcd.
17 (103a)	pale yellow oil	-	(CDCl <sub>3</sub> ) 1.11-2.03 (m, 9H), 2.35 (s, 2H), 2.38-2.81 (m, 4H), 3.90 (s, 3H), 5.29 (s, 1H), 7.01 (s, 1H), 7.02 (s, 1H), 7.06-7.42 (m, 10H), 7.92 (dd, J=1.1, 9.7 Hz, 1H), 7.99 (s, 1H)	(CHCl <sub>3</sub> ) 1716, 1299, 1279, 1110	483 (M <sup>+</sup> )	-
18 (E/Z-85a)	colorless oil	-	(CDCl <sub>3</sub> ) 0.8-2.0 (m, 7H), 2.1-2.9 (m, 8H), 3.78 (s, 3H), 5.18 (ds, 2H), 5.65 (t, J=7.0 Hz, 2 form), 5.99 (t, J=7.0 Hz; E form), 6.6-8.0 (m, 12H), 7.93 (d, J=2.4 Hz, 1H)	(neat) 2916, 1718, 1604, 1118, 1004	467 (M <sup>+</sup> )	-
19 (E-73a)	colorless amorphous	-	(CDCl <sub>3</sub> ) 2.3-3.4 (m, 10H), 3.83 (s, 3H), 4.7-5.6 (m, 2H), 6.25 (t, J=5.5 Hz, 1H), 6.6-7.5 (m, 9H), 7.73 (dd, J=2.2, 8.6 Hz, 1H), 8.00 (d, J=2.2 Hz, 1H)	(KBr tablet) 1707, 1605, 1504, 1240, 1002	458 (M <sup>+</sup> )	-
20 (E/Z-76a)	colorless oil	-	(CDCl <sub>3</sub> ) 2.35-2.7 (m, 4H), 3.0-3.35 (m, 2H), 3.88 (s, 3H), 3.65-4.0 (m, 4H), 4.6-5.6 (m, 2H), 6.30 (t, J=6.8 Hz; E form), 6.45 (t, J=4.8 Hz; E form), 6.79 (d, J=8.6 Hz, 1H), 7.0-7.5 (m, 5H), 7.78 (dd, J=2.2, 8.6 Hz, 1H), 8.05 (d, J=2.2 Hz, 1H), 8.27 (d, J=4.6 Hz, 2H)	(KBr tablet) 1713, 1585, 1357, 1244, 1002	442 (M <sup>+</sup> )	-

Example No. (Compound No.)	Appearance	MP (°C) [Solvent for re-crystallization]	NMR (solvent) $\delta$ , ppm	IR (method) $\text{cm}^{-1}$	MS m/z	Elemental analysis (%) Upper column: found Lower column: calcd.
21 (E-96a)	colorless oil	-	(CDCl <sub>3</sub> ) 1.0-3.2(m, 1H), 3.47(s, 2H), 3.61(s, 3H), 4.85-5.3(m, 2H), 6.14(t, J=6.8Hz, 1H), 6.66(d, J=9Hz, 1H), 6.8-7.34(m, 1H)	(neat) 2918, 1735, 1489, 1011	467(M <sup>+</sup> )	-
22 (E-98a)	colorless oil	-	(CDCl <sub>3</sub> ) 1.35-3.35(m, 1H), 3.54(s, 2H), 3.68(s, 3H), 4.65-5.5(m, 2H), 6.22(t, J=6.6Hz, 1H), 6.73(d, J=8.4Hz, 1H), 6.8-7.5(m, 1OH)	(CHCl <sub>3</sub> ) 1689, 1486, 1148, 1008	509(M <sup>+</sup> )	-

Example 23

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11-[2-(4-Phenyl-1-piperazinyl)ethyl]thio-6,11-dihydrodibenz[b,e]oxepin-2-carboxylic acid \* monohydrate  
(Compound 1b):

10 Compound 1a, 0.8 g, obtained in Example 1 was dissolved in a solvent mixture of 300 ml of methanol and 10 ml of water and, 80 mg of sodium hydroxide was added to the solution followed by heating under reflux for 4 hours. After allowing to cool, 4 N hydrochloric acid aqueous solution was added to the mixture to adjust pH to 7. The mixture was concentrated under reduced pressure. Precipitated crystals were taken by filtration. After washing with water, 0.7 g of the objective compound was obtained as white solid.

15 In Examples 24 through 30 below, the objective compound was prepared by hydrolyzing esters of the corresponding oxepine derivatives in a manner similar to Example 23.

Example 24

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11-[2-[4-(4-Fluorophenyl-1-piperazinyl)ethyl]thio-6,11-dihydrodibenz[b,e]oxepin-2-carboxylic acid  
(Compound 8b)

Example 25

25

11-[2-[4-(2-Methoxyphenyl-1-piperazinyl)ethyl]thio-6,11-dihydrodibenz[b,e]oxepin-2-carboxylic acid  
(Compound 9b)

Example 26

30

11-[2-(4-Benzyl-1-piperazinyl)ethyl]thio-6,11-dihydrodibenz[b,e]oxepin-2-carboxylic acid \* monohydrate  
(Compound 11b)

Example 27

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11-[2-(4-Piperonyl-1-piperazinyl)ethyl]thio-6,11-dihydrodibenz[b,e]oxepin-2-carboxylic acid (Compound 15b)

Example 28

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11-[2-[4-(Diphenylmethyl)-1-piperazinyl]ethyl]thio-6,11-dihydrodibenz[b,e]oxepin-2-carboxylic acid  
(Compound 16b)

Example 29

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11-[2-(4-Phenyl-1-piperidinyl)ethyl]thio-6,11-dihydrodibenz[b,e]oxepin-2-carboxylic acid \* 0.5 hydrate  
(Compound 37b)

Example 30

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11-[2-(4-Benzyl-1-piperidinyl)ethyl]thio-6,11-dihydrodibenz[b,e]oxepin-2-carboxylic acid (Compound 38b)

Example 31

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Sodium 11-[2-(4-Benzyl-1-piperidinyl)ethyl]thio-6,11-dihydrodibenz[b,e]oxepin-2-carboxylate (Compound 38b)

Compound 38b, 45 g, obtained in Example 30 was suspended in 500 ml of methanol and 5.3 g of sodium methoxide was added to the suspension followed by stirring at room temperature for 4 hours. The



solvent was distilled off under reduced pressure. The resulting crude product was solidified with hexane to give 47 g of the objective compound.

8 Example 32

11-[2-{4-[2(3H)-Benzimidazol-1-yl]piperidino}ethyl]thio-6,11-dihydrodibenz[b,e]oxepin-2-carboxylic acid\*0.5 Isopropanol\*0.5 hydrate (Compound 47b):

10 The objective compound was prepared by hydrolyzing Compound 47a obtained in Example 10 in a manner similar to Example 23.

Example 33

15 Sodium 11-[2-{4-[2(3H)-Benzimidazol-1-yl]piperidino}ethyl]thio-6,11-dihydrodibenz[b,e]oxepin-2-carboxylate\*2.5 hydrate (Compound 47b')

Compound 47b' obtained in Example 32 was converted into the sodium salt in a manner similar to Example 31.

In Examples 34 through 38 below, the objective compound was prepared by hydrolyzing esters of the corresponding oxepine derivatives in a manner similar to Example 23.

25 Example 34

11-[2-(4-Phenyl-4-hydroxy-1-piperidinyl)ethyl]thio-6,11-dihydrodibenz[b,e]oxepin-2-carboxylic acid\*0.75 hydrate (Compound 50b')

30 Example 35

11-[2-(4-Benzyl-4-hydroxy-1-piperidinyl)ethyl]thio-6,11-dihydrodibenz[b,e]oxepin-2-carboxylic acid\*0.2 isopropanol (Compound 51b)

35 Example 36

11-[2-(4-(4-Chlorobenzoyl-1-piperidinyl)ethyl]thio-6,11-dihydrodibenz[b,e]oxepin-2-carboxylic acid\*0.1 isopropanol\*0.5 hydrate (Compound 53b)

40 Example 37

11-[2-(1-Phenyl-1,3,8-triazaspiro[4.5]decan-4-on-8-yl)ethyl]thio-6,11-dihydrodibenz[b,e]oxepin-2-carboxylic acid\*0.25 hydrate (Compound 58b)

45 Example 38

2-Methyl-2-[11-[2-(4-benzyl-1-piperidinyl)ethyl]thio-6,11-dihydrodibenz[b,e]oxepin-2-yl]propionate (Compound 41)

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Example 39

5-[2-(4-Benzyl-1-piperidinyl)ethyl]thio-10,11-dihydro-5H-dibenzo[a,d]cyclohepten-3-carboxylic acid\*0.5 fumarate (Compound 101b):

55

Compound 101a, 0.33 g, obtained in Example 16 was hydrolyzed in a manner similar to Example 23 to give 0.3 g of the corresponding Compound 101b. This compound was dissolved in 50 ml of acetone and 80 mg of fumaric acid was added to the solution followed by stirring at room temperature for 2 hours. The

solvent was distilled off under reduced pressure. The resulting crude product was recrystallized from isopropanol to prepare 0.14 g of the objective compound.

5 Example 40

5-[2-(4-Benzyl-1-piperidinyl)ethyl]thio-5H-dibenzo[a,d]cycloheptan-3-carboxylic acid\*0.5 fumarate\*0.5 isopropanol (Compound 103b):

10 The objective compound was prepared using Compound 103a obtained in Example 17 in a manner similar to Example 39.

Example 41

15 (E,Z)-11-[3-(4-Benzyl-1-piperazinyl)propylidene]-6,11-dihydrodibenz[b,e]oxepin-2-carboxylic acid\*1.5 hydrate (Compound 85b):

The objective compound was prepared by hydrolyzing Compound E/Z-85a obtained in Example 18 in a manner similar to Example 23.

A ratio of E/Z was 38 : 62.

In Examples 42 and 43 below, the objective compound was prepared by hydrolyzing esters of the corresponding oxepine derivatives in a manner similar to Example 23.

25 Example 42

(E)-11-[2-[4-(4-Fluorophenyl)-1-piperazinyl]ethylidene]-6,11-dihydrodibenz[b,e]oxepin-2-carboxylic acid\*isopropanol\* monohydrate (Compound E-73b)

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Example 43

(E,Z)-11-[2-[4-(2-Pyrimidyl)-1-piperazinyl]ethylidene]-6,11-dihydrodibenz[b,e]oxepin-2-carboxylic acid (Compound E/Z- 76b)

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A ratio of E/Z was 85 : 15.

40 Example 44

(E)-11-[2-(4-Benzyl-1-piperidinyl)ethylidene]-6,11-dihydrodibenz[b,e]oxepin-2-acetic acid\*0.5 fumarate (Compound E-88b):

45 Compound E-88a, 2.0 g, obtained in Example 21 was hydrolyzed in a manner similar to Example 23 to give 1.7 g of the corresponding Compound E-88b. This compound was dissolved in 200 ml of acetone and 0.43 g of fumaric acid was added to the solution followed by stirring at room temperature for an hour. The precipitated crystals were taken by filtration to give 1.5 g of the objective compound.

50 Example 45

(E)-11-[2-[4-[2(3H)-Benzimidazol-1-yl]piperidinyl]ethylidene]-6,11-dihydrodibenz[b,e]oxepin-2-acetic acid\* monohydrate (Compound E-98b):

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The objective compound was prepared by hydrolyzing Compound E-98a obtained in Example 22 in a manner similar to Example 23.

**Example 46**

Sodium (E)-11-[2-[4-[2(3H)-benzimidazol-1-yl]piperidino]ethylidene]-6,11-dihydrodibenz[b,e]oxepin-2-acetate (Compound E-98b<sup>1</sup>):

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Compound E-98b<sup>1</sup> obtained in Example 45 was converted into the sodium salt in a manner similar to Example 31.

Physicochemical properties of the compounds obtained in Examples 23 through 46 are shown in Table 7-2.

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Table 7-2

Example No. (Compound No.)	Appearance	MP (°C) [Solvent for re-crystallization]	NMR (solvent) $\delta$ , ppm	IR (method) $\text{cm}^{-1}$	MS m/z	Elemental analysis (%) Upper column: found Lower column: calcd.
23 (1b')	white crystal	215 (dec.) [isopropanol]	(DMSO- $d_6$ ) 2.50-3.83(m, 12H), 5.08 & 6.23(ABq, J=12.5Hz, 2H), 5.51(s, 1H), 6.79-8.04(m, 12H)	(KBr tablet) 3400, 2936, 2824, 1694, 1599, 1232	460 (M <sup>+</sup> )	$\text{C}_{27}\text{H}_{28}\text{N}_2\text{O}_3\cdot\text{H}_2\text{O}$ C H N 67.39 6.06 5.59 67.76 6.32 5.85
24 (8 b)	white amorphous	165 - 166 (dec.) [isopropanol]*	(DMSO- $d_6$ ) 2.3-2.8(m, 8H), 3.0 -3.1(m, 4H), 5.05 & 6.28(ABq, J=12.6Hz, 2H), 5.45(s, 1H), 6.85-7.1(m, 5H), 7.4-7.5(m, 4H), 7.71 (dd, J=2.2, 8.7Hz, 1H), 7.99(d, J=2.2Hz, 1H)	(KBr tablet) 3400, 1607, 1509, 1384, 1232	478 (M <sup>+</sup> )	$\text{C}_{27}\text{H}_{27}\text{FN}_2\text{O}_3\text{S}$ C H N 67.55 5.65 5.58 67.76 5.69 5.85
25 (9 b)	white amorphous	171 - 175 [isopropanol]*	(DMSO- $d_6$ ) 3.0-3.7(m, 12H), 3.81 (s, 3H), 5.10 & 6.16(ABq, J=12.9 Hz, 2H), 5.54(s, 1H), 6.90(d, J=8.5Hz, 1H), 6.9-7.1(m, 4H), 7.4-7.65(m, 4H), 7.74(dd, J=2.2, 8.5Hz, 1H), 8.06(d, J=2.2Hz, 1H)	(KBr tablet) 1694, 1609, 1499, 1455, 1200, 1115, 1013	490 (M <sup>+</sup> )	$\text{C}_{28}\text{H}_{30}\text{N}_2\text{O}_4\text{S}$ C H N 68.23 6.35 5.88 68.55 6.16 5.71
26 (11b')	white crystal	112 - 115 (dec.) [isopropyl ether]	(DMSO- $d_6$ ) 2.2-2.7(m, 12H), 3.42 (s, 2H), 5.01 & 6.24(ABq, J=12.7 Hz, 2H), 5.39(s, 1H), 6.81(d, J=8.6Hz, 1H), 7.2-7.5(m, 9H), 7.69 (dd, J=2.2, 8.6Hz, 1H), 7.95(d, 2.2Hz, 1H)	(KBr tablet) 3400, 1609, 1384, 1232, 1008	474 (M <sup>+</sup> )	$\text{C}_{28}\text{H}_{30}\text{N}_2\text{O}_5\cdot\text{H}_2\text{O}$ C H N 68.33 6.25 5.54 68.27 6.55 5.69

Example No. (Compound No.)	Appearance	MP (°C) [Solvent for re-crystallization]	NMR (solvent) $\delta$ , ppm	IR (method) $\text{cm}^{-1}$	MS m/z	Elemental analysis (%) Upper column: found Lower column: calcd.
27 (15b)	white crystal	218 - 220 (dec.) [isopropanol]	(DMSO- $d_6$ ) 2.2-2.7(m, 12H), 3.34 (s, 2H), 5.04 & 6.26(ABq, J=12.7 Hz, 2H), 5.43 (s, 1H), 5.98 (s, 2H), 6.72 (d, J=7.9 Hz, 1H), 6.8-6.85 (m, 2H), 6.85 (d, J=8.6 Hz, 1H), 7.35-7.5 (m, 4H), 7.71 (dd, J=2.2, 8.6 Hz, 1H), 7.98 (d, J=2.2 Hz, 1H)	(KBr tablet) 1670, 1609, 1491, 1242, 1040	518 (M <sup>+</sup> )	$\text{C}_{29}\text{H}_{30}\text{N}_2\text{O}_5$ C H N 67.09 5.95 5.24 67.16 5.83 5.40
28 (16b)	white crystal	145 - 148 [isopropanol]	(DMSO- $d_6$ ) 2.2-2.7(m, 12H), 4.26 (s, 1H), 5.04 & 6.23(ABq, J=12.6 Hz, 2H), 5.44 (s, 1H), 6.86 (d, J=8.6 Hz, 1H), 7.70 (dd, J=2.2, 8.6 Hz, 1H), 7.98 (d, J=2.2 Hz, 1H)	(KBr tablet) 1683, 1608, 1493, 1451, 1232, 1006	550 (M <sup>+</sup> )	$\text{C}_{34}\text{H}_{34}\text{N}_2\text{O}_5$ C H N 73.98 6.35 5.00 74.15 6.22 5.09
29 (37b')	white crystal	182 - 184 [isopropanol]	(DMSO- $d_6$ ) 1.65-3.0 (m, 13H), 5.06 & 6.27(ABq, J=12.7 Hz, 2H), 5.48 (s, 1H), 6.88 (d, J=8.5 Hz, 1H), 7.15-7.5 (m, 9H), 7.72 (dd, J=2.2, 8.5 Hz, 1H), 8.01 (d, J=2.2 Hz, 1H)	(KBr tablet) 1680, 1605, 1494, 1318, 1233, 1104, 1007	459 (M <sup>+</sup> )	$\text{C}_{28}\text{H}_{29}\text{NO}_3 \cdot 0.5\text{H}_2\text{O}$ C H N 71.85 6.23 2.98 71.77 6.45 2.99
30 (38b)	white crystal	205 - 206 [isopropanol]	(DMSO- $d_6$ ) 1.0-1.9 (m, 7H), 2.2-2.85 (m, 8H), 5.03 & 6.25(ABq, J=12.6 Hz, 2H), 5.42 (s, 1H), 6.86 (d, J=8.6 Hz, 1H), 7.1-7.5 (m, 9H), 7.70 (dd, J=2.2, 8.6 Hz, 1H), 7.97 (d, J=2.2 Hz, 1H)	(KBr tablet) 2912, 1687, 1608, 1494, 1449, 1323, 1229, 1019	473 (M <sup>+</sup> )	$\text{C}_{29}\text{H}_{31}\text{NO}_3$ C H N 73.48 6.72 2.92 73.54 6.60 2.96

Example No. (Compound No.)	Appearance	MP (°C) [Solvent for re-crystallization]	NMR (solvent) $\delta$ , ppm	IR (method) $\text{cm}^{-1}$	MS m/z	Elemental analysis (%) Upper column: found Lower column: calcd.
31 (38b')	white solid	119 - 121 [hexane]*	-	(KBr tablet) 2918, 1612, 1582, 1548, 1385, 1252, 1227, 1105, 1011	-	$\text{C}_{29}\text{H}_{30}\text{NO}_3\text{SNa} \cdot 0.5\text{H}_2\text{O}$ C H N 68.80 6.44 2.93 69.03 6.19 2.78
32 (47b')	white crystal	185 - 188 [aceto-nitrile / isopropanol]	(DMSO- $d_6$ ) 1:55-3.1 (m, 12H), 4.14 (m, 1H), 5.07 & 6.28 (ABq, J=12.7 Hz, 2H), 5.48 (s, 1H), 6.89 (d, J=8.6 Hz, 1H), 6.95-7.5 (m, 8H), 7.72 (dd, J=2.2, 8.6 Hz, 1H), 8.06 (d, J=2.2 Hz, 1H), 10.83 (s, 1H)	(KBr tablet) 1695, 1485, 1386	515 (M+)	$\text{C}_{29}\text{H}_{29}\text{N}_3\text{O}_4\text{S} \cdot 0.5\text{C}_2\text{H}_3\text{O}$ C H N 66.41 5.90 8.45 66.75 6.22 8.57
33 (47b'')	white crystal	224 - 227 [methanol]	-	(KBr tablet) 1683, 1583, 1543, 1485, 1379, 1256, 1229, 1109, 1006	-	$\text{C}_{29}\text{H}_{28}\text{N}_3\text{O}_4\text{SNa} \cdot 2.5\text{H}_2\text{O}$ C H N 59.77 5.72 7.10 59.78 5.71 7.21
34 (50b')	white crystal	150 - 154 [isopropanol]	(DMSO- $d_6$ ) 1.5-1.6 (m, 2H), 1.8-2.0 (m, 2H), 2.35-2.75 (m, 8H), 5.05 & 6.28 (ABq, J=12.7 Hz, 2H), 6.87 (d, J=8.5 Hz, 1H), 7.15-7.5 (m, 9H), 7.71 (dd, J=2.1, 8.5 Hz, 1H), 8.01 (d, J=2.1 Hz, 1H)	(KBr tablet) 1611, 1584, 1550, 1375, 1228, 1006	475 (M+)	$\text{C}_{28}\text{H}_{29}\text{N}_3\text{O}_4\text{S} \cdot 0.75\text{H}_2\text{O}$ C H N 68.60 6.41 2.88 68.76 6.29 2.86

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Example No. (Compound No.)	Appearance	MP (°C) [Solvent for re-crystallization]	NMR (solvent) $\delta$ , ppm	IR (method) $\text{cm}^{-1}$	MS m/z	Elemental analysis (%) Upper column: found Lower column: calcd.
35 (51b')	white crystal	232 [isopropanol]	(DMSO- $d_6$ ) 1.25-1.5(m, 4H), 2.2-2.75(m, 8H), 2.63(s, 2H), 5.02 s 6.25(ABq, J=12.6Hz, 2H), 5.41(s, 1H), 6.85(d, J=8.6Hz, 1H), 7.15-7.45(m, 9H), 7.70(dd, J=2.2, 8.6 Hz, 1H), 7.96(d, J=2.2Hz, 1H)	(KBr tablet) 1608, 1492, 1383, 1232, 1107, 1009	489(M+)	$\text{C}_{29}\text{H}_{31}\text{NO}_4 \cdot 0.2\text{C}_3\text{H}_8\text{O}$ C H N 70.98 6.43 2.49 70.87 6.55 2.79
36 (53b')	pale yellow solid	138-139 [isopropyl ether]*	(DMSO- $d_6$ ) 1.4-3.5(m, 13H), 5.05 s 6.27(ABq, J=12.6Hz, 2H), 5.46 (s, 1H), 6.87(d, J=8.6Hz, 1H), 7.35-7.5(m, 4H), 7.59(d, J=8.6Hz, 2H), 7.71(dd, J=2.1, 8.6Hz, 1H), 7.97(d, J=8.6Hz, 2H), 7.99(d, J=2.1Hz, 1H)	(KBr tablet) 1676, 1608, 1587, 1376, 1229, 1091, 1009	522(M+)	$\text{C}_{29}\text{H}_{28}\text{ClNO}_4 \cdot 0.1$ $\text{C}_6\text{H}_{14}\text{O} \cdot 0.5\text{H}_2\text{O}$ C H N 65.81 5.39 2.22 65.46 5.68 2.58
37 (58b')	white solid	186-189 [isopropanol]	(DMSO- $d_6$ ) 1.56(d, J=13.6Hz, 2H), 2.4-2.9(m, 10H), 4.57(s, 2H), 5.06 s 6.26(ABq, J=12.6Hz, 2H), 5.54(s, 1H), 6.75-7.2(m, 5H), 6.88(d, J=8.5Hz, 1H), 7.72(dd, J=2.1, 8.5Hz, 1H), 8.04(d, J=2.1Hz, 1H), 8.64(s, 1H)	(KBr tablet) 1710, 1598, 1497, 1379, 1105, 1009	529(M+)	$\text{C}_{30}\text{H}_{31}\text{N}_3\text{O}_5 \cdot 0.25\text{H}_2\text{O}$ C H N 67.29 5.82 7.69 67.46 5.94 7.87
38 (41b)	white solid	84-86 [water]*	(DMSO- $d_6$ ) 1.1-1.3(m, 2H), 1.35 -1.6(m, 3H), 1.43 s 1.45(each s, 6H), 1.85-2.0(m, 2H), 2.35-2.9 (m, 8H), 5.33(s, 1H), 4.91 s 6.13 (ABq, J=12.8Hz, 2H), 6.76(d, J=8.4Hz, 1H), 7.1-7.45(m, 11H)	(KBr tablet) 2926, 1720, 1496, 1453, 1231, 1123, 1014	515(M+)	$\text{C}_{32}\text{H}_{37}\text{NO}_3$ C H N 74.21 7.45 2.66 74.53 7.23 2.72

Example No. (Compound No.)	Appearance	MP (°C) [Solvent for re-crystallization]	NMR (solvent) $\delta$ , ppm	IR (method) $\text{cm}^{-1}$	MS m/z	Elemental analysis (%) Upper column: found Lower column: calcd.
39 (101b')	pale yellow crystal	127 - 130 [isopropanol]	(CD <sub>3</sub> OD) 1.20-1.86(m, 5H), 2.35-3.20(m, 12H), 3.56-3.94(m, 2H), 5.21(s, 1H), 7.03-7.33(m, 10H), 7.75-7.86(m, 2H)	(KBr tablet) 3420, 2920, 1695, 1585, 1453, 1372, 1244	469(M <sup>+</sup> )	$\text{C}_{30}\text{H}_{31}\text{NO}_2 \cdot 0.5\text{C}_4\text{H}_4\text{O}_4$ C H N 72.60 6.60 2.55 72.84 6.30 2.65
40 (103b')	pale yellow crystal	100 - 115 (dec.) [isopropanol]	(DMSO-d <sub>6</sub> ) 1.18-2.11(m, 5H), 2.26-2.60(m, 8H), 2.63-2.92(m, 2H), 5.44(s, 1H), 6.99(s, 2H), 7.01-7.49(m, 10H), 7.84(dd, J=1.3, 9.0Hz, 1H), 8.02(s, 1H)	(KBr tablet) 3400, 2922, 1698, 1582, 1372	467(M <sup>+</sup> )	$\text{C}_{30}\text{H}_{29}\text{NO}_2 \cdot 0.5\text{C}_4\text{H}_4\text{O}_4 \cdot 0.5\text{C}_3\text{H}_8\text{O}$ C H N 72.23 6.51 2.41 72.41 6.35 2.52
41 (E/Z-85b')	white solid	146 - 149 [water]*	(DMSO-d <sub>6</sub> ) 5.70(t, J=6.7Hz; Z form), 6.06(t, J=7.5Hz; E form), 6.80(d, J=8.4Hz; E form), 6.88(d, J=8.6Hz; Z form), 7.76(d, J=2.2Hz; Z form), 7.87(d, J=2.0Hz; E form)	(KBr tablet) 2920, 1605, 1486, 1454, 1367, 1242, 1004	453(M <sup>+</sup> )	$\text{C}_{30}\text{H}_{31}\text{NO}_3 \cdot 1.5\text{H}_2\text{O}$ C H N 75.09 7.22 2.82 74.97 7.13 2.91
42 (E-73b')	white crystal	236 - 238 (dec.) [isopropanol]	(DMSO-d <sub>6</sub> ) 2.4-3.7(m, 10H), 4.95-5.65(m, 2H), 6.17(t, J=6.7Hz, 1H), 6.83(d, J=8.5Hz, 1H), 6.85-7.55(m, 8H), 7.72(dd, J=2.2, 8.5Hz, 1H), 7.92(d, J=2.2Hz, 1H)	(KBr tablet) 1608, 1509, 1370, 1224	444(M <sup>+</sup> )	$\text{C}_{27}\text{H}_{25}\text{FN}_2\text{O}_3 \cdot \text{C}_3\text{H}_8\text{O} \cdot \text{H}_2\text{O}$ C H N 69.10 6.60 5.33 68.95 6.75 5.36



Example No. (Compound No.)	Appearance	MP (°C) [Solvent for re-crystallization]	NMR (solvent) $\delta$ , ppm	IR (method) $\text{cm}^{-1}$	MS m/z	Elemental analysis (%) Upper column: found Lower column: calcd.
43 (E/Z-76b)	pale yellow crystal	161 - 165 [isopropanol]	(DMSO- $d_6$ ) 2.3-3.8(m, 10H), 4.9-5.65(m, 2H), 5.86(t, J=7.1Hz; 2 form), 6.18(t, J=6.6Hz; 8 form), 6.6-8.4(m, 10H)	(KBr tablet) 1584, 1551, 1487, 1450, 1360	428 (M <sup>+</sup> )	$\text{C}_{25}\text{H}_{24}\text{N}_2\text{O}_3$ C 69.91 5.78 13.22 H 70.08 5.65 13.08
44 (E-96b')	white solid	203-204.5 [acetone]*	(DMSO- $d_6$ ) 1.1-2.55(m, 9H), 2.46(d, J=6.6Hz, 2H), 2.92(d, J=6.6 Hz, 1H), 3.48(s, 2H), 4.0-5.5(m, 2H), 6.08(t, J=6.6Hz, 1H), 6.57(s, 1H), 6.67(d, J=8.4Hz, 1H), 7.0-7.5(m, 11H)	(KBr tablet) 1704, 1488, 1357, 1249, 1227, 1003	453 (M <sup>+</sup> )	$\text{C}_{30}\text{H}_{31}\text{NO}_3 \cdot 0.5\text{C}_4\text{H}_4\text{O}_4$ C 74.92 6.64 2.79 H 75.12 6.50 2.74
45 (E-98b')	white solid	260 - 262 [isopropanol]*	(DMSO- $d_6$ ) 1.5-3.6(m, 10H), 3.51(s, 2H), 4.05-4.2(m, 1H), 4.93-5.40(each bs, 2H), 6.13(t, J=6.7 Hz, 1H), 6.68(d, J=8.5Hz, 1H), 6.95-7.5(m, 10H)	(KBr tablet) 1699, 1489, 1386, 1003	495 (M <sup>+</sup> )	$\text{C}_{30}\text{H}_{29}\text{N}_3\text{O}_4 \cdot \text{H}_2\text{O}$ C 70.31 6.45 8.35 H 70.16 6.08 8.18
46 (E-98b'')	white crystal	218-219.5 [isopropanol]	(DMSO- $d_6$ ) 1.5-2.5(m, 4H), 2.9-3.4(m, 6H), 3.15(s, 2H), 4.0-4.15(m, 1H), 4.2-5.6(m, 2H), 6.10(t, J=6.7Hz, 1H), 6.57(d, J=8.3Hz, 1H), 6.9-7.5(m, 10H), 8.29(s, 1H)	(KBr tablet) 1681, 1567, 1485, 1381, 1009	-	$\text{C}_{30}\text{H}_{28}\text{N}_3\text{O}_4 \cdot \text{Na}$ C 69.66 5.52 8.22 H 69.62 5.45 8.12

Example 47

Methyl 11-[2-(4-cinnamyl-1-piperazinyl)ethyl]thio-6,11-dihydrodibenz[b,e]oxepin-2-carboxylate (Compound 20a) :

Compound k, 2.0 g, obtained in Reference Example 11 and 1.2 g of cinnamyl bromide were stirred in 100 ml of isopropanol for 4 hours at room temperature. The mixture was extracted with 300 ml of ethyl acetate. The extract was washed in succession with saturated sodium bicarbonate aqueous solution and saturated sodium chloride aqueous solution. After drying over anhydrous sodium sulfate, the solvent was distilled off under reduced pressure. The obtained residue was purified by silica gel column chromatography (eluent; hexane : ethyl acetate : triethylamine = 10 : 10 : 1) to give 0.8 g of the objective compound as pale yellow oil.

Example 48

(E)-11-[3-(4-cinnamyl-1-piperazinyl)ethylidene]-6,11-dihydrodibenz[b,e]oxepin-2-carboxylate (Compound E-75a) :

The objective compound was prepared using Compound s obtained in Reference Example 19 and cinnamyl bromide in a manner similar to Example 47.

In Examples 49 and 50 below, the objective compound was prepared by hydrolyzing esters of the corresponding oxepine derivatives and then converting the hydrolyzate into the fumarate in a manner similar to Example 44.

Example 49

11-[2-(4-Cinnamyl-1-piperazinyl)ethyl]thio-6,11-dihydrodibenz[b,e]oxepin-2-carboxylic acid \* difumarate \* 0.5 hydrate (Compound 20b)

Example 50

(E)-11-[3-(4-Cinnamyl-1-piperazinyl)ethylidene]-6,11-dihydrodibenz[b,e]oxepin-2-carboxylic acid \* difumarate \* 0.5 hydrate (Compound E-75b)

Physicochemical properties of the compounds obtained in Examples 47 to 50 are shown in Table 7-3.

Table 7-3

Example No. (Compound No.)	Appearance	MP (°C) [Solvent for re-crystallization]	NMR (solvent) $\delta$ , ppm	IR (method) $\text{cm}^{-1}$	MS m/z	Elemental analysis (%) Upper column: found Lower column: calcd.
47 (20a)	pale yellow oil	-	(CDCl <sub>3</sub> ) 2.32-2.80 (m, 12H), 3.14 (d, J=5.6Hz, 2H), 3.89 (s, 3H), 4.89 & 6.45 (ABq, J=14.5Hz, 2H), 5.10 (s, 1H), 6.21-8.10 (m, 14H)	(neat) 2944, 2808, 1708, 1610, 1571, 1497, 1007	514 (M <sup>+</sup> )	-
48 (E-75a)	colorless oil	-	(CDCl <sub>3</sub> ) 2.48 (m, 8H), 3.0-3.3 (m, 4H), 3.81 (s, 3H), 4.9-5.7 (m, 2H), 6.05-6.5 (m, 2H), 6.71 (d, J=9.2Hz, 1H), 7.0-7.5 (m, J=9.2Hz, 9H), 7.73 (dd, J=2.2, 9.2Hz, 1H), 7.99 (d, J=2.2Hz, 1H)	(neat) 1715, 1606, 1245, 1006	480 (M <sup>+</sup> )	-
49 (20b')	yellow solid	210-211 [aceto-nitrile]*	(DMSO-d <sub>6</sub> ) 2.35-2.7 (m, 12H), 3.25 (d, J=6.6Hz, 2H), 5.05 & 6.25 (ABq, J=10.6Hz, 2H), 5.44 (s, 1H), 6.25-6.3 (m, 1H), 6.87 (d, J=8.5Hz, 1H), 7.2-7.5 (m, 10H), 7.71 (dd, J=2.2, 8.5Hz, 1H), 7.98 (d, J=2.2Hz, 1H)	(KBr tablet) 1713, 1612, 1300, 1252, 1170	500 (M <sup>+</sup> )	C <sub>38</sub> H <sub>40</sub> N <sub>2</sub> O <sub>11</sub> ·0.5H <sub>2</sub> O C H N 61.48 5.42 3.88 61.53 5.57 3.78
50 (E-75b')	white crystal	246 (dec.) [aceto-nitrile]	(DMSO-d <sub>6</sub> ) 2.4-3.3 (m, 10H), 3.24 (d, J=6.7Hz, 2H), 4.9-5.7 (m, 2H), 6.14 (t, J=6.7Hz, 1H), 6.2-6.35 (m, 1H), 6.82 (d, J=8.5Hz, 1H), 7.2-7.55 (m, 10H), 7.72 (dd, J=2.2, 8.5Hz, 1H), 7.90 (d, J=2.2Hz, 1H)	(KBr tablet) 1703, 1659, 1605, 1250, 1010, 980	466 (M <sup>+</sup> )	C <sub>38</sub> H <sub>38</sub> N <sub>2</sub> O <sub>11</sub> ·0.5H <sub>2</sub> O C H N 64.53 5.43 4.05 64.49 5.55 3.96

Example 51

Methyl 11-[2-[4-(4-fluorophenylsulfonyl)-1-piperazinyl]ethyl]thio-6,11-dihydrodibenz[b,e]oxepin-2-carboxylate (Compound 22a):

Compound k, 0.8 g, obtained in Reference Example 11 was dissolved in 50 ml of pyridine and 0.5 g of 4-fluorobenzenesulfonyl chloride was added to the solution followed by stirring at room temperature for 2 hours. The solvent was distilled off under reduced pressure. The residue was extracted with 200 ml of ethyl acetate. The extract was washed in succession with saturated sodium bicarbonate aqueous solution and saturated sodium chloride aqueous solution. After drying over anhydrous magnesium sulfate, the solvent was distilled off under reduced pressure. The obtained residue was subjected to silica gel column chromatography (eluent; hexane : ethyl acetate : triethylamine = 50 : 20 : 1) to give 0.6 g of the crude product. The crude product obtained was recrystallized from isopropyl ether to give 0.4 g of the objective compound.

In Examples 52 and 53 below, the objective compound was prepared using Compound k and the corresponding sulfonyl chloride compound in a manner similar to Example 51.

#### Example 52

Methyl 11-[2-[4-(styrylsulfonyl)-1-piperazinyl]ethyl]thio-6,11-dihydrodibenz[b,e]oxepin-2-carboxylate (Compound 24a)

#### Example 53

Methyl 11-[2-[4-(2-thienylsulfonyl)-1-piperazinyl]ethyl]thio-6,11-dihydrodibenz[b,e]oxepin-2-carboxylate (Compound 25a)

In Examples 54 and 55 below, the objective compound was prepared using Compound k and the corresponding acid chloride compound in a manner similar to Example 51.

#### Example 54

Methyl 11-[2-[4-(2,3,4-trimethoxybenzoyl)-1-piperazinyl]ethyl]thio-6,11-dihydrodibenz[b,e]oxepin-2-carboxylate (Compound 27a)

#### Example 55

Methyl 11-[2-[4-(cinnamoyl)-1-piperazinyl]ethyl]thio-6,11-dihydrodibenz[b,e]oxepin-2-carboxylate (Compound 28a)

#### Example 56

Methyl 11-[2-[4-(phenoxycarbonyl)-1-piperazinyl]ethyl]thio-6,11-dihydrodibenz[b,e]oxepin-2-carboxylate (Compound 30a):

Compound k, 2.0 g, obtained in Reference Example 11 was dissolved in 100 ml of isopropanol and 0.75 ml of phenyl chloroformate was added to the solution followed by stirring at room temperature for 4 hours. The solvent was distilled off under reduced pressure and the resulting residue was extracted with 200 ml of ethyl acetate. The extract was washed in succession with saturated sodium bicarbonate aqueous solution and saturated sodium chloride aqueous solution. After drying over anhydrous magnesium sulfate, the solvent was distilled off under reduced pressure. The obtained residue was subjected to silica gel column chromatography (eluent; hexane : ethyl acetate : triethylamine = 50 : 30 : 1) to give 1.7 g of the crude product. The crude product obtained was solidified with isopropyl ether to give 1.6 g of the objective compound.

#### Example 57

Methyl 11-[2-[4-(benzyloxycarbonyl)-1-piperazinyl]ethyl]thio-8,11-dihydrodibenz[b,e]oxepin-2-carboxylate (Compound 31a)

The objective compound was prepared using Compound k and benzyl chloroformate in a manner similar to Example 58.

In Examples 58 and 59 below, the objective compound was prepared from Compound s obtained in Reference Example 19 and the corresponding sulfonyl chloride compound in a manner similar to Example 51.

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#### Example 58

Methyl (E)-11-[2-[4-(styrylsulfonyl)-1-piperazinyl]ethylidene]-8,11-dihydrodibenz[b,e]oxepin-2-carboxylic acid (Compound E-78a)

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#### Example 59

Methyl (E)-11-[2-[4-(2-thienylsulfonyl)-1-piperazinyl]ethylidene]-8,11-dihydrodibenz[b,e]oxepin-2-carboxylic acid (Compound E-79a)

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#### Example 60

Methyl 11-[2-[4-(1-naphthalenesulfonyl)-1-homopiperazinyl]ethyl]thio-8,11-dihydrodibenz[b,e]oxepin-2-carboxylate (Compound 34a):

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The objective compound was prepared using 1.4 g of Compound u obtained in Reference Example 21 and 1.0 g of 1-naphthalenesulfonyl chloride in a manner similar to Example 51.

Physicochemical properties of the compounds obtained in Examples 51 to 60 are shown in Table 7-4.

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Table 7-4

Example No. (Compound No.)	Appearance	MP (°C) [Solvent for re-crystallization]	NMR (solvent) $\delta$ , ppm	IR (method) $\text{cm}^{-1}$	MS m/z	Elemental analysis (b) Upper column: found Lower column: calcd.
51 (22a)	yellow amorphous	80 - 82 [iso-propyl ether]*	(CDCl <sub>3</sub> ) 2.25-3.15(m, 12H), 3.80 (s, 3H), 4.85 $\pm$ 6.39(ABq, J=13.6 Hz, 2H), 5.00(s, 1H), 6.77(d, J=8.6Hz, 1H), 6.95-8.0(m, 10H)	(KBr tablet) 1702, 1494, 1357, 1326, 1244, 1172, 1004	556 (M <sup>+</sup> )	C <sub>28</sub> H <sub>29</sub> N <sub>2</sub> O <sub>5</sub> S <sub>2</sub> C H N 60.60 5.39 4.88 60.41 5.25 5.03
52 (24a)	colorless oil	-	(CDCl <sub>3</sub> ) 2.4-2.75(m, 8H), 3.15-3.35(m, 4H), 3.79(s, 3H), 4.97 (s, 1H), 4.82 $\pm$ 6.37(ABq, J=19.2 Hz, 2H), 6.7-6.85(m, 2H), 7.1-7.5 (m, 10H), 7.71(dd, J=3.0, 12.9Hz, 1H), 7.90(d, J=3.0Hz, 1H)	(KBr tablet) 1715, 1607, 1310, 1248, 1152, 1000	550 (M <sup>+</sup> )	-
53 (25a)	yellow solid	80 - 81 [iso-propyl ether]*	(CDCl <sub>3</sub> ) 2.28-3.26(m, 12H), 3.81 (s, 3H), 5.00(s, 1H), 4.86 $\pm$ 6.36 (ABq, J=13.2Hz, 2H), 6.68-7.98(m, 10H)	(KBr tablet) 1714, 1609, 1353, 1242, 1007	544 (M <sup>+</sup> )	C <sub>26</sub> H <sub>28</sub> N <sub>2</sub> O <sub>5</sub> S <sub>3</sub> C H N 57.31 5.30 5.09 57.33 5.18 5.14
54 (27a)	colorless oil	-	(CDCl <sub>3</sub> ) 2.15-2.75(m, 10H), 3.15 -3.45(m, 2H), 3.8-3.9(m, 12H), 5.04(s, 1H), 4.87 $\pm$ 6.38(ABq, J=17.1Hz, 2H), 6.62(d, J=12.9Hz, 1H), 6.80(d, J=12.9Hz, 1H), 6.91 (d, J=12.9Hz, 1H), 7.1-7.47(m, 4H), 7.74(dd, J=3.0, 12.9Hz, 1H), 7.93(d, J=3.0Hz, 1H)	(CHCl <sub>3</sub> ) 1713, 1613, 1461, 1295, 1120, 1009	592 (M <sup>+</sup> )	-

Example No. (Compound No.)	Appearance	MP (°C) [Solvent for re-crystallization]	NMR (solvent) $\delta$ , ppm	IR (method) $\text{cm}^{-1}$	MS m/z	Elemental analysis (%) Upper column: found Lower column: calcd.
55 (28a)	yellow amorphous	-	(CDCl <sub>3</sub> ) 2.19-2.70(m, 8H), 3.36-3.83(m, 4H), 3.83(s, 3H), 5.03(s, 1H), 4.85 $\pm$ 6.41(ABq, J=12.8Hz, 2H), 6.61-8.06(m, 14H)	(KBr tablet) 1711, 1645, 1606, 1245, 982	528 (M <sup>+</sup> )	-
56 (30a)	white solid	133 - 135 [isopropyl ether]	(CDCl <sub>3</sub> ) 2.2-2.8(m, 8H), 3.3-3.7(m, 4H), 3.85(s, 3H), 4.87 $\pm$ 6.43(ABq, J=12.8Hz, 2H), 5.04(s, 1H), 6.83(d, J=8.4Hz, 1H), 6.95-7.5(m, 9H), 7.76(dd, J=2.4, 8.4Hz, 1H), 7.97(d, J=2.4Hz, 1H)	(KBr tablet) 1712, 1611, 1497, 1293, 1248, 1199, 1118	518 (M <sup>+</sup> )	C <sub>29</sub> H <sub>30</sub> N <sub>2</sub> O <sub>5</sub> C 67.00 5.90 5.12 H 67.16 5.83 5.40 N
57 (31a)	colorless oil	-	(CDCl <sub>3</sub> ) 2.05-2.8(m, 8H), 3.4-3.65(m, 4H), 3.80(s, 3H), 4.84 $\pm$ 6.39(ABq, J=12.6Hz, 2H), 5.02(s, 1H), 5.08(s, 2H), 6.81(d, J=8.5 Hz, 1H), 7.0-7.5(m, 9H), 7.74(dd, J=2.4, 8.5Hz, 1H), 7.93(d, J=2.4Hz, 1H)	(neat) 2948, 1710, 1610, 1433, 1237, 1118, 1006	532 (M <sup>+</sup> )	-
58 (8-78a)	pale yellow crystal	193 - 196 [isopropanol]	(CDCl <sub>3</sub> ) 2.45-2.9(m, 4H), 3.0-3.6(m, 6H), 3.91(s, 3H), 4.7-5.7(m, 2H), 6.23(t, J=7.6Hz, 1H), 6.5-8.1(m, 14H)	(KBr tablet) 1715, 1607, 1310, 1248, 1152, 1000, 945	530 (M <sup>+</sup> )	C <sub>30</sub> H <sub>30</sub> N <sub>2</sub> O <sub>5</sub> C 67.55 5.90 5.39 H 67.90 5.70 5.28 N

Example No. (Compound No.)	Appearance	MP (°C) [Solvent for re-crystallization]	NMR (solvent) $\delta$ , ppm	IR (method cm <sup>-1</sup> )	MS m/z	Elemental analysis (%) Upper column: found Lower column: calcd.
59 (E-79a)	colorless oil	-	(CDCl <sub>3</sub> ) 2.45-2.6(m, 4H), 3.0-3.3(m, 6H), 3.86(s, 3H), 4.7-5.7(m, 2H), 6.13(t, J=6.8Hz, 1H), 7.0-7.65(m, 7H), 7.78(dd, J=2.2, 8.6Hz, 1H), 6.77(d, J=8.6Hz, 1H), 7.98(d, J=2.2Hz, 1H)	(KBr tablet) 1711, 1605, 1351, 1241, 999	510 (M <sup>+</sup> )	-
60 (34a)	colorless oil	-	(CDCl <sub>3</sub> ) 1.6-2.0(m, 2H), 2.45-2.85(m, 8H), 3.3-3.6(m, 4H), 3.86(s, 3H), 4.88 & 6.43(ABq, J=12.1 Hz, 2H), 4.99(s, 1H), 7.25-8.85(m, 13H)	(KBr tablet) 1713, 1609, 1321, 1243, 1005	602 (M <sup>+</sup> )	-



In Examples 61 through 63 below, the objective compound was prepared by hydrolyzing esters of the corresponding oxepine derivatives in a manner similar to Example 23.

5

#### Example 61

11-[2-[4-(Styrylsulfonyl)-1-piperazinyl]ethyl]thio-6,11-dihydrodibenz[b,e]oxepin-2-carboxylic acid\*0.5 hydrate (Compound 24b)

10

#### Example 62

11-[2-[4-(2-Thienylsulfonyl)-1-piperazinyl]ethyl]thio-6,11-dihydrodibenz[b,e]oxepin-2-carboxylic acid\*0.5 isopropanol\*dihydrate (Compound 25b)

15

#### Example 63

11-[2-[4-(2,3,4-Trimethoxybenzoyl)-1-piperazinyl]ethyl]thio-6,11-dihydrodibenz[b,e]oxepin-2-carboxylic acid (Compound 27b)

20

#### Example 64

11-[2-[4-(2,3,4-Trimethoxybenzoyl)-1-piperazinyl]ethyl]thio-6,11-dihydrodibenz[b,e]oxepin-2-carboxylic acid\* monohydrochloride\*1.1 hydrate (Compound 27b)

25

The objective compound was prepared as the hydrochloride from Compound 27b obtained in Example 63 in a manner similar to Example 2.

In Examples 65 through 68 below, the objective compound was prepared by hydrolyzing esters of the corresponding oxepine derivatives in a manner similar to Example 23.

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#### Example 65

11-[2-[4-(Cinnamoyl)-1-piperazinyl]ethyl]thio-6,11-dihydrodibenz[b,e]oxepin-2-carboxylic acid (Compound 28b)

35

#### Example 66

11-[2-[4-(Benzyloxycarbonyl)-1-piperazinyl]ethyl]thio-6,11-dihydrodibenz[b,e]oxepin-2-carboxylic acid.dihydrate (Compound 31b):

40

#### Example 67

(E)-11-[2-[4-(Styrylsulfonyl)-1-piperazinyl]ethylidene]-6,11-dihydrodibenz[b,e]oxepin-2-carboxylic acid\*0.1 isopropanol\*0.1 hydrate (Compound E-78b)

45

#### Example 68

(E)-11-[2-[4-(2-Thienylsulfonyl)-1-piperazinyl]ethylidene]-6,11-dihydrodibenz[b,e]oxepin-2-carboxylic acid\*0.5 hydrate (Compound E-78b)

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Physicochemical properties of the compounds obtained in Examples 61 to 68 are shown in Table 7-5.

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Table 7-5

Example No. (Compound No.)	Appearance	MP (°C) [Solvent for recrystallization]	NMR (solvent) $\delta$ , ppm	IR (method) $\text{cm}^{-1}$	MS m/z	Elemental analysis (%) Upper column: found Lower column: calcd.
61 (24b')	white crystal	232 - 235 [isopropanol]	(DMSO- $d_6$ ) 2.3-2.7 (m, 8H), 3.0-3.3 (m, 4H), 5.44 (s, 1H), 5.10 s 6.31 (ABq, J=19.2Hz, 2H), 6.86 (d, J=13.5Hz, 1H), 7.25-7.65 (m, 10H), 7.65-7.9 (m, 2H), 7.98 (d, J=3.0Hz, 1H)	(KBr tablet) 1685, 1603, 1313, 1246, 1157, 1002, 957	-	$\text{C}_{29}\text{H}_{30}\text{N}_2\text{O}_5 \cdot 0.5\text{H}_2\text{O}$ C H N 62.30 5.63 4.82 62.23 5.58 5.00
62 (25b')	white crystal	157 - 168 [isopropanol]	(DMSO- $d_6$ ) 2.50-3.46 (m, 12H), 5.05 s 6.17 (ABq, J=12.7Hz, 2H), 5.47 (s, 1H), 6.86-8.11 (m, 10H), 12.76 (bs, 1H)	(KBr tablet) 3400, 2860, 1687, 1607, 1354, 1159	530 (M+)	$\text{C}_{25}\text{H}_{26}\text{N}_2\text{O}_5 \cdot 0.5\text{H}_2\text{O}$ $\text{C}_3\text{H}_8\text{O} \cdot 2\text{H}_2\text{O}$ C H N 53.53 5.20 4.55 53.34 5.41 4.69
63 (27b)	pale yellow amorphous	-	(DMSO- $d_6$ ) 2.85-3.65 (m, 12H), 3.75-3.85 (m, 9H), 5.52 (s, 1H), 5.09 s 6.15 (ABq, J=12.9Hz, 2H), 6.85-7.00 (m, 3H), 7.4-7.5 (m, 4H), 7.73 (dd, J=2.1, 8.6Hz, 1H), 8.04 (d, J=2.1Hz, 1H)	-	578 (M+)	-
64 (27b')	white crystal	165 - 168 [isopropyl ether]	-	(KBr tablet) 2934, 1583, 1462, 1227, 1011	-	$\text{C}_{31}\text{H}_{34}\text{N}_2\text{O}_5 \cdot \text{HCl} \cdot \text{H}_2\text{O}$ C H N 58.63 6.23 4.23 58.64 5.90 4.41

Example No. (Compound No.)	Appearance	MP (°C) [Solvent for re-crystallization]	NMR (solvent) $\delta$ , ppm	IR (method) $\text{cm}^{-1}$	MS $m/z$	Elemental analysis (%) Upper column: found Lower column: calcd.
65 (28b)	white crystal	218 - 219 [isopropanol]	(DMSO- $d_6$ ) 2.25-3.75(m, 12H), 5.05 & 6.28(ABq, J=12.5Hz, 2H), 5.45(s, 1H), 6.87(d, J=8.5Hz, 1H), 7.24(d, J=15.5Hz, 1H), 7.3-7.5(m, 5H), 7.48(d, J=15.5Hz, 1H), 7.65-7.8(m, 3H), 8.00(d, J=2.1Hz, 1H)	(KBr tablet) 1692, 1642, 1566, 1227, 1004, 987	514 (M <sup>+</sup> )	$\text{C}_{30}\text{H}_{30}\text{N}_2\text{O}_5$ C H N 69.91 5.96 5.39 70.02 5.88 5.44
66 (31b')	white crystal	207 - 208 [toluene]	(DMSO- $d_6$ ) 2.8-4.2(m, 12H), 5.12(s, 2H), 5.09 & 6.15(ABq, J=12.9 Hz, 2H), 5.52(s, 1H), 6.90(d, J=8.6Hz, 1H), 7.3-7.6(m, 9H), 7.73(dd, J=2.2, 8.6Hz, 1H), 8.04(d, J=2.2Hz, 1H), 11.25(bs, 1H)	(KBr tablet) 1714, 1677, 1611, 1436, 1200, 1161, 1005	518 (M <sup>+</sup> )	$\text{C}_{29}\text{H}_{30}\text{N}_2\text{O}_5 \cdot 2\text{H}_2\text{O}$ C H N 62.43 6.51 4.96 62.80 6.18 5.05
67 (E-78b')	white crystal	233 - 235 (dec.) [isopropanol]	(DMSO- $d_6$ ) 2.2-3.5(m, 10H), 5.00 & 5.52(each bs, 2H), 6.10(t, J=6.6Hz, 1H), 6.83(d, J=8.5Hz, 1H), 7.2-7.9(m, 11H)	(KBr tablet) 1685, 1603, 1570, 1348, 1313, 1246, 1002, 957	516 (M <sup>+</sup> )	$\text{C}_{29}\text{H}_{28}\text{N}_2\text{O}_5$ $0.5\text{H}_2\text{O} \cdot 0.1\text{C}_3\text{H}_8$ C H N 66.37 5.70 5.11 66.20 5.65 5.27
68 (E-79b')	white crystal	250 - 251 (dec.) [isopropanol]	(DMSO- $d_6$ ) 2.3-3.5(m, 10H), 4.99 & 5.50(each bs, 2H), 6.07(bs, 1H), 6.82(d, J=8.5Hz, 1H), 7.2-7.65(m, 6H), 7.71(dd, J=2.2, 8.5 Hz, 1H), 7.87(d, J=2.2Hz, 1H), 8.05-8.10(m, 1H)	(KBr tablet) 1685, 1607, 1353, 1316, 1246, 1166, 1009, 954	496 (M <sup>+</sup> )	$\text{C}_{25}\text{H}_{24}\text{N}_2\text{O}_5 \cdot 0.5\text{H}_2\text{O}$ C H N 59.33 4.74 5.47 59.39 4.98 5.54

Pharmaceutical Preparation 1 Tablet

A tablet having the following composition is prepared in a conventional manner.

Compound 1b	100 mg
Lactose	60 mg
Potato starch	30 mg
Polyvinyl alcohol	2 mg
Magnesium stearate	1 mg
Tar pigment	trace

Pharmaceutical Preparation 2 Powder

Powders having the following composition are prepared in a conventional manner.

Compound 38b	100 mg
Lactose	300 mg

Pharmaceutical Preparation 3 Syrup

Syrup having the following composition is prepared in a conventional manner.

Compound 38b	100 mg
Refined sugar	30 mg
Ethyl p-hydroxybenzoate	40 mg
Propyl p-hydroxybenzoate	10 mg
Strawberry flavor	0.1 cc
Water is added until the whole volume is 100 cc.	

Pharmaceutical Preparation 4 Tablet

A tablet having the following composition is prepared in a conventional manner.

Compound E-78b	100 mg
Lactose	60 mg
Potato starch	30 mg
Polyvinyl alcohol	2 mg
Magnesium stearate	1 mg
Tar pigment	trace

Pharmaceutical Preparation 5 Syrup

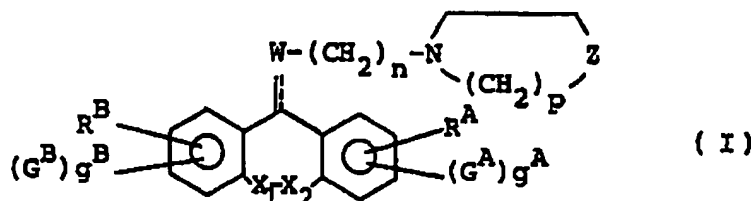
Syrup having the following composition is prepared in a conventional manner.

Compound E-78b'	100 mg
Refined sugar	30 g
Ethyl p-hydroxybenzoate	40 mg
Propyl p-hydroxybenzoate	10 mg
Strawberry flavor	0.1 cc
Water is added until the whole volume is 100 cc.	

While the invention has been described in detail and with reference to specific embodiments thereof, it will be apparent to one skilled in the art that various changes and modifications can be made therein without departing from the spirit and scope thereof.

### Claims

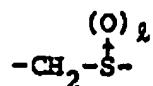
1. A tricyclic compound represented by formula (I):



wherein

— represents single bond or double bond;

$\overline{X_1-X_2}$  represents  $-\text{CH}_2\text{O}-$ .



wherein  $l$  represents 0, 1 or 2,  $-\text{CH}_2-\text{CH}_2-$ , or  $-\text{CH}=\text{CH}-$ ;

$W$  represents  $-\text{S}-$  or  $=\text{CH}-$ ;

$n$  is 1, 2, 3, or 4;

one of  $R^A$  and  $R^B$  represents hydrogen and the other represents  $-\text{Y}-\text{M}$  wherein  $Y$  represents single bond,  $-\text{CR}^1\text{R}^2-(\text{CH}_2)_m-$ , or  $-\text{CR}^1=\text{CR}^2-(\text{CH}_2)_m-$  wherein each of  $R^1$  and  $R^2$  independently represents hydrogen or lower alkyl and  $m$  is 0, 1, 2, 3 or 4, in which the left side of each formula is bound to the mother nucleus;

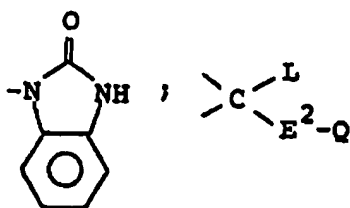
and  $M$  represents  $-\text{COOR}^3$  wherein  $R^3$  represents hydrogen or lower alkyl,  $-\text{CONR}^{3a}\text{R}^{3b}$  wherein each of  $R^{3a}$  and  $R^{3b}$  independently has the same significances for  $R^3$  as described above, or tetrazolyl;

each of  $G^A$  and  $G^B$  independently represents lower alkyl, halogen, hydroxyl, or lower alkoxy;

each of  $g^A$  and  $g^B$  independently represents 0, 1, 2 or 3;

$Z$  represents  $\text{>N-E-Q}$  wherein  $E$  represents single bond,  $-\text{CO}-$ ,  $-\text{COO}-$  wherein the left side of the formula is bound to the nitrogen atom, or  $-\text{SO}_2-$ ; and

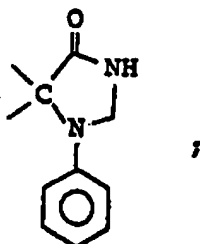
$Q$  represents optionally substituted aryl, optionally substituted aralkyl, optionally substituted aralkenyl, aromatic heterocyclic group, or



10 wherein  $L$  represents hydrogen, hydroxyl, or lower alkoxy;  $E^2$  represents single bond,  $-CO-$ , or  $-\overset{\text{O}}{\underset{\text{R}^4}{\text{CH}}}-$

wherein  $R^4$  represents hydrogen or lower alkyl; and  
 $Q$  has the same significance as described above;

15  $\geq C=CH-Q$  wherein  $Q$  has the same significance as described above; or



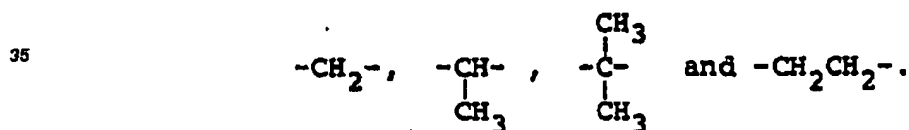
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$p$  is 1, 2 or 3;

and a pharmaceutically acceptable salt thereof.

2. A compound according to claim 1, wherein one of  $R^A$  and  $R^B$  represents hydrogen and the other represents  $-Y-COOH$ .

3. A compound according to claim 2, wherein  $Y$  is a member selected from the group consisting of single bond,

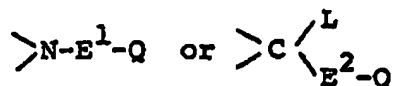


4. A compound according to claim 3, wherein  $X_1-X_2$  represents  $-\text{CH}_2\text{O}-$ ,  $n$  is 2 and  $W$  represents  $-S-$ .

5. A compound according to claim 3, wherein  $X_1-X_2$  represents  $-\text{CH}_2\text{O}-$ ,  $n$  is 1 or 2 and  $W$  represents  $=\text{CH}-$ .

6. A compound according to any one of claims 4 and 5, wherein  $Z$  is

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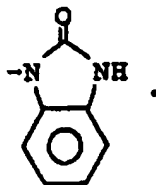
50 and  $p$  is 2 or 3.

7. A compound according to claim 8, wherein  $Q$  represents optionally substituted aryl having 6 to 10 carbon atoms, optionally substituted aralkyl having 7 to 20 carbon atoms or optionally substituted aralkenyl having 8 to 18 carbon atoms.

8. A compound according to claim 7, the optional substituent on the aryl, aralkyl or aralkenyl means 1 to 3 substituents on the aromatic ring and the substituent(s) is/are independently selected from the group consisting of lower alkyl, halogen, trifluoromethyl, hydroxyl, lower alkoxy and methylenedioxy formed together with the ortho-position of the aromatic ring.

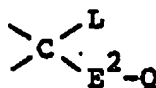
9. A compound according to claim 8, wherein Q represents the aromatic heterocyclic group selected from the group consisting of furyl, thienyl, pyridyl, pyrimidinyl, quinolyl and isoquinolyl.

10. A compound according to any one of claims 4 and 5, wherein p is 2; and Z is  $\text{>N-E}^1\text{-Q}$  wherein  $\text{E}^1$  represents single bond or  $-\text{SO}_2-$ , Q is selected from the group consisting of optionally substituted phenyl, optionally substituted benzyl, optionally substituted benzhydryl, phenethyl, styryl, cinnamyl, thienyl and

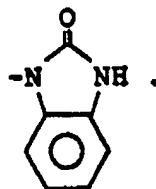


11. A compound according to claim 10, which is selected from the group consisting of
- 11-[2-(4-phenyl-1-piperazinyl)ethyl]thio-8,11-dihydrodibenz[b,e]oxepin-2-carboxylic acid,
  - 11-[2-(4-(4-fluorophenyl)-1-piperazinyl)ethyl]thio-8,11-dihydrodibenz[b,e]oxepin-2-carboxylic acid,
  - 11-[2-(4-(2-methoxyphenyl)-1-piperazinyl)ethyl]thio-8,11-dihydrodibenz[b,e]oxepin-2-carboxylic acid,
  - 11-[2-(4-benzyl-1-piperazinyl)ethyl]thio-8,11-dihydrodibenz[b,e]oxepin-2-carboxylic acid,
  - 11-[2-(4-(2-thienylsulfonyl)-1-piperazinyl)ethyl]thio-8,11-dihydrodibenz[b,e]oxepin-2-carboxylic acid,
  - 11-[3-(4-benzyl-1-piperazinyl)propylidene]-8,11-dihydrodibenz[b,e]oxepin-2-carboxylic acid,
  - 11-[2-(4-(4-fluorophenyl)piperazinyl)ethylidene]-8,11-dihydrodibenz[b,e]oxepin-2-carboxylic acid,
  - 11-[2-(4-cinnamyl-1-piperazinyl)ethylidene]-8,11-dihydrodibenz[b,e]oxepin-2-carboxylic acid,
  - 11-[2-(4-(styrylsulfonyl)-1-piperazinyl)ethylidene]-8,11-dihydrodibenz[b,e]oxepin-2-carboxylic acid and
  - 11-[2-(4-(2-thienylsulfonyl)-1-piperazinyl)ethylidene]-8,11-dihydrodibenz[b,e]oxepin-2-carboxylic acid.

12. A compound according to any one of claims 4 and 5, wherein p is 2; and Z is



wherein L represents hydrogen or hydroxyl,  $\text{E}^2$  represents single bond or  $-\text{CO}-$ , and Q is selected from the group consisting of optionally substituted phenyl, optionally substituted benzyl, optionally substituted benzhydryl, phenethyl, styryl, cinnamyl, thienyl and



13. A compound according to claim 12, which is selected from the group consisting of
- 11-[2-(4-phenyl-1-piperidinyl)ethyl]thio-8,11-dihydrodibenz[b,e]oxepin-2-carboxylic acid,
  - 11-[2-(4-benzyl-1-piperidinyl)ethyl]thio-8,11-dihydrodibenz[b,e]oxepin-2-carboxylic acid,
  - 11-[2-(4-benzyl-1-piperidinyl)ethyl]thio-8,11-dihydrodibenz[b,e]oxepin-2-acetic acid,
  - 2-methyl-2-[11-[2-(4-benzyl-1-piperidinyl)ethyl]thio-8,11-dihydrodibenz[b,e]oxepin-2-yl]propionic acid,
  - 11-[2-(4-(4-chlorobenzyl)-1-piperidinyl)ethyl]thio-8,11-dihydrodibenz[b,e]oxepin-2-carboxylic acid,
  - 2-methyl-2-[11-[2-(4-(4-chlorobenzyl)-1-piperidinyl)ethyl]thio-8,11-dihydrodibenz[b,e]oxepin-2-yl]propionic acid,
  - 11-[2-(4-[2-(3H)-benzimidazol-1-yl]piperidino)ethyl]thio-8,11-dihydrodibenz[b,e]oxepin-2-carboxylic acid,
  - 11-[3-(4-benzyl-1-piperidinyl)propylidene]-8,11-dihydrodibenz[b,e]oxepin-2-carboxylic acid,
  - 11-[2-(4-benzyl-1-piperidinyl)ethylidene]-8,11-dihydrodibenz[b,e]oxepin-2-acetic acid and
  - 11-[2-(4-[2-(3H)-benzimidazol-1-yl]piperidino)ethylidene]-8,11-dihydrodibenz[b,e]oxepin-2-acetic acid.

14. A compound according to claim 1, wherein said salt is selected from the group consisting of acid addition salt, metal salt, ammonium salt, organic amine addition salt and amino acid addition salt.

15. A pharmaceutical composition comprising a pharmaceutical carrier and, as an active ingredient, an effective amount of a tricyclic compound as defined by claim 1.

16. A compound according to any of claims 1 to 14 usable as a medicament.

17. Use of a compound according to any of claims 1 to 14 for the manufacture of a medicament having  
5 a thromboxane A<sub>2</sub> antagonizing activity.

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European Patent  
Office

# EUROPEAN SEARCH REPORT

Application Number

EP 88 12 0889

DOCUMENTS CONSIDERED TO BE RELEVANT			
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int. CL.4)
Y,D	EP-A-0 235 796 (KYOWA HAKKO KOGYO) * whole document * ---	1,15	C 07 D 313/12 C 07 D 337/12
Y,D	EP-A-0 188 802 (KYOWA HAKKO KOGYO) * whole document * ---	1,15	C 07 D 211/10 C 07 D 333/34 C 07 D 405/06
Y,D	EP-A-0 214 779 (THE WELLCOME FOUNDATION) * whole document * ---	1,15	C 07 D 405/12 C 07 D 409/06 C 07 D 409/14 C 07 D 409/12
Y,D	EP-A-0 130 555 (KYOWA HAKKO KOGYO CO) * page 35, claim 1, page 1, lines 32-37 *; & JP - A - 28972, 1985 ---	1,15	C 07 D 471/10 A 61 K 31/445 A 61 K 31/50 A 61 K 31/55
A	GB-A-1 347 935 (YOSHITOMI PHARMACEUTICALS) * page 1, column 1, lines 10-25, page 2, column 2, lines 19-25 * ---	1	
A	GB-A-1 330 966 (YOSHITOMI PHARMACEUTICALS) * page 1, columns 1,2, lines 10-23, page 3, column 1, lines 7-12 * ---	1	
Y	CHEMICAL ABSTRACTS vol. 72, no. 1, 5th January 1970, page 308, abstract no. 3387d, Columbus, Ohio, US; M. PROTIVA et al.: "Tricyclic 3-(4-phenylpiperidino)propylidene derivatives."; & Czech 128.004, 15-01-1968 ---	1,15	TECHNICAL FIELDS SEARCHED (Int. CL.4) C 07 D 313/00 C 07 D 337/12 C 07 D 405/00 C 07 D 409/00
A	DE-A-2 407 661 (JANSSEN PHARMACEUTICA) * page 20, claim 1, page 8, lines 1-10 * --- -/-	1	
The present search report has been drawn up for all claims			
Place of search BERLIN		Date of completion of the search 10-04-1989	Searcher KYRIAKAKOU G
<b>CATEGORY OF CITED DOCUMENTS</b> X : particularly relevant if taken alone Y : particularly relevant if combined with another document of the same category A : technological background O : non-written disclosure P : intermediate document T : theory or principle underlying the invention E : earlier patent document, but published on, or after the filing date D : document cited in the application L : document cited for other reasons A : member of the same patent family, corresponding document			

EPO FORM 150 (04/88) (P0401)



DOCUMENTS CONSIDERED TO BE RELEVANT			
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	CLASSIFICATION OF THE APPLICATION (Int. Cl. 4)
Y	CHEMICAL ABSTRACTS vol. 81, no. 5, 5th August 1974, page 424, column 1, abstract no. 25566z, Columbus, Ohio, US; & JP - A - 73 30064, 17-09-73 ---	1,15	
Y,D	US-A-4 282 365 (J. ROKACH et al.) * column 1, lines 10-60, column 2, lines 40-63 * ---	1,15	
A	EUROPEAN JOURNAL OF MEDICINAL CHEMISTRY - CHEMICA THERAPEUTICA vol. 9, no. 3, May-June 1974, pages 259-262; P. DOSTERT et al.: "Composés tricycliques portant une chaîne alkylaminoalkylthio Synthèse et activité pharmacologique." * whole document * ---	1,15	
A,D	JOURNAL OF MEDICINAL CHEMISTRY no. 29, 1986, pages 2347-2351, Columbus, Ohio, US; L.D. WATERBURG et al.: "Antiinflammatory and Aldose Reductase Inhibitory Activity of Some Tricyclic Arylacetic Acids." * whole document * -----	1,15	
The present search report has been drawn up for all claims			TECHNICAL FIELDS SEARCHED (Int. Cl. 4)
Place of search BERLIN		Date of completion of the search 10-04-1989	Examiner KYRIAKAKOU G
<b>CATEGORY OF CITED DOCUMENTS</b> X : particularly relevant if taken alone Y : particularly relevant if combined with another document of the same category A : technological background O : non-written disclosure P : intermediate document  T : theory or principle underlying the invention E : earlier patent document, but published on, or after the filing date D : document cited in the application L : document cited for other reasons  * : member of the same patent family, corresponding document			